

## SUPPLEMENTAL MATERIAL

to

### **Combustion chemistry and flame structure of furan group biofuels using molecular-beam mass spectrometry and gas chromatography – Part III: 2,5-dimethylfuran**

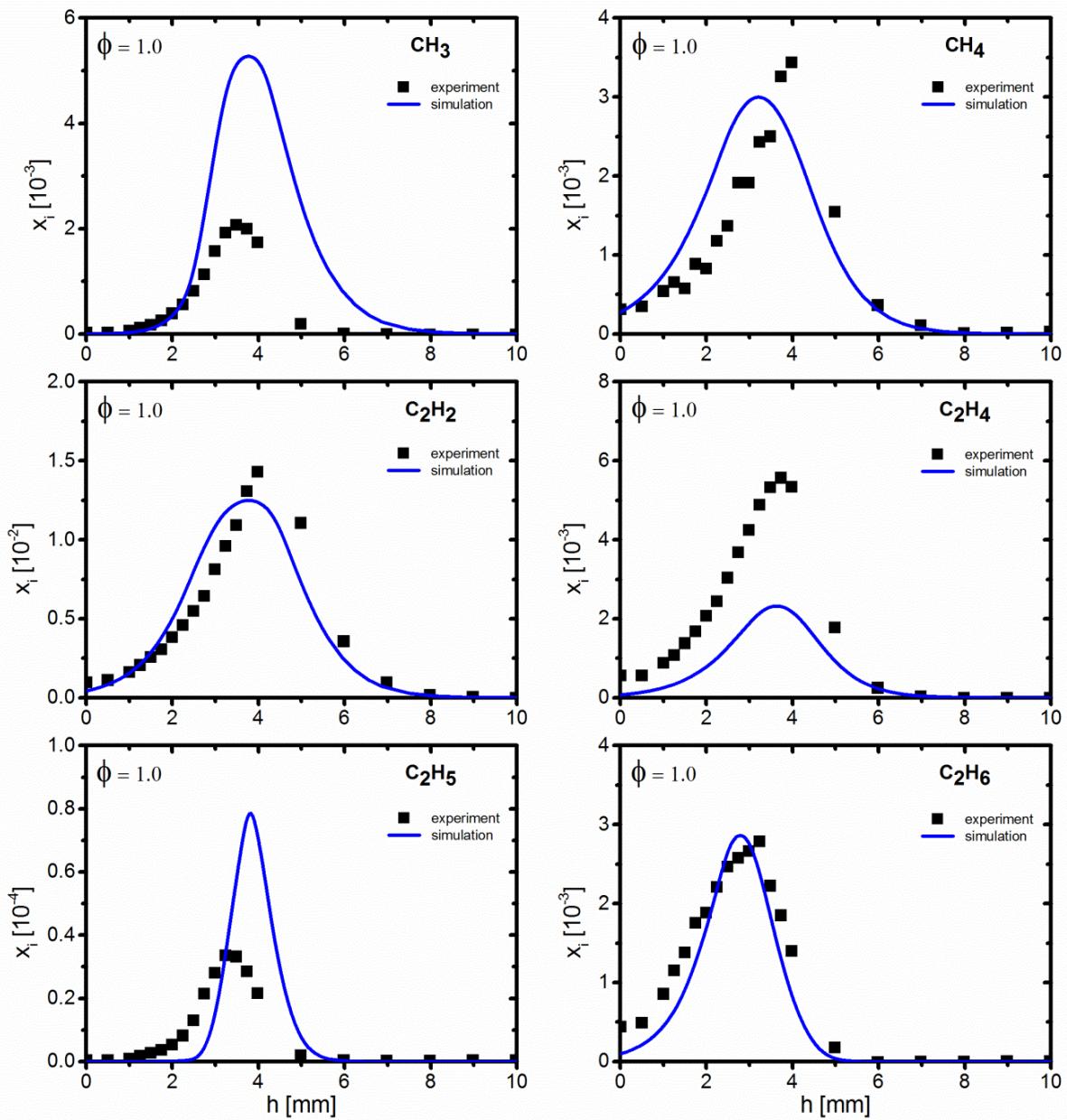
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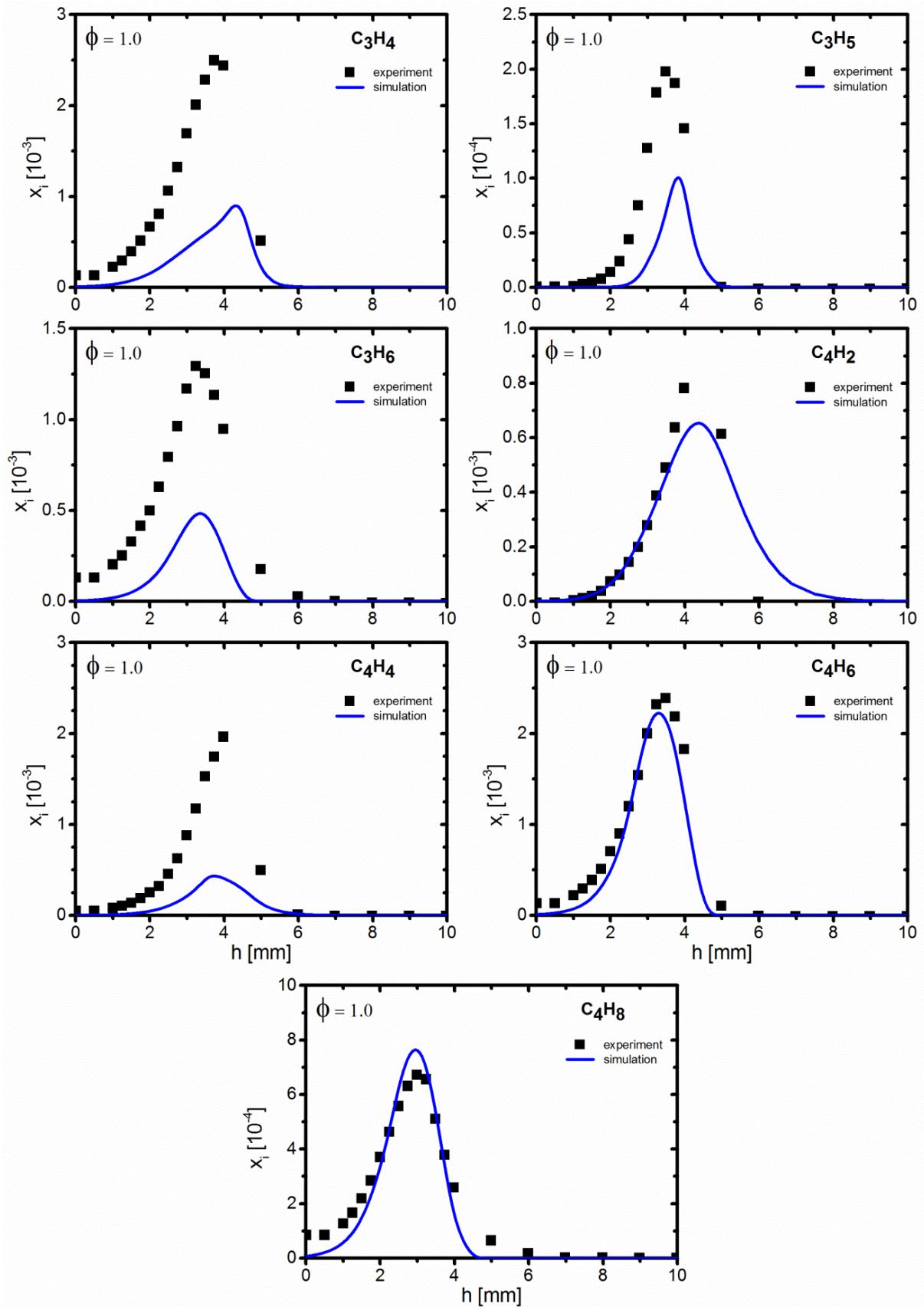
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# now at German Aerospace Center (DLR), Institute of Combustion Technology, Pfaffenwaldring 38-40, D-70569 Stuttgart, Germany

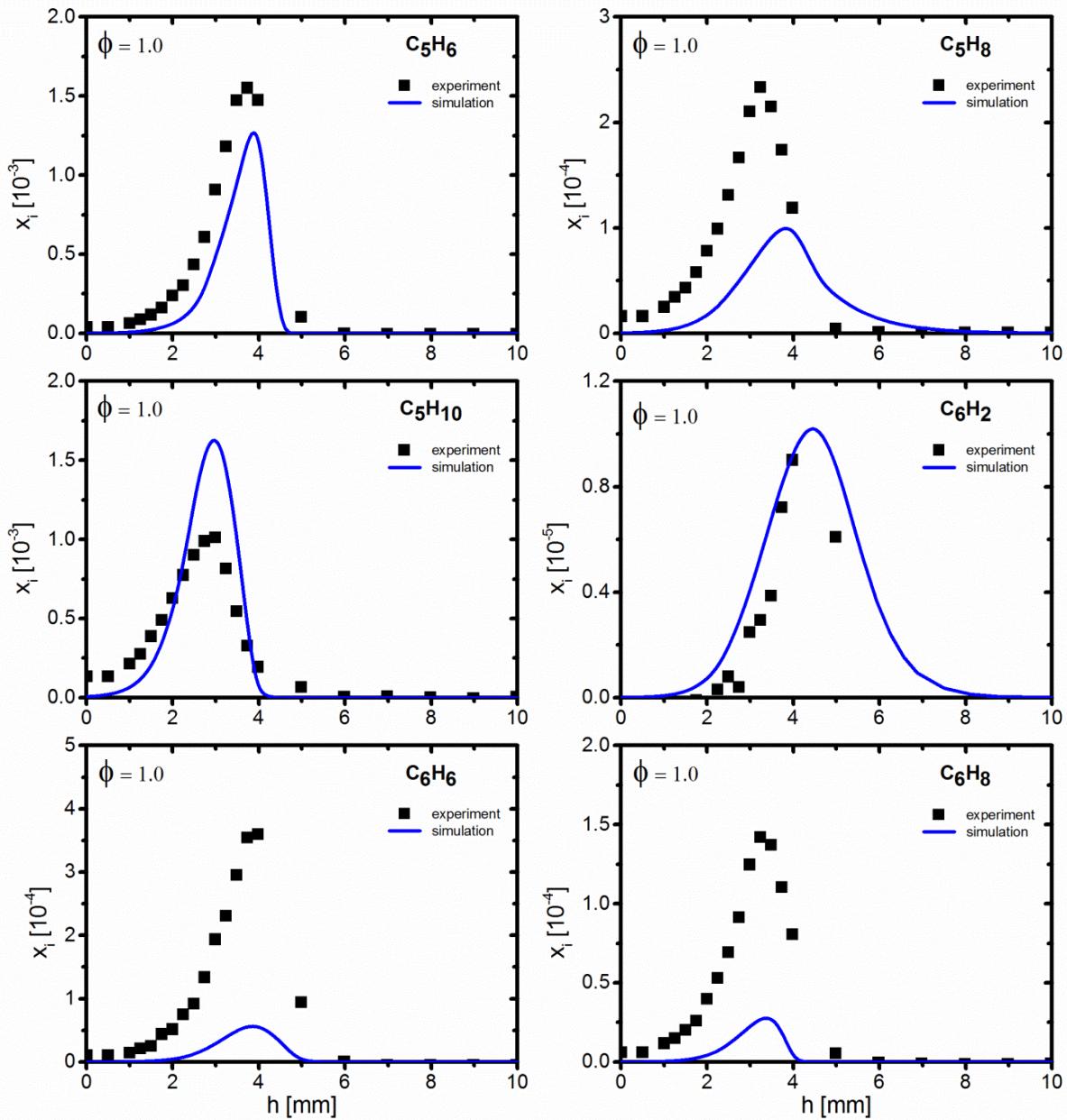
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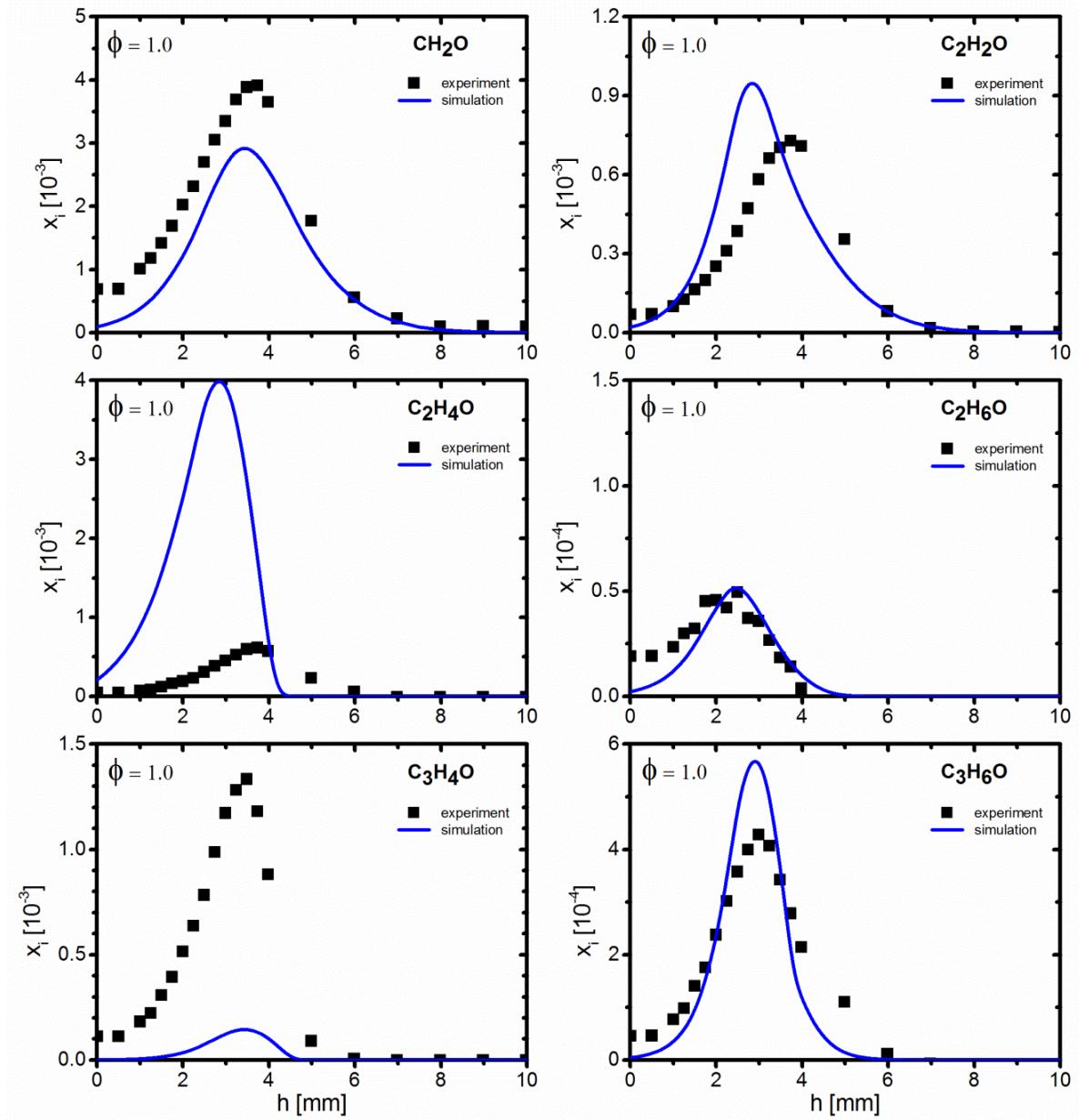
**Figure S1.** Mole fraction profiles of selected  $\text{C}_1$  and  $\text{C}_2$  species profiles for  $\phi = 1.0$ . Symbols: experiment; lines: simulation.



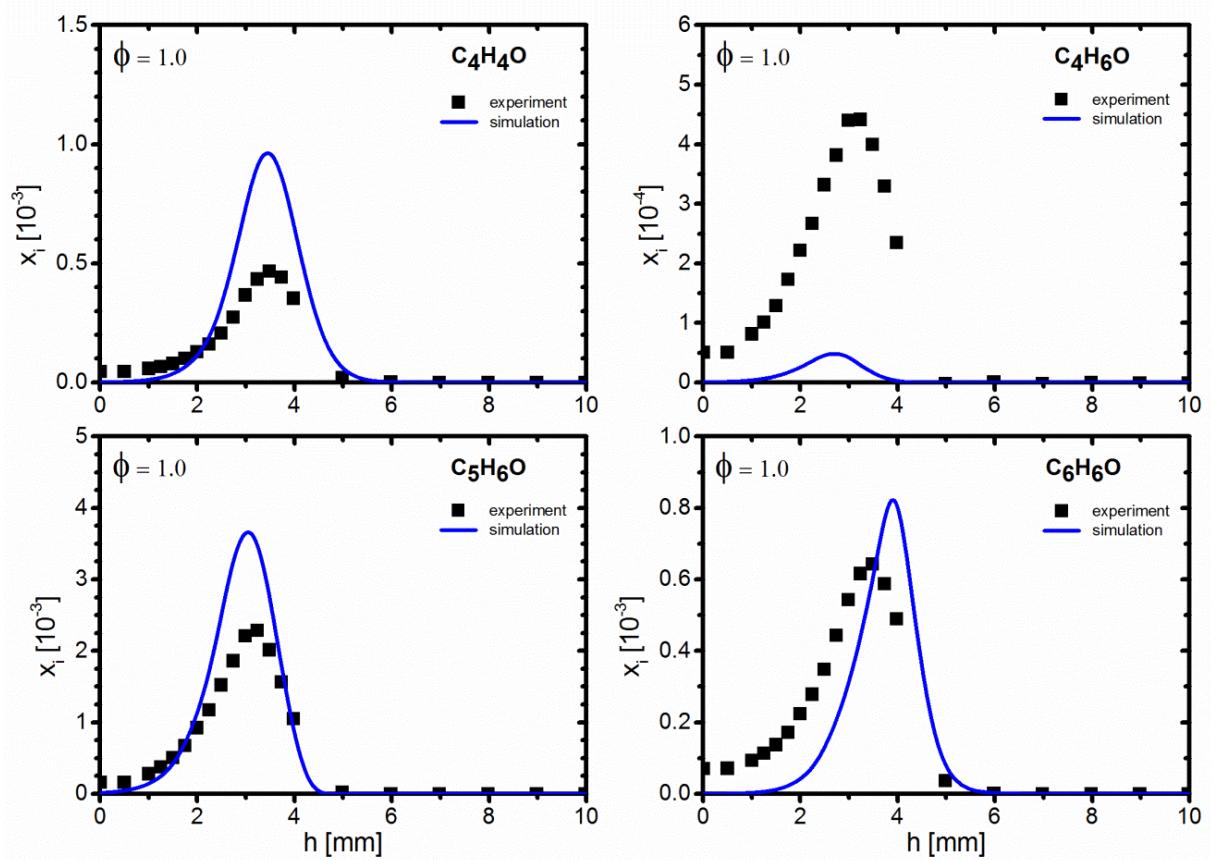
**Figure S2.** Mole fraction profiles of selected  $C_3$  and  $C_4$  species profiles for  $\phi = 1.0$ . Symbols: experiment; lines: simulation.



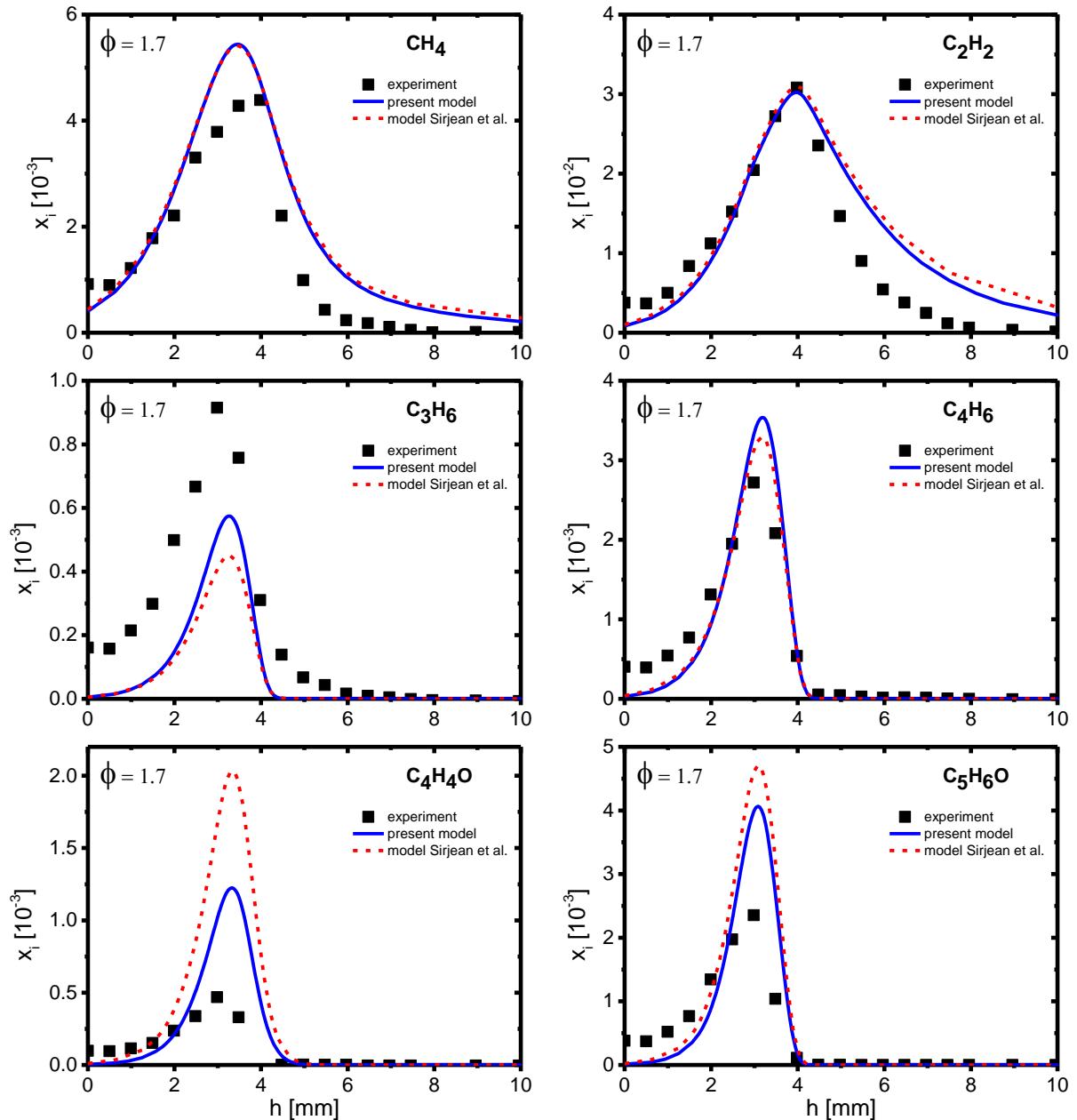
**Figure S3.** Mole fraction profiles of selected  $C_5$  and  $C_6$  species profiles for  $\phi = 1.0$ . Symbols: experiment; lines: simulation.



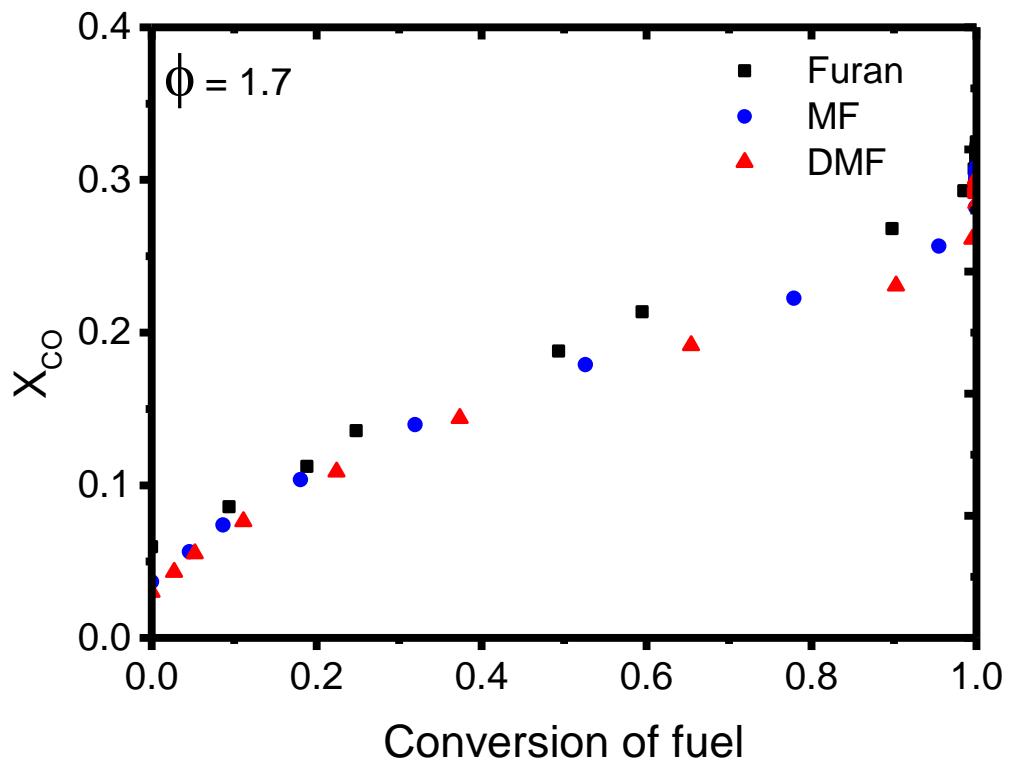
**Figure S4.** Mole fraction profiles of selected  $\text{C}_1\text{-}\text{C}_3$  oxygenated species profiles for  $\phi = 1.0$ . Symbols: experiment; lines: simulation.



**Figure S5.** Mole fraction profiles of selected C<sub>4</sub>-C<sub>6</sub> oxygenated species profiles for  $\phi = 1.0$ . Symbols: experiment; lines: simulation.



**Figure S6.** Examples for the comparison between the present model and that of Sirjean et al. [28].



**Figure S7.** Mole fraction profiles of CO profiles as a function of fuel conversion.

**Table S1**

Comparison between the maximum mole fractions ( $x$ ) of the main intermediate species measured in the three fuel flames. The largest mole fractions for each species obtained in the three fuels are written in bold.

Species quantified		$\phi = 1.0$			$\phi = 1.7$		
Formula	Name	$x_{\text{Furan}}$ (ppm)	$x_{\text{MF}}$ (ppm)	$x_{\text{DMF}}$ (ppm)	$x_{\text{Furan}}$ (ppm)	$x_{\text{MF}}$ (ppm)	$x_{\text{DMF}}$ (ppm)
$\text{CH}_3$	Methyl	1140	1920	<b>2070</b>	1490	1530	<b>1960</b>
$\text{CH}_4$	Methane	2360	3050	<b>3440</b>	3300	4260	<b>4390</b>
$\text{C}_2\text{H}_2$	Acetylene	<b>18200</b>	15900	14300	<b>34900</b>	33400	30800
$\text{C}_2\text{H}_3$	Vinyl	<b>6.32</b>	-	5.30	5.77	<b>5.95</b>	5.65
$\text{C}_2\text{H}_4$	Ethylene	5690	<b>6210</b>	5570	6720	<b>8040</b>	7050
$\text{C}_2\text{H}_5$	Ethyl	26.9	<b>41.0</b>	33.7	23.6	<b>25.2</b>	24.5
$\text{HCO}$	Formyl	<b>71.1</b>	44.6	34.7	<b>55.0</b>	29.7	25.2
$\text{C}_2\text{H}_6$	Ethane	1450	2370	<b>2790</b>	1980	2550	<b>3000</b>
$\text{CH}_2\text{O}$	Formaldehyde	3950	<b>3980</b>	3920	<b>3840</b>	3300	2910
$\text{CH}_3\text{O}$	Methanol	<b>492</b>	227	350	<b>436</b>	143	306
$\text{C}_3\text{H}_3$	Propargyl	37.2	<b>182</b>	160	104	248	<b>355</b>
$\text{C}_3\text{H}_4$	Propyne	1820	<b>2950</b>	2500	2960	<b>4760</b>	3790
$\text{C}_3\text{H}_5$	Allyl	152	<b>305</b>	198	151	<b>190</b>	165
$\text{C}_2\text{H}_2\text{O}$	Ketene	361	637	<b>2470</b>	358	551	<b>752</b>
$\text{C}_3\text{H}_6$	Propene	1430	<b>1830</b>	1290	1110	<b>1130</b>	915
$\text{C}_2\text{H}_4\text{O}$	Acetaldehyde	<b>3740</b>	1040	618	<b>3180</b>	608	674
$\text{C}_2\text{H}_6\text{O}$	Dimethylether	23.9	34.0	<b>49.6</b>	39.0	25.6	<b>42.0</b>
$\text{C}_4\text{H}_2$	1,3-Butadiyne	134	509	<b>782</b>	1080	2150	<b>4200</b>
$\text{C}_4\text{H}_4$	1-Buten-3-yne	163	1500	<b>3250</b>	482	2160	<b>3585</b>
$\text{C}_4\text{H}_5$	But-2-yn-1-yl	2.38	30.7	<b>42.2</b>	5.33	38.3	<b>122</b>
$\text{C}_4\text{H}_6$	1,3-Butadiene	248	2230	<b>2390</b>	456	2020	<b>2720</b>
$\text{C}_4\text{H}_7$	But-3-en-1-yl	7.75	<b>171</b>	81.0	14.0	<b>146</b>	42.7
$\text{C}_4\text{H}_8$	1-Butene	510	<b>1120</b>	672	643	<b>862</b>	409
$\text{C}_3\text{H}_4\text{O}$	Propenal	<b>2750</b>	1960	1330	<b>2450</b>	1370	847
$\text{C}_3\text{H}_6\text{O}$	Propanal	<b>859</b>	354	428	<b>417</b>	175	138
$\text{C}_5\text{H}_4$	1,3-Pentadiyne	5.93	22.5	<b>39.9</b>	28.2	57.9	<b>126</b>
$\text{C}_5\text{H}_6$	1,3-Cyclopentadiene	49.2	158	<b>1550</b>	101	202	<b>2500</b>
$\text{C}_5\text{H}_7$	3-Cyclopentenyl	0.9	2.16	<b>7.24</b>	1.3	2.23	<b>17.0</b>
$\text{C}_4\text{H}_4\text{O}$	Furan	Fuel	<b>3080</b>	468	Fuel	<b>3100</b>	469
$\text{C}_5\text{H}_8$	1,3-Pentadiene	n/a	58	<b>233</b>	n/a	68.9	<b>230</b>
$\text{C}_4\text{H}_6\text{O}$	2-Butenal	<b>396</b>	n/a	n/a	<b>291</b>	n/a	n/a
$\text{C}_4\text{H}_6\text{O}$	2-Butenone	n/a	395	<b>441.8</b>	n/a	241	<b>243</b>
$\text{C}_5\text{H}_{10}$	2-Pentene	13.0	<b>110</b>	101	24.7	<b>124</b>	69.2
$\text{C}_4\text{H}_8\text{O}$	Isobutanal	105	<b>114</b>	n/a	102	<b>421</b>	n/a
$\text{C}_4\text{H}_8\text{O}$	2-Butenone	n/a	n/a	32.1	n/a	n/a	12.4
$\text{C}_6\text{H}_2$	Triacetylene	3.22	<b>12.2</b>	9.02	36.6	60.9	<b>130</b>
$\text{C}_6\text{H}_4$	Benzyne	1.74	9.19	<b>11.5</b>	11.9	29.5	<b>54.6</b>
$\text{C}_6\text{H}_6$	Benzene	42.3	141	<b>360</b>	131	275	<b>1000</b>
$\text{C}_6\text{H}_8$	1,4-Cyclohexadiene	21.4	29.7	<b>142</b>	41	42.8	<b>308</b>
$\text{C}_5\text{H}_4\text{O}$	Allenylketene	n/a	<b>5.54</b>	0	n/a	5.49	<b>46.8</b>
$\text{C}_5\text{H}_5\text{O}$	2-Furanylmethyl	n/a	82.2	<b>179</b>	n/a	97.4	<b>272</b>
$\text{C}_5\text{H}_6\text{O}$	2-Methylfuran	68.8	Fuel	<b>2290</b>	88.9	Fuel	<b>2350</b>
$\text{C}_5\text{H}_8\text{O}$	2,5-Dihydro-2MF	20.5	n/a	n/a	10.7	n/a	n/a
$\text{C}_5\text{H}_8\text{O}$	3-Pentene-2-one	n/a	<b>516</b>	79.5	n/a	<b>240</b>	40.9
$\text{C}_7\text{H}_8$	Toluene	2.98	24.6	<b>43.1</b>	9.63	42.4	<b>45.9</b>
$\text{C}_6\text{H}_6\text{O}$	Phenol	6.15	29.0	<b>445</b>	9.70	25.5	<b>626</b>
$\text{C}_6\text{H}_7\text{O}$	2-(5-Methyl)furanylmethyl	n/a	n/a	<b>253</b>	n/a	n/a	<b>467</b>
$\text{C}_6\text{H}_8\text{O}$	2,5DMF	5.8	<b>259</b>	Fuel	5	<b>284</b>	Fuel

n/a: not applicable

**Table S2.**

Name and structure of some species involved in the detailed mechanism of the DMF oxidation.

Species	Name	Species	Name	Species	Name
	DMF		R8C <sub>6</sub> H <sub>7</sub> O		Furyl-2
	MF		R9C <sub>6</sub> H <sub>7</sub> O		Furyl-3
	Furan		M2C <sub>6</sub> H <sub>6</sub> O		C <sub>4</sub> H <sub>5</sub> O-3
	R1C <sub>6</sub> H <sub>7</sub> O		M3C <sub>6</sub> H <sub>6</sub> O		C <sub>4</sub> H <sub>5</sub> O-2
	R2C <sub>6</sub> H <sub>7</sub> O		B1C <sub>6</sub> H <sub>6</sub> O		C <sub>5</sub> H <sub>5</sub> #
	R3C <sub>6</sub> H <sub>7</sub> O		C <sub>6</sub> H <sub>5</sub> #		1,3-C <sub>5</sub> H <sub>6</sub>
	R4C <sub>6</sub> H <sub>7</sub> O		C <sub>6</sub> H <sub>5</sub> OH		C <sub>5</sub> H <sub>7</sub> Y
	R5C <sub>6</sub> H <sub>7</sub> O		Bicycle		C <sub>5</sub> H <sub>7</sub> #Y
	R6C <sub>6</sub> H <sub>7</sub> O		furylCH <sub>2</sub>		C <sub>5</sub> H <sub>7</sub> -1s
	R7C <sub>6</sub> H <sub>7</sub> O		M5F-2yl		C <sub>5</sub> H <sub>7</sub> #