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# STEAM EXPLOSION OF BEECH WOOD– EFFECT OF THE PARTICLE SIZE ON THE XYLANS RECOVERY

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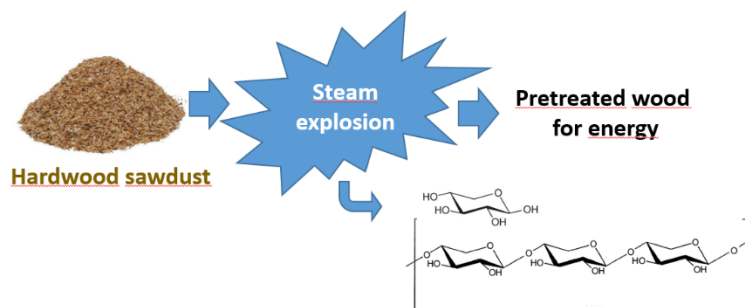
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## Abstract

In this work the effect of particle size and of the severity factors were investigated to find the optimum condition of steam explosion pretreatment on xylan recovery of beech wood. The beech wood particle with sizes of 0.16, 1, or 2 mm were steamed at 150°C-210°C for 2.5-15 minutes before the explosive decompression. The results showed that the maximum xylan recovery was about 10% w/w wood with low concentration of the inhibitors, which were obtained when the particle size is 1 mm and  $R_0 = 3.65$  (190°C, 10 min). The smaller particle size may result in overcooking of biomass, lead to easily and high degradation of hemicellulose sugars, whereas the larger particle size may result in incomplete autohydrolysis in biomass and lower extractability of hemicellulose sugars. The obtained optimum condition for xylan recovery will improve the subsequent utilization (such as in food industry and other chemical products), prior to subsequent transformation of steam explosion pretreated wood (bioethanol and pellet).

**Keywords:** Steam explosion pretreatment; beech wood; severity factor; biomass particle size; xylan recovery.

## Graphical abstract



## Statement of Novelty :

The steam explosion pretreatment of beech sawmill has been studied. The effect of hardwood chips particle size on the sugars recovery (xylose and xylans) in the liquid stream has been examined for the first time. The final goal is the improved valorization of the xylans from hardwood sawmill waste.

## 1. Introduction

The steam explosion (SE) process was developed by Mason in 1924 for the production of chipboard panels [1] and then improved for lignocellulosic materials disintegration used in the manufacture of paper, board pulp and the like [2]. The main equipment of this process is a steam generator, high-pressure reactor, an explosion tank to receive the material after explosion and a condenser [3]. SE is a pretreatment process to open up the biomass fibers. It uses high temperature and pressure steam for certain duration of times proceeded by a sudden pressure drop causing the disintegration of lignocellulosic structure and lower the bulk density of biomass [4-5]. SE is a thermo-mechanico-chemical pretreatment which induces a breakdown of lignocellulose by chemical modifications from in situ generated acids which cause the depolymerisation of hemicelluloses and by mechanical action due to high shearing forces applied to the biomass during the explosive release. This process improves the recovery of sugars (or other useful compounds) and so, makes the biopolymers more accessible for subsequent processes such as fermentation or densification. It can be used as a pretreatment process to produce solid biofuel pellets, to increase the calorific value, to improve the pelletizing properties of the biomass or to make cellulose more accessible to enzymes in order to convert it to fermentable simple sugars [6-7].

SE has been extensively described for pretreatment to enhance (1) the enzymatic hydrolysis of the resulting cellulosic pulp (poplar wood, sugar cane bagasse, pine wood, douglas wood, corn stover, and wheat bran [8-13] and (2) the properties of pellets from agricultural-based biomass, poplar wood, and douglas wood [4, 11-12]. In addition to the pretreated cellulosic pulp, high yields of hemicellulosic sugars are recovered in the water effluent of SE process. These sugars consist of monomers, oligomers and polymers which constitute attractive building blocks for further transformations. The kinetics of hydrolysis of corn cob and eucalyptus globulus xylans have been accurately described during hydrothermal treatments (150°C-190°C, few hours). However, the parametric factors affecting the hemicellulosic sugars recovery during the cooking step of a SE treatment are not fully understood. No clear correlation between temperature, residence time, biomass particle size and hemicellulosic sugars yields has been found in the literature.

Severity factor is a model broadly used for evaluating the process of pretreatment. It is based on the assumption that overall the pretreatment process is hydrolytic, and it follows a first order of kinetics process and obeys the Arrhenius law [14]. It combines the treatment temperature ( $T$ ) and residence time ( $t_s$ ) in one value.

During the cooking step of SE, hydronium ions from both water and in-situ generated compounds catalyze the hemicellulose depolymerization into oligomers and monomers (auto-hydrolysis process) [15-16]. Under harsh conditions (high temperature, low pH), monomeric sugars can be further degraded to furan derivatives (furfural from C5 and hydroxymethylfurfural from C6 sugars).

However, one of the limitations of the severity index model is the lacking particle size informations that have a major influence on the kinetics of the hydrolytic process. Particle size strongly affects the kinetics of the hydrolytic processes, the efficiency of the vapor soaking, the particles heat transfer and physical modifications of the biomass [12, 17-18]. Size reduction of biomass before a pretreatment is crucial for the optimization of sugars conversion but also strongly affect the milling power and the overall cost of the process [10].

Large particle size can hamper heat transfer and may result in incomplete auto-hydrolysis of its interior part whereas small particle may result in overcooking and in a degradation of its components. In the literature, the effect of chip size on SE process has been examined regarding

the sugars recovery starting from agricultural residues [12] and on softwood [10, 19]. It was concluded that larger particle produced higher enzymatic cellulosic pulp digestibility. The effect of the chips size on the sugars recovery and the furans production have also been examined during the SE of Douglas-fir [17]. To the best of our knowledge, the effect of hardwood chips particle size on SE performance and on sugars recovery in the liquid stream of the SE has never been examined.

European Beech (*Fagus sylvatica* L.) is one of the most abundant species and the most harvested hardwood in France. It is used to produce sawn wood with an average yield of 43-50% [20-21]. As a result, considerable amount of wood residues is produced which could constitute a potential source of biopolymers and/ bioenergy. Xylans are the main hemicellulosic components of beech wood. They constitute a valuable potential source for the production of high-added value ingredients for functional foods (xylo-oligosaccharides and/or xylitol from xylose) [22-23].

In the present study, the effect of the size of beech sawdust pretreated by varying experimental SE conditions (temperature, residence time) on the sugars conversion was examined in order to assess which SE conditions produce the maximum concentration of sugars (monomers and oligomers) in the water stream. The final goal is to improve the extraction of xylan of beech sawmill waste.

## **2. Material and methods**

### **2.1 Process flow chart**

Fig. 1 presents the different steps used to characterize beech wood (extractibles, ash, lignin, cellulose and hemicelluloses) and to process to beech wood pretreatment by using steam explosion (SE) process.

### **2.2 Sample preparation**

The wood used in this study was beech (*Fagus sylvatica* L.) from Lorraine forest. The wood samples were milled and sieved with three different sizes: 2 mm, 1 mm, and 0.160 mm. Then the sample were dried and stored at room temperature until steam explosion pretreatment.

### **2.3 Chemical composition**

**Extractives:** In a Soxhlet apparatus, 2 g dry sample with particle size 0.160 mm were extracted for 16 cycles with 300 mL of toluene and ethanol (2/1 ; v/v). The solvent was then evaporated under vacuum and the extracts were weighed, whereas the beech sample was dried with temperature 40°C for 24 hours. Extractives content were calculated by using these two-final masses.

**Holocellulose:** 1.5 g of beech free extractive samples with particle size 0.160 mm, 1 g of sodium chlorite ( $\text{NaClO}_2$ ), 1 mL acetic acid ( $\text{CH}_3\text{COOH}$ ) glacial and 125 mL ultrapure water (UPW) were mixed into a 250 mL flask. The flask was brought to reflux with continuous stirring for 2 hours at 70°C. Every 2 hours, 1 g  $\text{NaClO}_2$  and 1 mL  $\text{CH}_3\text{COOH}$  were added until the wood sample is completely white, meaning a total delignification. The solid fraction of holocellulose was then washed with ultrapure water then filtered, and finally dried at 40°C for 24 hours and then weighed. Holocellulose content was calculated by using the mass change.

**Cellulose:** 1 g of previous holocellulose was mixed with 50 mL of sodium hydroxide ( $\text{NaOH}$ -17,5%) in a 100 mL Erlenmeyer flask for 30 minutes at 25°C. Then, about 5 mL ultrapure water was added and stirred for 30 minutes at 25°C. The solid part, composed of cellulose, was filtered and washed with 50 mL of a solution of acetic acid (1% in volume). The solid was dried in oven at 40°C for 24 hours and weighed. Cellulose content was calculated by using the mass variation.

*Lignin*: 0.175 g of wood free extractive with particle size 0.160 mm was put inside a plastic centrifuge-tube. About 1.5 mL sulfuric acid ( $\text{H}_2\text{SO}_4$  - 72%) were mixed with the sample and then incubated in a rotary water bath for at 30°C for 1 hour. The sample was taken out and 42 mL of ultrapure water were added. The tube was then closed and put inside an autoclave for 1 hour 30 minutes. The solid, composed of insoluble lignin – also known as Klason lignin -, was filtered, washed, dried at 105°C for 24 h and finally weighed. The liquid phase, containing monomeric sugars from cellulose and hemicelluloses, was completed with ultrapure water to 100 mL and freeze-dried for further analyses. Lignin content was calculated by using the mass variation, whereas monomeric sugars were quantified by High-Performance Anion Exchange Chromatography coupled with Pulsed Amperometric Detection (HPAE-PAD).

*Ash*: 1 g of dry sample was weighed into a tared crucible and then put inside a muffle furnace (Carbolite WF-1100) with a temperature of 525°C for 24 hours. After 24 hours, the crucible was stored in a desiccator and weighed. The ash content was obtained by the decrease in mass.

*Oligomeric sugars post-hydrolysis*: To be quantified, oligomeric sugars contained in the liquid fraction after SE were first hydrolyzed to monomeric sugars. 10 mL of SE liquid phase were mixed in a plastic centrifuge-tube with 349  $\mu\text{L}$  of sulfuric acid ( $\text{H}_2\text{SO}_4$  - 72 %) and autoclaved at 121°C for 1h. After a subsequent dilution (200 times), the monosaccharide content of the sample was then determined by HPAE-PAD.

## 2.4 Chromatographic analyses

*HPAE-PAD analysis*: Monosaccharide contents of soluble fractions were analyzed by HPAE-PAD (ICS-3000 Dionex) equipped with a Dionex CarboPac<sup>TM</sup> PA-20 (3x150 mm) analytical column. Filtered samples (20  $\mu\text{L}$ ) were eluted at 35°C and at 0.4mL/min with the following composition: UPW 99,2% /250mM NaOH 0,8% : 0→20 min ; UPW 75%/250mM NaOH 20% /NaOAc (1M)- NaOH (20mM) 5% 20→37min ; UPW 40% /250mM NaOH 20%/NaOAc (1M)- NaOH (20mM) 40% 37→41min. Each elution was followed by a wash and subsequent equilibration time. External sugar and uronic acids standards were used for calibration (7 points per curve) : fucose, glucose, xylose, galactose, mannose, rhamnose, arabinose and galacturonic acid, glucuronic acid (all provided by Sigma-Aldrich).

*High-Performance Liquid Chromatography (HPLC) analysis*: During SE, acetic acid can be released from the wood because of the partial hydrolysis of its cell wall components; whereas furfural and 5-hydroxymethylfurfural (HMF) can be produced respectively from pentose sugars and hexose sugars by a dehydration process. At last formic and levulinic acids are resulting from both furans degradation.

All these “by-products” were analyzed by HPLC (Ultimate 3000) using a Rezex<sup>TM</sup> RHM-Monosaccharide  $\text{H}^+$  (8%) column with sulfuric acid ( $\text{H}_2\text{SO}_4$  - 5.10-3 M) as the mobile phase with an isocratic flow rate of 0,8 mL/min and a temperature of 65°C. The wavelength for detecting the three acid compounds was 210 nm, whereas it was 280 nm for furfural and HMF. Each elution was followed by a wash and subsequent equilibration time. External standards provided by Sigma-Aldrich were used for calibration (5 points per curve).

## 2.5 Steam explosion pretreatment

The wood samples (1 mm, 2 mm, and 0.160 mm) were impregnated in the ultrapure water for 24 hours at the room temperature beforehand; with a ratio between water and wood sample of 20:1 (v/w). After the sample was soaked, it was filtered; the filtered-water was stored for the analysis of sugar content. The framework of steam explosion process is shown in Fig. 2; the apparatus is composed of three main parts : a vapor generator, a reactor and a discharge tank.

The 20 g of beech wood sample was loaded into the reactor and exposed to the several high-pressure steams [3,7 bar – 18,1 bar] for several residence times [2 min-15 min] and temperatures [150°C-210°C]. The samples were then exploded by sudden-dropped in pressure and released in the discharge tank. The exploded solid biomass was separated from the liquid fraction by vacuum filtration. It was dried at 40°C for 48 hours, whereas the volume of liquid fraction was measured; only a small part was frozen until it was analyzed for monomeric and oligomeric sugars content and other byproducts.

*Severity factor:*

$$R_0 = \int_0^{t_s} \exp\left(\frac{T - 100}{14.75}\right) dt$$

Where:

Ro = Severity factor

Tr = Temperature of reaction (°C)

Tb = Base temperature (100°C)

ts = Residence time (min)

14,75 = Conventional activation energy with assumption overall process is hydrolytic, follows the first order of kinetics process, and obey Arrhenius law.

### 3. Results and Discussion

#### 3.1 Chemical composition

The untreated beech wood used in this study comprises 37.8 % cellulose, 35.7 % hemicellulose, 24.5 % Klason lignin, 1.8 % extractives (solubility in toluene and ethanol; 2/1 ; v/v, respectively) and 0.2% ash. Compared to some reports from literature, the sample has lower amount of cellulose and higher or comparable content of lignin [24–27].

#### 3.2 Overall mass loss to severity factor

The SE experimental conditions used were as follows: temperature from 150°C to 210°C, residence time from 2.5 min to 15 min, sawdust sizes of 0.16 mm, 1 mm, and 2 mm. The corresponding severity factors are gathered in Table 1. As seen in Figure 3, the mass loss was increasing with the severity factor. It appears that below a severity factor of 3.0, the mass loss was varying very slightly whatever the size of the particle. For  $R_0 > 3.0$ , dramatic increase of the mass loss was observed. At the highest severity ( $R_0 = 4.41$ ), ~ 40% of the wood was solubilized. This result is in accordance with previous works performed starting with hardwood chips using both continuous [28] or batch [28-29] SE systems which showed that a significant degree of solubilization of wood requires treatment severity beyond  $R_0 = 3.0$  with a maximum (~ 30% of mass loss) for  $R_0 > 3.75$ . From Fig. 3 it can be seen that the particle sizes impact the mass loss. The larger particles exhibited a significantly lower degree of wood solubilization.

#### 3.3 Xylan and glucan yields

Figure 4A, B and C give respectively the total monosaccharides yields, the monomeric glucose and the monomeric xylose released in the liquid phase during the steam explosion process as a function of the severity factors. As expected, the sugars yields recovered in the liquid phase follow the same trend as the mass losses (Fig. 3). For  $R_0 < 3.0$ , the sugars yields were very low and slightly increase with  $R_0$ . For  $R_0 > 3.0$ , an increase is observed.

As expected, the xylose concentrations (Fig 4C) are higher than those of the other sugars, xylan being the dominant hemicellulose of beech. For instance, with a severity factor of 4.41, xylose represents about 75% of the sugars released whatever the sawdust size. Its yield was increasing

drastically when the  $R_0 > 3.0$ , confirming that hydrolysis of hemicelluloses is more efficient when increasing severity factor. Regarding glucose content, its value is very low compared to the total yield of monosaccharides. Glucose yields are more difficult to interpret because glucose is produced by the hydrolysis of both hemicellulose and cellulose. The low glucose content in the hydrolysates showed that cellulose was not or few affected by steam explosion process in our conditions.

Interestingly, higher monosaccharides content is recovered in solution for the intermediate particle size (1 mm); for larger (2 mm) or smaller (0.16 mm) particles lower sugars concentrations were detected. This result is in agreement with previous studies performed on softwood (Douglas-Fir), in which fine particles (0.42 mm) released fewer sugars than woodchips [17].

The yields of the oligomeric total sugars, glucans and xylans released during the SE process in the liquid phase are shown in Fig. 5A, B and C respectively. Fig. 5B and C confirm that xylans are the dominant polysaccharides (> 77% of total sugars). Compared to monosaccharides, the oligomeric contents are higher and the shape of the curves are different; the oligosaccharides are released at lower  $R_0$  (from  $R_0 = 2.5$ ) and reach a maximum at  $R_0 = 3.65$  (190°C, 10 minutes). At that severity, the amount of extracted oligosaccharides was 10g/100 g of dry wood for xylans and 0.8g/100 g of dry wood for glucans.

A further increase in severity led to a decrease in the recovery of oligosaccharides. This can be explained by the kinetic pathway in which oligomers are released first during the steam treatment and then are further hydrolyzed into monosaccharides [16, 31]. Very similar results were previously reported for the non-catalyzed SE of poplar using a continuous 4t/h pilot plant [28]. These authors described a maximum pentosans recovery (monomers and oligomers) of 65% at  $R_0 = 3.8$ . In contrast, using acid-catalyzed SE, pentosans and hexosans were primarily recovered as monosaccharides. At  $R_0 = 3.03$ , less than 20% of xylans were found as oligomers [32].

Regarding the influence of the particle size, as it was previously observed for monomeric sugars, the oligomeric recovery in the liquid phase is higher starting from a sawdust size of 1 mm.

### 3.4 Furfural and 5-HMF yields

During the cooking step of SE process, the increase of severity led to an increasing content of hemicelluloses solubilized and to an increasing dehydration rate of sugars derived from hemicellulose, producing byproducts such as furfural and 5-hydroxymethylfurfural (HMF) [33]. In a biomass-to-biofuel process, furfural and HMF might affect enzymes in hydrolysis process and reduce glucose conversion during fermentation process [34]. As shown in Fig. 6, the concentration of HMF and furfural in the SE liquid effluents were undetectable or very low for  $R_0 < 3.5$  and increased above this value. For the highest severity ( $R_0 = 4.41$ ) and the smallest particle size (0.16 mm), maximum concentration of HMF (0.2% w/w) and furfural (0.7% w/w) were detected. Stoffel et al (2017) reported comparable trends but the generation of higher levels of degradation products during the SE of pine sawdust [32]. In this study the higher furans content in the hydrolysates can be rationalized by the utilization of 1-3% of sulfuric acid as catalyst which catalysis the sugars dehydration reactions.

The impact of the wood particle size on the production of degradation products is clearly shown in Fig. 6 the smaller the particle, the higher the furfural and HMF concentration. Similar results were described by Cullis et al. (2004) who observed during the SE of softwood a substantial decrease of the production of furfural and HMF as the wood chip size increased (from 1.5 cm to 5 cm) [17]. Overcooking of the small particles during the cooking step of SE may promote hemicelluloses depolymerization and degradation. Surprisingly, Liu et al. (2013) reported the

opposite trend for the SE of corn stover : the amount of furan derivatives was higher and the sugars recoveries were lower for larger biomass particules [12].

#### 4. Conclusion

The optimization of pretreatment of beech wood by uncatalyzed SE for hemicellulose recovery in the forms of monomeric and oligomeric is as a function of the severity factor and particle size. The maximum of oligomers recovery was obtained when the particle size is 1 mm and  $R_0 = 3.65$ , 10% w/w wood with low concentration of the inhibitors. The particle size is a very important factor in the hemicellulose recovery and in this work the optimum size was found to be 1 mm. Smaller particle size (0.16 mm) resulted in overcooking of biomass, leading to higher degradation of hemicellulose sugars, whereas larger particle size (2 mm) resulted in incomplete autohydrolysis in biomass and lower extractability of hemicellulose sugars. The optimum condition for xylan recovery will optimize the subsequent utilization (such as in food industry and other chemical products), prior to subsequent transformations of SE pretreated wood (bioethanol and pellet).

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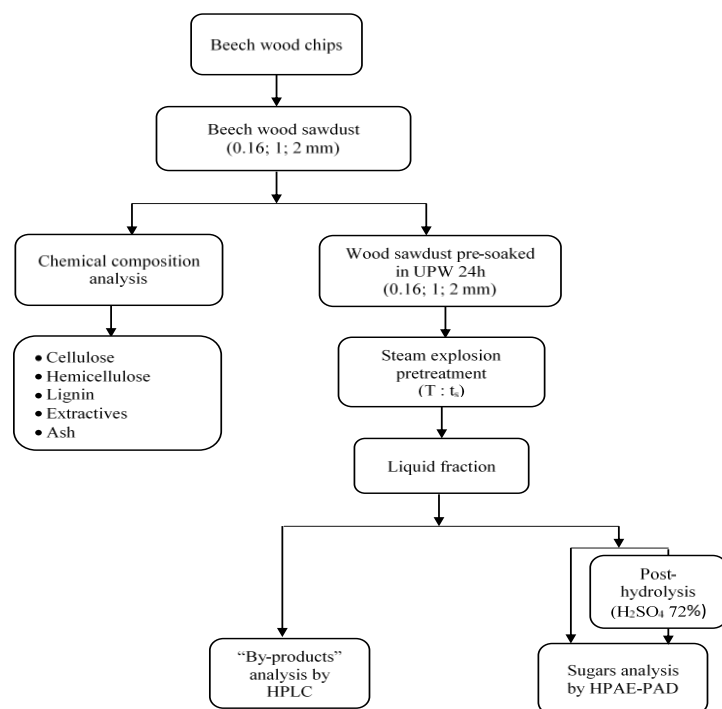
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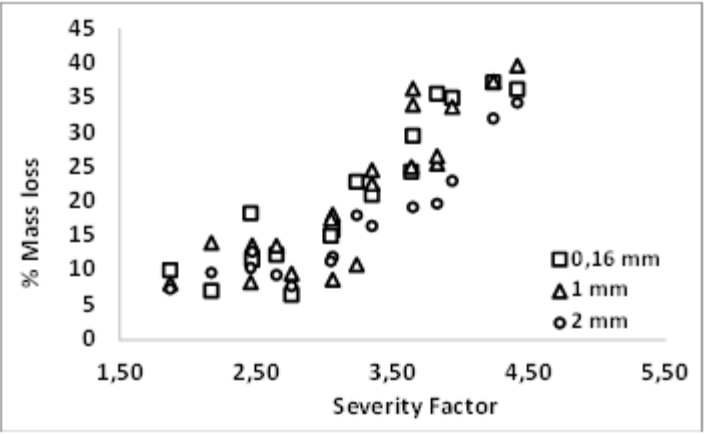
$t_s$ (min) \ T(°C)	150	170	190	210
2.5	1.87	2.46	3.05	3.64
5	2.17	2.76	3.35	3.94
10	2.47	3.06	3.65	4.24
15	2.65	3.24	3.83	4.41

**Table 1** Severity factor ( $R_0$ ) values  
Software used: Microsoft Word



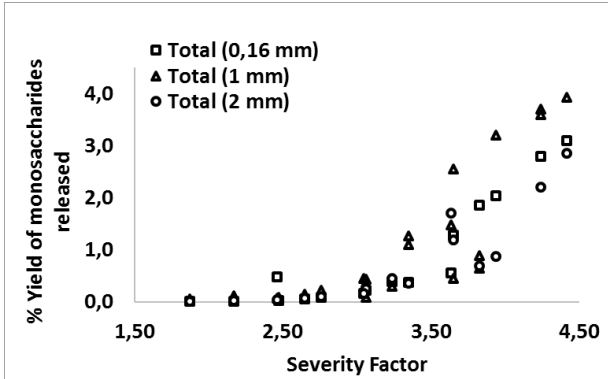
**Figure 1** Flowchart of beech sawdust pretreatment.  
Software used: Microsoft Word

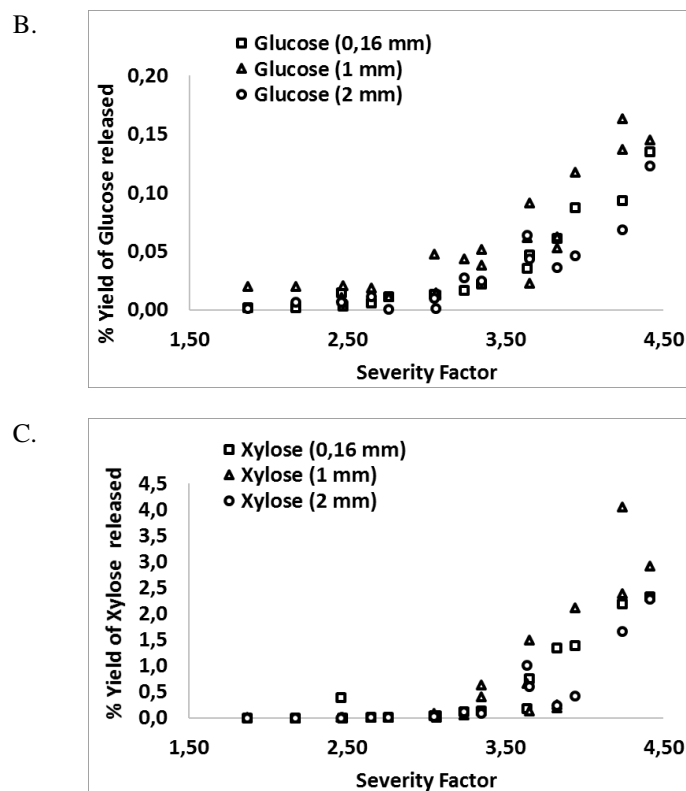




**Figure 3** Mass loss vs  $R_0$ .  
Software used: Microsoft Excel

A.

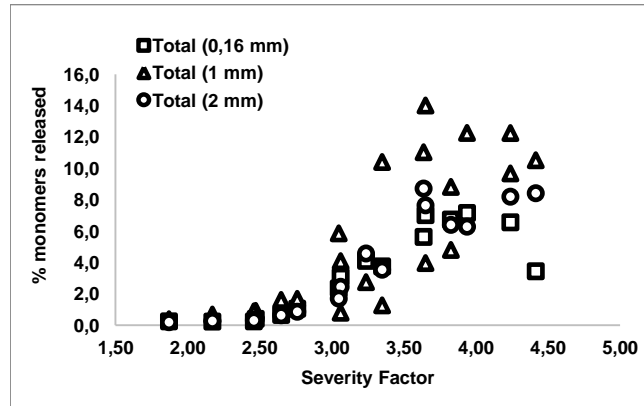




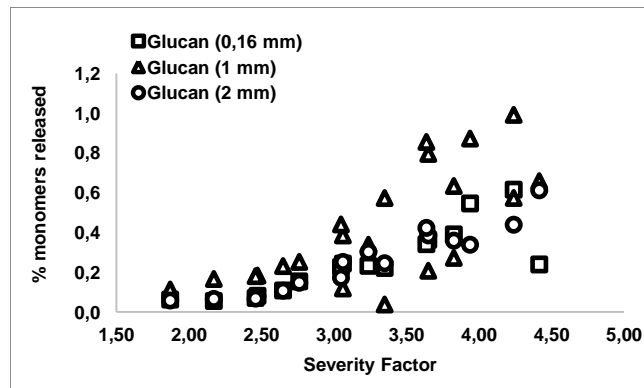
**Figure 4** Yield of monosaccharides released in the liquid phase vs  $R_0$ . A) the total monosaccharides yields; B) the monomeric glucose; C) the monomeric xylose. Software used: Microsoft Excel



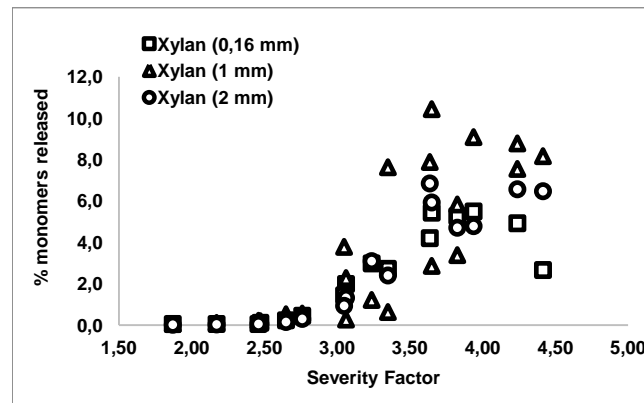
A.



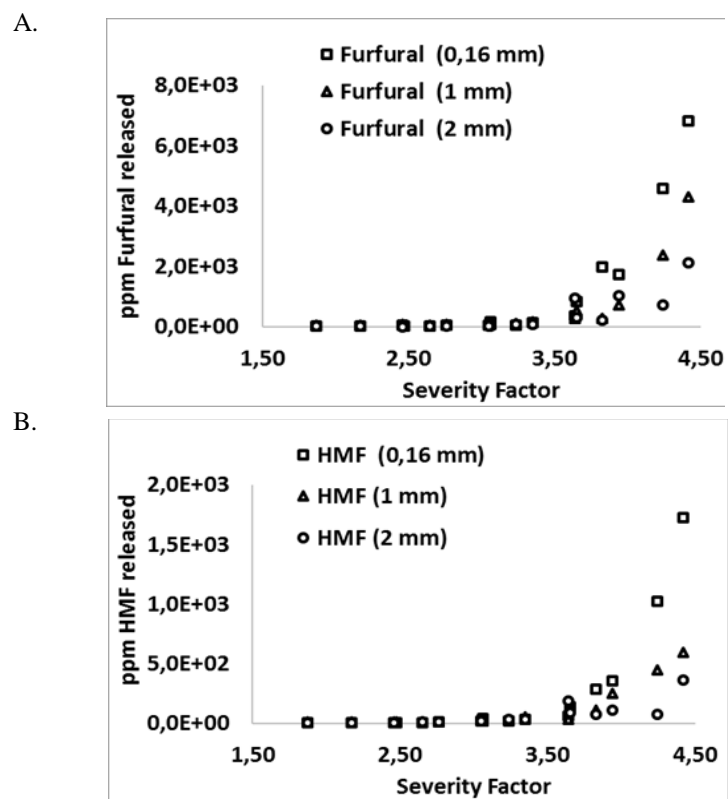
B.



C.



**Figure 5** Yield of oligosaccharides released in the liquid phase vs  $R_0$ . A) the oligomeric total sugars; B) the oligomeric glucans; C) the oligomeric xylans.  
Software used: Microsoft Excel



**Figure 6** Yield of A) furfural and B) HMF released in the liquid phase vs  $R_0$ .  
Software used: Microsoft Excel

