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Constraints on the loss of matrix-sited helium during vacuum crushing of mafic phenocrysts

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Abstract

Vacuum crushing is an efficient technique to selectively release the mantle-derived helium component trapped within olivine and pyroxene phenocrysts. However, contrary to previous assumptions, recent studies have shown that this method may liberate significant matrix-sited cosmogenic ³He (³He_c) or radiogenic ⁴He (⁴He^{*}). Because this loss may bias both the determination of magmatic ³He/⁴He ratios and the accuracy of ³He_c measurements, it is essential to understand what mechanism is responsible and under what conditions matrix helium loss is manifest. To address this question, olivines and pyroxenes with various amounts of matrix-sited 3 He (from 10^{7} to 10^{11} at. g^{-1}) were crushed in air or in vacuum using several crushing devices. Sample temperature was controlled during each crushing experiment, and ranged from 25 to 325 °C. The resulting powders were then sieved to obtain several homogeneous grain fractions ranging between <10 and >300 μm. The ³He_c concentrations measured in each fraction clearly show that significant ³He_c loss (>20%) affects only the finest fraction (<10 μm) and, importantly, only under hot conditions (here $T \ge 300$ °C). Even the smallest fractions (<10 μ m) quantitatively retain matrix-sited 3 He_c when crushed under cold conditions ($T \le 25$ °C), regardless of the duration and energy of crushing. These results invalidate the model previously proposed by (Yokochi R., Marty B., Pik R. and Burnard P. (2005) High ³He/⁴He ratios in peridotite xenoliths from SW Japan revisited: evidence for cosmogenic ³He released by vacuum crushing. Geochem. Geophys. Geosyst. 6, doi:10.1029/2004GC000836) that involved spallation tracks and implied that the magnitude of loss was mainly controlled by the grain size. Moreover, new diffusion experiments were carried out to constrain the diffusivity of matrix-sited helium in crushed olivines. When used to model diffusive ³He_c loss as a function of grain size during crushing, these new data predict the observed release fairly well. Therefore, we conclude that temperature-enhanced volume diffusion is one of the main mechanisms controlling the release of ³He_c during crushing. For future applications, special attention should thus be paid to control the grain size, the crushing duration, and the temperature of the sample. © 2008 Elsevier Ltd. All rights reserved.

1. INTRODUCTION

Helium isotopes trapped in inclusions in mafic phenocrysts are a very useful geochemical proxy for characterizing mantle sources (e.g., Kurz et al., 1982; Stuart et al., 2003). In addition, cosmogenic ³He in the matrix of mafic phases (Kurz, 1986a) is a useful tool for understanding Earth-surface processes, not only because its nuclear stability permits

quantitative constraints over several million years of exposure (Blard et al., 2006a), but also because its low detection limit compared to its high production rate allow quantification of young exposure events (<10 ka) (e.g., Kurz et al., 1990). However, accurate determination of magmatic helium isotopic ratios and cosmogenic ³He (³He_c) abundances require very specific analytical conditions to isolate each of these helium components. In the case of olivines and pyroxenes, which are the most common minerals used for such applications, vacuum crushing has been widely adopted because of its supposed ability to selectively release the magmatic helium contained within fluid and melt inclusions,

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without extracting matrix-sited helium (Kurz, 1986b). However, recent studies have demonstrated that this assumption is not always valid and that prolonged crushing may cause the extraction of a significant amount of matrix-sited helium (Scarsi, 2000). The demonstration that such loss can occur has led (Yokochi et al., 2005) and (Moreira and Madureira, 2005) to revisit several determinations of magmatic helium isotopic ratios. It also caused (Blard et al., 2006b) to question the accuracy of certain cosmogenic ³He measurements.

Correction for magmatic ³He is usually performed according to the equation (Kurz, 1986a):

$${}^{3}\mathrm{He_{c}} = {}^{3}\mathrm{He_{fusion}} - \left({}^{3}\mathrm{He}/{}^{4}\mathrm{He}\right)_{\mathrm{crush}} \times {}^{4}\mathrm{He_{fusion}} \tag{1}$$

where ³He_{fusion} and ⁴He_{fusion} are the concentrations measured by vacuum fusion of the phenocrysts and (³He/ ⁴He)_{crush} is the helium isotopic ratio determined by vacuum crushing. Consequently, extraction of a significant amount of cosmogenic ³He during the crushing step may induce overestimation of the fluid inclusion ³He/⁴He ratio. If ignored, this partial ³He_c release may also lead to underestimation of the actual ³He_c concentration, as a result of the combined effects of (i) inaccurately high measured crushing ³He/⁴He ratios and (ii) unquantified loss of the isotope of interest (³He_c). Moreover, the mechanisms controlling the release of matrix-sited ³He_c are not yet understood and this uncertainty hampers the development of alternative strategies to avoid matrix helium loss during crushing. Several recent studies (Moreira and Madureira, 2005; Yokochi et al., 2005; Blard et al., 2006b) proposed that the crushing release of ³He_c was controlled either by (i) a mechanism involving spallation tracks, that may provide an efficient pathway for helium extraction when tracks encounter newly created fractures or (ii) helium release at the fractured surface due to lattice distortion caused by plastic deformation of the grain during crushing. Both proposed mechanisms are strongly dependent on granulometry and are supposed to be efficient even at low temperature. Moreira and Madureira (2005) and Yokochi et al. (2005) ruled out the possibility that volume diffusion was the main mechanism responsible of crushing loss, basing their conclusions on the observed range of temperatures reached during crushing (T <100 °C) and the diffusion data of (Trull et al., 1991). However these hypotheses still need definitive testing.

In this study, we present results from new experiments designed to address these issues and to establish the mechanisms controlling release of matrix-sited helium during vacuum crushing. We carried out several crushing experiments, both in vacuum and in air, using different crushing devices. In particular, the temperature of the samples was controlled during crushing. After crushing, ³He_c was determined on samples that were split into several size fractions from <10 to >300 µm. These experiments reveal that significant loss (>20%) only affects the finest fraction (<10 μm) and, importantly, only under hot conditions during crushing (here the experiment was performed at ≥ 300 °C). The present study also provides new helium diffusivity data using proton-irradiated olivines (Shuster et al., 2004). As discussed in Section 4, these results provide support that temperature-dependent diffusion may control the release of matrix-sited ³He_c during vacuum crushing.

2. METHODS

2.1. Sample description

2.1.1. Geological samples with natural spallation-induced ${}^{3}He_{c}$

The studied mafic phenocrysts come from two different moraine boulders exposed to cosmic rays since the Late Pleistocene glacial retreat, but with distinct geological settings and exposure histories.

Sample MK9 is a cobble of 20 cm diameter that was collected on a ~15-ka-old recessional moraine deposited at 3570 m on the southern slope of Mauna Kea volcano (Hawaii, 19°N) (Blard et al., 2007). This cobble is an alkali basalt that is very rich (>10 %) in coarse olivine (Fo₈₃) and clinopyroxene (diopside) phenocrysts (average diameter \sim 5 mm). Both minerals have fluid and melt inclusions ranging in size between 5 and 40 µm. Given that the eruption age of this basalt is ~100 ka and its U and Th contents are, respectively, lower than 0.3 and 0.8 ppm (Wolfe et al., 1997), the concentration of the ingrown radiogenic ⁴He is very low within the phenocrysts ($<3 \times 10^9$ at. g⁻¹). A previous measurement (Blard et al., 2007) established that the cosmogenic ³He content of MK9D phenocrysts is 8.7 ± 0.3 and $8.5 \pm 0.3 \times 10^6$ at. g⁻¹ for olivines and pyroxenes, respectively. Therefore, even if radiogenic ⁴He was neglected, this would not bias the corrected ³He_c by more than 0.5%. The helium in these phenocrysts can thus be considered as a two component system: ³He is only cosmogenic and magmatic and ⁴He is only magmatic, and the use of Eq. (1) is therefore appropriate. Moreover, because magmatic 4 He is quite low in these phenocrysts ($<3 \times 10^{10}$ at. g $^{-1}$), correction for magmatic 3 He is not more than 5% of the total ³He_c and does not represent a significant source of uncertainty in the published experiment (Blard et al., 2007). Consequently, the total uncertainty associated with the ³He_c determination should be approximately the total reproducibility of the analytical system, i.e. few % (Farley et al., 2006). This point is important given that the experiments described in the following section aim to detect variations in ³He concentrations of as little as 10%. Because several grams of pure MK9D olivines and clinopyroxenes are available, this sample is ideal for the multiple crushing experiments carried out here on separate mineral phases.

Sample TU8 was collected on the flat surface at the top of a boulder (~2 m in height) located at 3750 m on an 18- to 15-ka terminal fluvio-glacial moraine (Blard et al., 2008), on the southern flank of Cerro Tunupa, which borders the Salar de Uyuni (Bolivian Altiplano, 20°S). This rock is an andesite bearing <1 mm phenocrysts of clinopyroxene (diopside composition: (Ca0.83,Na0.04)(Mg0.82,Fe0.29, Al0.06)(SiO₃)₂). These phenocrysts have melt inclusions of 10–40 µm diameter, but they appear to be almost free of fluid inclusions. Although the eruption age of this andesite is not precisely known, the emplacement of the Tunupa volcano is younger than 5 Ma (Villeneuve et al., 2002). Thus, over this time period, given the Li concentration measured in these diopsides (<60 ppm, Blard et al., 2008), nucleogenic ³He (³He_n) (Andrews and Kay, 1982; Andrews, 1985) is less

than 5% of the ${}^{3}\text{He}_{c}$ concentration of $1.88 \pm 0.04 \times$ 10⁷ at. g⁻¹that has been measured in the TU8 pyroxenes (Blard et al., 2008). This estimate takes into account the proportion of ³He_n that is produced in the andesite matrix and implanted in these $\sim 300 \, \mu m$ grains (Farley et al., 2006). Given the helium closure age of these minerals (≤ 5 Ma), as well as the U and Th concentration in these andesites (U ≤ 2 ppm and Th ≤ 10 ppm, Blard et al., 2008) and the diposide-lava partition coefficient of U-Th (from 0.11 to 0.15, Blard et al., 2008), most of the ⁴He extracted by fusing TU8 clinopyroxenes ($5 \pm 0.5 \times 10^{12}$ at. g⁻¹) results from radiogenic ingrowth. Preliminary results of vacuum crushing (Blard et al., 2008) also revealed that the total magmatic ⁴He concentration is about 6×10^{10} at. g⁻¹. If we make the reasonable assumption that the uncertainty of this measured magmatic ⁴He concentration is 100% and that the trapped ³He/⁴He ratio is 8 Ra, then the magmatic ³He correction (6.7 \pm 6.7 \times 10^{5} at. $g^{-1})$ does not represent more than 1% of the total ${}^{3}\text{He}_{c}$ concentration (1.84 \pm 0.04 \times 10^7 at. g^{-1}). This consideration is important in the case of TU8 pyroxenes, because it implies that, despite this three component mixture, the magmatic ³He correction does not represent a significant source of uncertainty. Moreover, preliminary results (Blard et al., 2008) obtained by fusing several aliquots of unbroken phenocrysts with variable grain size (from 0.2-0.3 to 0.4-0.5 mm) yielded similar ³He concentrations for all aliquots, while different grain sizes are likely to have contrasting magmatic helium contents. This observation supports the assumption of a very low magmatic helium correction. Consequently, we will use in the following discussion a constant magmatic 3 He correction of $6.7 \pm 6.7 \times 10^5$ at g^{-1} for each TU8 aliquot. TU8 pyroxenes are thus suited for this study and will yield complementary informations to that obtained from MK9D olivines and pyroxenes. This will allow an independent test of the inter-sample reproducibility of our conclusions.

2.1.2. Proton-irradiated olivines

Proton-irradiated olivines were also used for several of our experiments because they have concentrations of matrix-sited ³He about 10³ times above the natural samples MK9D and TU8. These samples are thus useful to (i) place limits on the fraction of helium released by vacuum crushing and (ii) perform new helium diffusion experiments using crushed minerals. The new diffusion data will allow us to assess if the crushing modifies helium diffusivity. One of the samples selected for irradiation is a gemquality Fo₇₅-Fo₈₀ olivine from a Guadalupe Island dunite (Batiza, 1977). The selected fragment was almost free of fluid inclusions and analyses of other olivines from this area indicate that the magmatic helium content of this material is several orders of magnitude lower than the proton-induced ³He concentration (Shuster et al., 2004). This ensures that the magmatic ³He, which is contained within fluid inclusions, cannot bias the diffusion data obtained from the release pattern of the matrix-sited proton-induced ³He. The other irradiated olivine sample is a gem-quality Fo₉₂ crystal from xenoliths of San Carlos, Arizona (Frey and Prinz, 1976).

The Guadalupe olivine was irradiated in April 2002 under a fluence of 3.2×10^{14} protons cm⁻² over a 10-h period with a 147-MeV proton beam generated by a synchrocyclotron at Harvard Cyclotron Laboratory following the analytical conditions previously described in (Shuster et al., 2004). By contrast, the San Carlos olivine sample was irradiated in May 2007, under a fluence of 4.6×10^{15} protons cm⁻² with an energy of 220 MeV at the Francis H. Burr Proton Therapy Center of Massachusetts General Hospital.

2.2. Experimental techniques

2.2.1. Crushing experiments

2.2.1.1. Vacuum crushing. The purity of each aliquot of separated phenocrysts was carefully checked under a binocular microscope. Special attention was paid to remove any grains with adhering matrix. The starting size fraction of MK9D olivines and pyroxenes was 0.5-1 mm. Because this grain size is smaller than the initial diameter of the original grains (\sim 5 mm), it is likely that some of the fluid inclusions were broken, releasing a fraction of their total magmatic helium. The 0.3- to 0.5-mm fraction was processed in the case of the TU8 pyroxenes. In this diameter range, most of the grains were unbroken euhedral phenocrysts. Each phenocryst aliquot was weighed before being loaded in the crushing device (masses are listed in Table 1). Typical sample mass was \sim 1 g for samples with only natural helium (MK9D and TU8). Given the high amounts of proton-induced ³He, a small (69 mg) single fragment of the San Carlos olivine was loaded in the crushing tube.

Online crushing was performed using stainless steel tubes in which magnetic slugs of \sim 64 g are activated by three external solenoids. The slug used at Caltech is partially hollowed out over about 1/4 of the cylinder, while the slug used at CRPG (Nancy) is a full steel cylinder. The crushing frequency was ~60 strokes per minute, with a delivered energy of ~0.4 J per stroke. Because temperature is a crucial parameter controlling volume diffusion, the initial apparatus was slightly modified. For this, the tube was wrapped with a resistance heater tape and a thermocouple was attached at the bottom of the tube, where the sample resides. This design permitted us to measure $(\pm 10 \, ^{\circ}\text{C})$ and adjust (up to 325 $^{\circ}\text{C}$) the temperature of the sample during crushing. While TU8 and San Carlos olivines were crushed under cold conditions ($T \sim 25$ °C), MK9D phenocrysts (both olivines and pyroxenes) were crushed at different temperatures (from 25 up to 325 °C). Sequential crushing (steps of 5, 10 and 15 min) was carried out to observe the evolution of ³He and ⁴He release through time. The helium released after each crushing step was purified over SAES getters, cryofocussed and separated from Ne according to the procedure used at Caltech (Patterson and Farley, 1998). Then, helium was inlet into a MAP 215-50 mass spectrometer and the two isotopes analyzed by peak-jumping, ³He being counted on a pulse-counting electron multiplier and ⁴He on a Faraday cup. Abundances were determined by peak height comparison with gas standards of known composition. Linearity of ³He sensitivity

Table 1 Helium data from in vacuo crushing experim

Sample	Mass (g)	Crushing time	Temperature	³ He	⁴ He	³ He/ ⁴ He	⁴ He flux	Extracted	Fraction of
	(8)	(min)	(°C)	$(10^5 \text{ at. g}^{-1})$	$(10^{10} \text{ at. g}^{-1})$	(Ra)	$(10^9 \text{ at. g}^{-1} \text{ min}^{-1})$	${}^{3}\text{He}_{\text{cex}}{}^{a} (10^{5} \text{ at. g}^{-1})$	extracted ³ He _c (%)
MK9D olivines	1.0	5	25	7.1 ± 0.2	6.1 ± 0.1	8.4 ± 0.3	12.3 ± 0.1	< 0.22	0.0
0.5-1 mm		10	25	1.7 ± 0.2	1.4 ± 0.1	8.5 ± 1.0	1.4 ± 0.1	0.03 ± 0.2	0.0
		15	25	0.4 ± 0.2	0.3 ± 0.1	8.7 ± 4.7	0.3 ± 0.1	0.02 ± 0.2	0.0
Total		30		9.1 ± 0.3	7.9 ± 0.2			0.04 ± 0.4	0.1
MK9D olivines	1.4	5	300	5.8 ± 0.1	4.9 ± 0.1	8.5 ± 0.2	9.8 ± 0.1	0.1 ± 0.1	0.1
0.5–1 mm		10	300	2.7 ± 0.1	1.8 ± 0.1	10.9 ± 0.5	1.8 ± 0.1	0.6 ± 0.1	0.8
		15	325	3.4 ± 0.1	1.2 ± 0.1	20.3 ± 0.9	0.8 ± 0.1	2.0 ± 0.1	2.4
Total		30		11.8 ± 0.2	7.9 ± 0.2			2.7 ± 0.2	3.3
MK9D pyroxenes	0.9	5	300	1.3 ± 0.2	1.4 ± 0.1	6.5 ± 0.9	2.8 ± 0.1	< 0.25	0.0
0.5–1 mm		10	300	1.5 ± 0.2	0.5 ± 0.1	24 ± 3	0.5 ± 0.1	1.1 ± 0.2	1.4
		15	300	2.2 ± 0.2	0.3 ± 0.1	59 ± 9	0.3 ± 0.1	2.0 ± 0.3	2.5
		15	325	3.5 ± 0.2	0.1 ± 0.1	202 ± 57	0.1 ± 0.1	3.4 ± 0.3	4.3
Total		45		8.6 ± 0.4	2.3 ± 0.2			6.5 ± 0.4	8.1
TU8 pyroxenes 0.3-0.5 mm	0.6	0.5	25	0.1 ± 0.1	0.5 ± 0.1	0.8 ± 1.2	11.0 ± 0.6	< 0.24	0.0
		3	25	0.3 ± 0.1	1.4 ± 0.1	1.5 ± 0.5	4.6 ± 0.1	0.1 ± 0.1	0.1
		10	25	1.30 ± 0.1	4.7 ± 0.1	2.0 ± 0.1	4.7 ± 0.1	0.8 ± 0.2	0.4
Total		13.5		1.6 ± 0.2	6.7 ± 0.2			0.9 ± 0.3	0.5
Irradiated	0.069	5	25	$18,000 \pm 500$	1.7 ± 0.1		0.36 ± 0.01		0.5
San Carlos		10	25	$20,100 \pm 600$	2.1 ± 0.2		0.40 ± 0.01		0.5
olivine 1 fragment		10	25	$30,\!800\pm900$	2.8 ± 0.2		0.31 ± 0.01		0.8
Total		25		$68,900 \pm 1200$	6.6 ± 0.4				1.8

Crushing frequency is about 60 strokes per minute. $^{\rm a}$ $^{\rm 3}$ He_{cex} is calculated from Eq. (3).

was established by adjusting the standard size, so that the amount of ⁴He in the mass spectrometer is similar for samples and standards (Burnard and Farley, 2000). Sensitivities were typically $\sim 1.7 \times 10^{-5}$ cps at.⁻¹ and $\sim 3 \times 10^{-7}$ mV at.⁻¹ for ³He and ⁴He, respectively. Uncertainties for the measured ³He and ⁴He concentrations included the internal reproducibility (<1%), the external reproducibility computed from the analysis of standards (3% for ³He and 1% for ⁴He), as well as the error attached to the blank correction. Because these sources of error are independent, a quadratic sum was made to calculate the total analytical uncertainty. All these uncertainties are here given at the 1σ level. The crushing apparatus is designed to degas the tube activating the slug before dropping the samples. This capability permits measurement (and lowering of) the blanks before each analysis. ³He and ⁴He blanks were $3.6 \pm 0.6 \times 10^3$ and $3 \pm 1 \times 10^8$ atoms per minute, respectively. Because this background was not temperaturedependent but only time-dependent, similar blank corrections were used for both cold and hot experiments. ³He blanks were between 4% and 34% of the gas extracted from the samples and ⁴He blanks between 2% and 65% of the sample. The uncertainties associated with these blank estimates are 16% and 40% for ³He and ⁴He, respectively, and this correction is thus the main source of uncertainty for these crushing analyses.

The characteristics (time and temperature) of these online crushing experiments are given in Table 2 for each sample. These results are used to assess the mechanisms controlling the release of the different helium components.

2.2.1.2. Crushing in an agate mortar. An aliquot of \sim 1 g MK9D olivines was crushed in air, using an agate mortar and pestle. This was done in ethanol, to ensure that the sample was kept at low temperature (\sim 25 °C) and to inhibit grain ejection. The mechanisms involved in this kind of crushing are different (time of impact is longer, experiment is done in air) from those within the vacuum crushing device. This experiment was therefore carried out as a comparative test, in order to check if the 3 He_c release is sensitive to the crushing conditions.

2.2.2. Measurement of cosmogenic ³He concentrations

Results from previous studies (Yokochi et al., 2005; Blard et al., 2006b) suggest that grain size is an important parameter controlling matrix-sited ³He_c release. Consequently, to assess grain size control on ³He loss, the crushed samples were sieved into different granulometric fractions (from <10 to >190 μm, see Table 2 for details). Fine (<50 μm) steel chips originating from the crusher were removed from the phenocrysts using a hand magnet, to minimize contamination. Indeed, this metallic component, together with possible contamination from previous samples, may have very high helium content (more than 10^{14} ⁴He at. g⁻¹, the ³He/⁴He ratio being <0.02 Ra). After sieving, each fraction was carefully checked under a binocular microscope, to ensure that grains were not constituted of agglomerated smaller grains. The ³He_c content of each fraction was then measured by degassing in a double-walled vacuum furnace. Extraction temperature was set at 1500–1600 °C for 20 min. For each sample, complete extraction yield was verified by a second heating step, which always yielded blank level helium. Measurement of $^3{\rm He}$ and $^4{\rm He}$ abundances was performed under similar conditions as those described in Section 2.2.1.1. Furnace hot blanks were $1.3\pm1.1\times10^4$ and $6.6\pm3\times10^9$ atoms for $^3{\rm He}$ and $^4{\rm He}$, respectively. These $^3{\rm He}$ blanks were <5% for most samples and do not represent a major source of uncertainty. In contrast, $^4{\rm He}$ blanks were up to 100% of the $^4{\rm He}$ extracted from the MK9D samples.

2.2.3. Diffusion experiments

Previous diffusion experiments (Trull et al., 1991; Trull and Kurz, 1993; Shuster et al., 2004) were performed using mafic phenocrysts that were not crushed intensely. It is unknown whether these results are also valid for minerals whose structure has suffered prolonged crushing such as the one experienced by the phenocrysts in the crushing apparatus. Consequently, we carried out new diffusion experiments using fragments of irradiated samples that were crushed under different conditions. Guadalupe olivine was crushed in air at low temperature (25 °C) using several hammer strokes. The resulting fragments were sieved and 0.59 mg of 100-300 µm grains were loaded in a diffusion cell (Farley et al., 1999). San Carlos olivine was crushed in vacuum in the tube, at low temperature (25 °C) and for 25 min, under the conditions described in Section 2.2.1.1. After sieving, the grains were cleaned with ethanol to remove the adhering powder. 0.95 mg of the 40- to 60-µm fraction of this sample was loaded in a diffusion cell. After microscopic inspection we concluded that the shape of these crushed fragments was roughly spherical. Olivines were held at a known temperature T_i during a time t_i under static vacuum. At each heating step, released ³He was purified (see Section 2.2.1.1) and measured by static mass spectrometry (MAP 215-50 and GV SFT). Maximum temperatures reached during diffusion experiments were 480 and 560 °C for San Carlos and Guadalupe olivine, respectively. After the diffusion experiment, samples were heated to ~1600 °C in the furnace (described in Section 2.2.2) to determine the remaining ³He content. Each heating step yielded between 1×10^4 and 2.9×10^7 at of ³He (Table 3). With the exception of the first low temperature step, blank contribution $(1.2 \times 10^4 \text{ atoms})$ was lower than 5% and the total relative uncertainty of the yield was from 3% to 5%. The ³He step-release fractions were converted into diffusion coefficients according to the analytical solutions of (Fechtig and Kalbitzer, 1966). To allow an absolute comparison between our results and the previous estimate, the measured diffusivities (D/a^2) were converted into intrinsic diffusivities (D) by dividing by the olivine grain radius, assuming that the physical grain equals the diffusion domain and that the geometry of the grains is spherical. For this computation a mass-weighted mean radius was computed for each sample, based on the analysis of the grain size distribution. This calculation yielded a weighted radius (a) of $123 \, \mu m$ for the 100- to 300-µm grains of the crushed Guadalupe

Sample	Grains size (µm)	Mass (mg)	³ He (10 ⁶ at. g ⁻¹)	⁴ He (10 ¹¹ at. g ⁻¹)	${}^{3}\mathrm{He_{c}}^{a}~(10^{6}~\mathrm{at.~g^{-1}})$	Fraction of He _c lost (%
Uncrushed samples						
MK9D olivines	500-1000	258.5	8.94 ± 0.20	0.23 ± 0.09	8.66 ± 0.22	
	500-1000	218.4	8.04 ± 0.19	0.27 ± 0.10	7.71 ± 0.22	
	500-1000	224.0	8.40 ± 0.19	0.31 ± 0.10	8.02 ± 0.23	
Data from Blard et al., 2007	320-1000	530.3	8.75 ± 0.27	0.21 ± 0.04	8.50 ± 0.28	
Weighted average $(\pm 1\sigma)$			0.26 ± 0.05	8.23 ± 0.27		
MK9D proxenes	500-1000	142.9	10.52 ± 0.25	1.61 ± 0.16	8.60 ± 0.31	
•	500-1000	197.3	9.69 ± 0.22	1.27 ± 0.11	8.18 ± 0.26	
	500-1000	279.1	9.28 ± 0.19	1.29 ± 0.08	7.75 ± 0.21	
Data from Blard et al., 2007	150-1000	565.4	9.14 ± 0.25	0.32 ± 0.04	8.76 ± 0.26	
Weighted average $(\pm 1\sigma)$				1.12 ± 0.56	8.24 ± 0.29	
Samples crushed at 25 °C						
MK9D olivines crushed in mortar at 25 °C	<40	84.3	8.63 ± 0.42	1.26 ± 0.26	8.32 ± 0.42	No loss
	40-65	183.8	8.53 ± 0.39	0.77 ± 0.12	8.22 ± 0.39	No loss
	65–105	137.0	8.61 ± 0.37	0.73 ± 0.16	8.31 ± 0.38	No loss
	105-190	45.6	8.60 ± 0.42	0.64 ± 0.49	8.29 ± 0.43	No loss
MK9D olivines crushed 30 min in tube at 25 °C	<10	27.4	8.48 ± 0.40	4.35 ± 0.81	8.17 ± 0.40	No loss
	10-40	31.1	8.63 ± 0.40	4.11 ± 0.71	8.33 ± 0.40	No loss
	40-75	56.4	8.49 ± 0.36	5.06 ± 0.39	8.19 ± 0.36	No loss
	75-125	23.4	8.55 ± 0.47	3.00 ± 0.95	8.25 ± 0.47	No loss
	125-190	35.4	8.38 ± 0.39	1.01 ± 0.63	8.07 ± 0.39	No loss
	125-1000	34.0	9.12 ± 0.42	3.36 ± 0.65	8.82 ± 0.43	No loss
TU8 pyroxenes crushed 13.5 min in tube at 25 °C	<45	16.8	18.03 ± 0.65	630.6 ± 1.6	17.4 ± 0.9	No loss
	45-65	13.7	19.54 ± 0.75	718.8 ± 1.9	18.9 ± 1.0	No loss
	65-105	21.5	18.76 ± 0.59	262.9 ± 1.1	18.1 ± 0.9	No loss
	>105	54.7	19.07 ± 0.47	34.0 ± 0.4	18.4 ± 0.8	No loss
Samples crushed up to 325 °C						
MK9D olivines crushed 30 min in tube between 300 and 325 °C		32.7	6.83 ± 0.36	1.36 ± 0.68	6.53 ± 0.37	21
	10-40	58.3	7.86 ± 0.30	4.16 ± 0.38	7.55 ± 0.31	8
	40-75	127.9	8.50 ± 0.30	0.77 ± 0.17	8.20 ± 0.31	No loss
	75-125	51.5	8.60 ± 0.33	0.19 ± 0.43	8.29 ± 0.33	No loss
	125-190	79.7	8.16 ± 0.30	0.76 ± 0.28	7.86 ± 0.31	No loss
	125-1000	107.4	8.46 ± 0.29	0.15 ± 0.21	8.16 ± 0.30	No loss
MK9D pyroxenes crushed 45 min in tube between	<10	17.7	5.07 ± 0.49	1.53 ± 1.26	4.69 ± 0.82	43
00 and 325 °C	10-40	38.2	7.88 ± 0.41	9.77 ± 0.58	7.50 ± 0.78	9
	40-75	34.8	7.92 ± 0.43	5.13 ± 0.64	7.54 ± 0.79	8
	104-190	19.5	8.72 ± 0.67	11.24 ± 1.14	8.35 ± 0.94	No loss

a In the case of uncrushed MK9D phenocrysts, correction for magmatic 3 He is calculated assuming that (i) the individual 4 He concentrations measured in the same aliquot equal the magmatic 4 He and (ii) the magmatic 3 He fratio is 8 Ra. In the case of crushed MK9D phenocrysts, contamination by 4 He-rich particles from the crushing tube is probable. So, correction for magmatic 3 He is computed using the weighted average of the 4 He measured within uncrushed phenocrysts and a magmatic ratio of 8 Ra. In the case of TU8 pyroxenes, magmatic correction is $6.7 \pm 6.7 \times 10^5$ at. g^{-1} of 3 He (see Section 2.1.1).

Table 3
Diffusion data from Guadalupe and San Carlos crushed and irradiated olivines

Crushed, irradiated Guadahupe olivine 100–300 µm (0.59 mg). Equivalent weighted radius is 123 micron 300 3600 1 17.5 310 3600 1 17.2 320 3600 1 16.9 330 1800 1 16.9 330 1800 1 16.6 340 1800 1 16.6 340 1800 2 15.6 340 1800 2 15.6 380 1800 2 15.6 380 1800 2 15.6 380 1800 3 15.1 400 1800 3 15.1 400 1800 3 15.1 400 1800 3 14.9 420 1800 5 14.4 440 1800 8 140 13.6 480 1800 12 13.3 500 1800 10 13.6 480 1800 12 13.3 500 1800 10 13.6 50 14.4 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 13.6 50 1800 10 12 13.3 500 1800 10 12 13.3 500 1800 10 12 13.3 500 1800 10 12 13.3 500 1800 10 12 13.3 500 1800 10 12 13.3 500 1800 10 12 13.3 500 1800 10 12 13.3 500 1800 10 12 13.3 500 1800 10 12 13.3 500 1800 24 12.3 560 1800 24 12.3 560 1800 24 12.3 560 1800 24 12.3 560 1800 10 11 23.6 50 7200 1.1 23.6 50 7200 1.1 23.6 50 15.0 7200 1.1 23.6 50 15.0 7200 1.1 23.6 50 15.0 7200 1.1 23.6 50 15.0 7200 1.1 23.6 50 15.0 7200 1.1 23.6 50 15.0 7200 1.1 23.6 50 15.0 7200 1.1 5.8 50 15.3 300 3600 38 17.5 50 15.3 300 3600 38 17.5 50 15.3 300 3600 38 16.9 50 1800 47 15.8 50 15.3 300 3600 38 16.9 50 15.8 300 30 300 300 300 300 300 300 300 300	$\ln D (\ln (\mathrm{cm}^2 \mathrm{s}^{-1}))$	$10^4/T (\mathrm{K}^{-1})$	³ He (10 ⁵ atoms)	Time (s)	Temperature (°C)
310		radius is 123 microns	n (0.59 mg). Equivalent weighted	dalupe olivine 100–300 µr	Crushed, irradiated Gua
320 3600 1 1 16.9 330 1800 1 16.6 340 1800 1 16.6 340 1800 1 16.3 350 1800 1 16.1 360 1800 2 15.8 370 1800 2 15.8 370 1800 2 15.6 380 1800 2 15.3 390 1800 3 15.1 400 1800 3 15.1 400 1800 3 14.9 420 1800 5 14.4 440 1800 8 14.0 460 1800 10 13.6 480 1800 12 13.3 500 1800 12 13.3 500 1800 20 12.6 540 1800 20 12.6 540 1800 20 12.6 540 1800 20 12.6 540 1800 20 12.6 540 1800 21 13.3 560 1800 16 12.9 520 1800 20 12.6 540 1800 20 12.6 540 1800 21 13.3 560 1800 16 12.9 520 1800 20 12.6 540 1800 20 12.6 540 1800 24 12.3 550 7200 12.6 520 1800 28 12.0 Total ³He after fusion (105 atoms) 769 Crushed, irradiated San Carlos olivine 40-60 μm (0.95 mg). Equivalent weighted radius is 26 microns 50 7200 0.1 75 7200 0.2 28.7 150 7200 1.1 23.6 200 7200 3.4 21.1 250 5400 14 19.1 300 3600 38 17.5 320 3600 38 17.5 320 3600 38 17.5 320 3600 38 16.9 340 1800 47 15.8 339 1800 65 15.3 360 1800 17 15.8 339 1800 5 16.3 339 1800 5 16.3 330 1800 17 15.8 339 1800 17 15.8 339 1800 5 16.3 330 1800 12 15.8 339 1800 12 15.8 340 1800 144 14.4 444 444 444 444 444 444 444 444 444	-30.4	17.5	3	3600	300
1800	-30.6	17.2	1	3600	310
1800	-30.4	16.9	1	3600	320
1800	-29.9		1		
1	-29.5		1		
15.8 15.8 15.6 15.6 15.6 15.6 15.6 15.6 15.6 15.6 15.6 15.6 15.6 15.0 15.0 15.0 15.0 15.0 15.0 15.0 15.1 15.0 15.1 15.0 15.1 15.0 15.1 15.0 15.1 15.0 15.1 15.0 15.1 15.0 15.1 15.0 15.0 15.1 15.0	-29.0		1		
1800 2 15.6	-28.7				
1800 1800 2 15.3	-28.4				
390 1800 3 15.1 400 1800 3 14.9 420 1800 5 14.4 440 1800 8 14.0 4460 1800 10 13.6 480 1800 12 13.3 500 1800 16 12.9 550 1800 20 12.6 540 1800 24 12.3 560 1800 28 12.0 Total ³He after fusion (10⁵ atoms) 769 **Crushed, irradiated San Carlos olivine 40–60 μm (0.95 mg). Equivalent weighted radius is 26 microns 50 7200 0.1 31.0 75 7200 0.2 28.7 150 7200 1.1 23.6 250 7200 1.1 23.6 250 5400 14 19.1 250 5400 14 19.1 250 5400 14 19.1 250 5400 14 19.1 2600 7200 3.8 17.5 27.0 28.7 29.0 29.0 20.0 20.0 20.0 20.0 20.0 20.0	-28.0				
1800 1800 3 14.9 1200 1800 5 14.4 1440 1800 8 14.0 1460 1800 10 13.6 1800 12 13.3 5000 1800 16 12.9 520 1800 20 12.6 540 1800 24 12.3 5600 1800 24 12.3 5600 1800 28 12.0 Total ³He after fusion (10⁵ atoms) 769 Crushed, irradiated San Carlos olivine 40−60 μm (0.95 mg). Equivalent weighted radius is 26 microns 50	-27.7				
1800 5	-27.4				
140	-26.6				
460	-25.9				
480 1800 12 13.3 500 1800 16 12.9 520 1800 20 12.6 540 1800 24 12.3 560 1800 28 12.0 Total ³ He after fusion (10 ⁵ atoms) 769 Crushed, irradiated San Carlos olivine 40–60 μm (0.95 mg). Equivalent weighted radius is 26 microns 50 7200 0.1 31.0 75 7200 0.2 28.7 150 7200 0.2 28.7 150 7200 1.1 23.6 200 7200 1.1 23.6 200 7200 3.4 21.1 250 5400 14 19.1 300 3600 38 17.5 320 3600 38 17.5 320 3600 38 16.9 340 1800 28 16.3 360 1800 47 15.8 380 1800 65 15.3 360 1800 65 15.3 360 1800 65 15.3 360 1800 65 15.3 379 1800 5 16.3 320 1800 65 15.3 339 1800 65 15.8 340 1800 65 15.8 340 1800 65 15.8 350 1800 65 15.8 360 1800 65 15.8 360 1800 65 15.8 360 1800 65 15.8 360 1800 65 15.8 360 1800 17 15.8 360 1800 17 15.8 360 1800 65 15.3 360 1800 65 15.3 360 1800 65 15.3 360 1800 65 15.3	-25.4				
1800 16 12.9 520 1800 20 12.6 540 1800 24 12.3 560 1800 28 12.0 Total ³ He after fusion (10 ⁵ atoms) 769 Crushed, irradiated San Carlos olivine 40–60 μm (0.95 mg). Equivalent weighted radius is 26 microns 50 7200 0.1 31.0 75 7200 0.2 28.7 150 7200 0.2 28.7 150 7200 1.1 23.6 200 7200 3.4 21.1 250 5400 14 19.1 300 3600 38 17.5 320 3600 38 16.9 340 1800 28 16.3 360 1800 47 15.8 380 1800 65 15.3 360 1800 17 15.8 380 1800 65 15.3 360 1800 17 15.8 339 1800 5 16.3 320 1800 5 16.3 320 1800 17 15.8 339 1800 61 14.9 440 1800 144 14.4 440 1800 149 14.0 440 1800 149 14.0	-23.4 -24.9				
1800 20 12.6 1800 24 12.3 12.0 1800 28 12.0 12.0 1800 28 12.0 1800 28 12.0 12.0 1800 38 12.0 12.0 1800 38 12.0 12.0 1800 38 12.0 1800 38 12.0 12.0 1800 38 12.0 12.0 1800 38 12.0 12.0 12.0 12.0 12.0 12.0 12.0 12.0	-24.3 -24.3				
1800	-24.3 -23.9				
1800 1800 28 12.0 Total ³ He after fusion (10 ⁵ atoms) 769 Crushed, irradiated San Carlos olivine 40–60 μm (0.95 mg). Equivalent weighted radius is 26 microns 10.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.	-23.4 -23.4				
Total ³ He after fusion (10 ⁵ atoms) Crushed, irradiated San Carlos olivine 40–60 μm (0.95 mg). Equivalent weighted radius is 26 microns 50 7200 0.1 31.0 75 7200 0.2 28.7 150 7200 1.1 23.6 200 7200 3.4 21.1 250 5400 14 19.1 300 3600 38 17.5 320 3600 38 16.9 340 1800 28 16.3 360 1800 47 15.8 380 1800 65 15.3 380 1800 65 15.3 390 1800 5 16.3 320 1800 5 16.3 320 1800 5 16.3 320 1800 61 15.8 320 1800 61 14.9 420 1800 104 14.4 440 1800 149 14.0 440 440 1800 149 14.0					
Crushed, irradiated San Carlos olivine 40–60 µm (0.95 mg). Equivalent weighted radius is 26 microns 50 7200 0.1 31.0 75 7200 0.2 28.7 150 7200 1.1 23.6 200 7200 3.4 21.1 250 5400 14 19.1 300 3600 38 17.5 320 3600 38 16.9 340 1800 28 16.3 360 1800 47 15.8 380 1800 65 15.3 360 1800 17 15.8 339 1800 5 16.3 320 1800 5 16.3 320 1800 5 16.3 320 1800 61 140 440 1800 61 14.9 440 1800 104 14.4 440 1800 104 14.4 440 1800 149 14.0 440 440 1800 149 14.0	-23.0	12.0	28	1800	300
75 7200 0.2 28.7 150 7200 1.1 23.6 200 7200 3.4 21.1 250 5400 14 19.1 300 3600 38 17.5 320 3600 38 16.9 340 1800 28 16.3 360 1800 47 15.8 380 1800 65 15.3 360 1800 17 15.8 339 1800 5 16.3 320 1800 5 16.3 320 1800 12 15.8 400 1800 12 15.8 400 1800 104 14.9 420 1800 104 14.4 440 1800 149 14.0 460 1800 248 13.6	-45.2			·	
150 7200 1.1 23.6 200 7200 3.4 21.1 250 5400 14 19.1 300 3600 38 17.5 320 3600 38 16.9 340 1800 28 16.3 360 1800 47 15.8 380 1800 65 15.3 360 1800 17 15.8 339 1800 5 16.3 320 1800 2 16.9 360 1800 12 15.8 400 1800 61 14.9 420 1800 104 14.4 440 1800 149 14.0 460 1800 248 13.6	-43.2 -42.4				
200 7200 3.4 21.1 250 5400 14 19.1 300 3600 38 17.5 320 3600 38 16.9 340 1800 28 16.3 360 1800 47 15.8 380 1800 65 15.3 360 1800 17 15.8 339 1800 5 16.3 320 1800 2 16.9 360 1800 12 15.8 400 1800 61 14.9 420 1800 104 14.4 440 1800 149 14.0 460 1800 248 13.6					
250 5400 14 19.1 300 3600 38 17.5 320 3600 38 16.9 340 1800 28 16.3 360 1800 47 15.8 380 1800 65 15.3 360 1800 17 15.8 339 1800 5 16.3 320 1800 2 16.9 360 1800 12 15.8 400 1800 61 14.9 420 1800 104 14.4 440 1800 149 14.0 440 1800 149 14.0 460 1800 248 13.6	-39.1				
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320 3600 38 16.9 340 1800 28 16.3 360 1800 47 15.8 380 1800 65 15.3 360 1800 17 15.8 339 1800 5 16.3 320 1800 2 16.9 360 1800 12 15.8 400 1800 61 14.9 420 1800 104 14.4 440 1800 149 14.0 460 1800 248 13.6	-33.6 21.0				
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400 1800 61 14.9 420 1800 104 14.4 440 1800 149 14.0 460 1800 248 13.6	-31.4				
420 1800 104 14.4 440 1800 149 14.0 460 1800 248 13.6	-29.4				
140 1800 149 14.0 1460 1800 248 13.6	-27.7				
460 1800 248 13.6	-26.9				
	-26.2				
480 1800 288 13.3	-25.3				
1000 200 13.3	-24.8	13.3	288	1800	480
Γotal ³ He after fusion (10 ⁵ atoms) 3708				_	

olivine and a radius of 28 μ m for the 40- to 60- μ m grains of the crushed San Carlos olivine. The radius-normalized diffusivities (*D*) were used together with the temperatures of each heating step to generate Arrhenius plots (Fig. 5). The activation energy $E_{\rm a}$ (J mol⁻¹) and the frequency factor D_0 (cm² s⁻¹) were then estimated by linear regression, assuming an Arrhenius relationship:

$$D(T) = D_0 \cdot \exp\left(-E_a/RT\right) \tag{2}$$

where $R = 8.314 \,\mathrm{J \, K^{-1} \, mol^{-1}}$ is the gas constant. Temperature cycling was incorporated in the diffusion experiments carried out on the San Carlos sample, but not on the Guadalupe sample.

3. RESULTS

3.1. Pattern of ³He_c release by vacuum crushing

Helium data from the vacuum crushing experiments are summarized in Table 1 and Fig. 1.

In the case of the Hawaiian phenocrysts, all of the first crushing steps (5 min) are characterized by 3 He/ 4 He ratios that are similar, within uncertainties, to the MORB-like reservoir value (8 Ra, Ra = 1.384×10^{-6}) that was previously observed for these Mauna Kea lavas (Kurz et al., 1983; Eiler et al., 1996). This suggests that the magmatic component is dominant during this crushing step. We calculated the

Helium loss during vacuum crushing

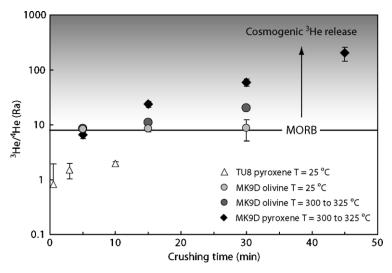


Fig. 1. Evolution of the measured ${}^{3}\text{He}/{}^{4}\text{He}$ ratio vs. time during step-crushing. Horizontal line shows the MORB-like value (Ra = 1.384×10^{-6} is the air ${}^{3}\text{He}/{}^{4}\text{He}$ ratio). Experiments were performed at various temperatures as indicated.

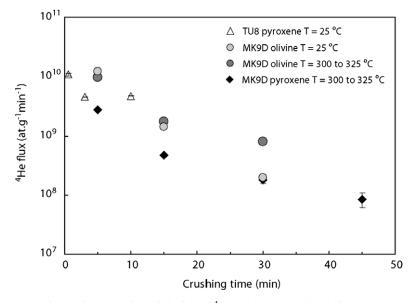


Fig. 2. Time evolution of the flux of ⁴ He released at each crushing step.

flux of helium extracted by crushing (Fig. 2). This was done following (Scarsi, 2000) by dividing the amount of released ⁴He by the crushing time corresponding to each step. The first crushing steps are characterized by high ⁴He fluxes (between 2.78 ± 0.07 and $12.29 \pm 0.07 \times 10^{9}$ at g^{-1} min⁻¹, Fig. 2), consistent with the order of magnitude observed by (Scarsi, 2000) under similar conditions and comparable magmatic concentrations. This author interpreted this primary and high-flux release as being characteristic of the activation of the magmatic helium reservoir, which actually corresponds to the helium component trapped within fluid or melt inclusions. Regardless of temperature, the flux of ⁴He released during the subsequent crushing steps decrease with time. This decrease can be described by a simple exponential law (Fig. 2), which suggests, according to the model of (Scarsi, 2000), that ⁴He has been extracted from a single

reservoir (fluid inclusions), even after several minutes of crushing. This interpretation is consistent with the fact that concentration of matrix-sited radiogenic ⁴He is quite low $(\le 3 \times 10^9 \text{ at. g}^{-1})$ in MK9D phenocrysts. However, for crushing times greater than 5 min with T > 300 °C, the ³He/⁴He ratio measured from MK9D olivines and pyroxenes varies between 10.9 ± 0.5 and 202 ± 58 Ra. This is far above the 8-Ra MORB value (Table 1 and Fig. 1). Such ³He/⁴He ratios are higher than those previously reported for Mauna Kea basalts (Kurz et al., 1983; Eiler et al., 1996) and some of these values (>50 Ra) are even higher than the highest ratios observed for plume-like volcanism (Hilton et al., 1999; Stuart et al., 2003). Consequently, as previously observed by Yokochi et al. (2005) and Moreira and Madureira (2005) after several minutes of crushing, it seems inescapable that cosmogenic ³He has been extracted

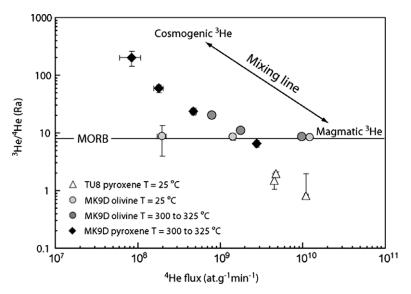


Fig. 3. Plot of the measured ${}^{3}\text{He}/{}^{4}\text{He}$ ratio against the corresponding flux of ${}^{4}\text{He}$ released by crushing. The anti-correlation obtained for MK9D phenocrysts crushed at high temperature can be interpreted as a mixing line between a pure magmatic helium reservoir and the addition of cosmogenic ${}^{3}\text{He}$. The measured ${}^{3}\text{He}/{}^{4}\text{He}$ is all the more sensitive to this addition of ${}^{3}\text{He}_{c}$ as the flux of extracted ${}^{4}\text{He}$ is smaller.

from the phenocryst lattice. This conclusion is also supported by the anti-correlation between the measured ³He/⁴He ratios and the flux of extracted ⁴He from samples crushed at T > 300 °C (Fig. 3). This empirical relationship reflects mixing between a magmatic helium component, having low ³He/⁴He, and cosmogenic ³He, whose relative contribution to the crushed ³He/⁴He increases as the flux of magmatic ⁴He is reduced (Fig. 3). Because the release of ³He_c is smallest during the short first crushing step (Yokochi et al., 2005), the best evaluation of the isotopic composition of the primordial (magmatic) trapped component is thus the 8.4 ± 0.3 Ra value obtained from the first crushing step of MK9D olivines (Table 1). Crushing steps of TU8 pyroxenes display ³He/⁴He ratios lower than the MORB value. (Scarsi, 2000) demonstrated that radiogenic ⁴He may be easily extracted by vacuum crushing. Thus, because TU8 samples are rich in radiogenic ⁴He, these low ³He/⁴He ratios probably result from a two (or three) component mixture: magmatic helium, radiogenic ⁴He (and cosmogenic ³He).

A helium budget calculation can be made to calculate the amount of ${}^{3}\text{He}_{c}$ extracted at each step, using this assumed magmatic ${}^{3}\text{He}/{}^{4}\text{He}$ ratio (${}^{3}\text{He}/{}^{4}\text{He}_{mag}$) in this sample. Because radiogenic ${}^{4}\text{He}$ can be neglected compared to the magmatic ${}^{4}\text{He}$ component, the concentrations of cosmogenic ${}^{3}\text{He}$ extracted (${}^{3}\text{He}_{cex}$ in at. g^{-1}) at each step is:

$$^{3}\text{He}_{\text{cex}} = {}^{4}\text{He}_{\text{crush}} \times ({}^{3}\text{He}/{}^{4}\text{He}_{\text{crush}} - {}^{3}\text{He}/{}^{4}\text{He}_{\text{mag}})$$
 (3)

where $^4\text{He}_{\text{crush}}$ (at. g^{-1}) is the amount of ^4He and $^3\text{He}/^4\text{He}_{\text{crush}}$ is the measured $^3\text{He}/^4\text{He}$ extracted at each step. ($^3\text{He}/^4\text{He}_{\text{mag}}$) is assumed to be 8.4 ± 0.3 Ra for MK9D olivines, 6.5 ± 0.9 Ra MK9D pyroxenes, and 0.8 ± 1.2 Ra for TU8 pyroxenes.

In the $\geqslant 300$ °C experiment, this calculation shows that the proportion of $^3\text{He}_c$ lost in each step ranges from 1.4% to 4.3% for MK9D pyroxenes and between 0.1% and

2.4% for MK9D olivines (Table 1). The sums of these losses represent total releases of 3.3% and 8.1%, for MK9D olivines and pyroxenes, respectively. Such values are comparable with the order of magnitude of ³He_c loss observed by (Yokochi et al., 2005; Blard et al., 2006b), but are significantly above the <0.5% loss reported by (Moreira and Madureira, 2005).

In contrast, the step-crushing performed on MK9D olivines under cold conditions (T < 25 °C) does not show any increase of the ${}^{3}\text{He}/{}^{4}\text{He}$ ratio, remaining similar to the 8 Ra MORB value, within uncertainties, even after 30 min of crushing (Table 1, Figs. 1 and 3). Consequently, MK9D olivines, as well as TU8 pyroxenes, released low amounts of ${}^{3}\text{He}_{cex}$ (<1%) in cold conditions, although the duration of crushing was similar to the experiment performed at high temperature (Table 1).

Moreover, given that the irradiated San Carlos olivine contains a very high concentration of matrix-sited 3 He, crushing data obtained from this sample permits a very low detection limit of the 3 He loss. These data from San Carlos (Table 1) permit to place a maximum limit of 1.8% (for 30 min of crushing, i.e. \sim 1800 strokes) on the magnitude of the loss, under cold conditions (T < 25 °C). This value is not significantly different from the 3 He_c loss observed in the non-irradiated samples crushed under the same conditions.

3.2. Cosmogenic ³He concentrations vs. grain size

The 3 He concentrations measured in the six aliquots of uncrushed MK9D olivines and pyroxenes subjected to total thermal outgassing ranged between $8.0 \pm 0.2 \times 10^6$ and $10.5 \pm 0.3 \times 10^6$ at. g⁻¹. These minerals have contrasting 4 He concentrations that range between $0.2 \pm 0.1 \times 10^{11}$ and $0.3 \pm 0.1 \times 10^{11}$ at. g⁻¹ for olivines and between $1.3 \pm 0.2 \times 10^{11}$ and $1.6 \pm 0.2 \times 10^{11}$ at. g⁻¹ for

pyroxenes. For each replicate, magmatic ³He was calculated multiplying the ⁴He concentration by the ³He/⁴He ratios measured from the first crushing step, i.e. $8.4 \pm$ $0.3 \, \mathrm{Ra}$ for olivines and $6.5 \pm 0.9 \, \mathrm{Ra}$ for pyroxenes. Cosmogenic ³He concentrations of each uncrushed replicate were then calculated by difference. Resulting ³He_c concentrations agree within 1σ and are consistent with the previous estimate of (Blard et al., 2007). Assuming that the absolute calibrations of the Caltech and the CRPG mass spectrometers match within analytical uncertainties, it is possible to use all these ³He_c data to calculate uncertainty weighted averages of $8.23 \pm 0.27 \times 10^6$ and $8.24 \pm 0.29 \times$ 10⁶ at. g⁻¹ for olivines and pyroxenes, respectively. Because uncrushed phenocrysts have not been affected by ³He_c loss, these values will be considered in the following as reference concentrations to which the concentrations measured in the crushed samples will be

Surprisingly, the ⁴He concentrations measured in the furnace in the crushed phenocrysts are significantly above the average concentrations obtained from the uncrushed phenocrysts. This discrepancy is roughly anti-correlated with grain size (Table 2), and this observation is in contradiction with the fact that small crushed phenocrysts (typically <100 μm) are depleted in inclusion-hosted magmatic helium (Kurz et al., 1990). Consequently, it seems reasonable to propose that the crushed phenocrysts were contaminated by small amounts of steel chips from the crushing tube that were not completely removed by the hand magnet separation. Indeed, given the high concentration of ⁴He contained within these small pieces of steel (>1014 ⁴He at. g⁻¹), even few micrograms of material are sufficient to bias significantly the measured ⁴He concentrations. Because the contaminating steel chips are fine (<50 μm), this undesirable bias is all the more significant for the smallest fractions that were enriched in this contaminant through the sieving process. In addition, the furnace blanks were also a major source of uncertainty in estimating the concentration of ⁴He of the smallest MK9D aliquots (<100 mg), because these blank corrections represent up to 65% of the gas extracted from these samples. These considerations imply that the individual ⁴He contents measured in the crushed phenocrysts will yield inaccurate ³He_c concentrations if they were used to correct for the magmatic ³He (see Eq. (1)).

Consequently, we have chosen to estimate the magmatic 3 He correction for the crushed MK9D phenocrysts in a different way: we used the average magmatic concentrations obtained by fusing the uncrushed phenocrysts $(2.9 \pm 0.6 \times 10^5 \text{ and } 1.2 \pm 0.6 \times 10^6 \text{ at. g}^{-1} \text{ for olivines and pyroxenes, respectively, Table 1). For further comparison, it must be kept in mind that this estimate of the magmatic <math>^3$ He yields a maximum correction. In the case of the TU8 pyroxenes, correction for magmatic helium was performed assuming all the replicates have 3 He concentrations of $6.7 \pm 6.7 \times 10^5 \text{ at. g}^{-1}$ (see Section 2.1.1 for explanation). We stress that the standard magmatic 3 He correction should ideally be applied, especially when the crushing device is clean enough to prevent any helium contamination. This however high-

lights how important it is to pay careful attention to the crushing conditions.

The corrected ³He_c concentrations show that the only samples that have been affected by loss that is statistically significant (outside 2σ uncertainty) are the smallest fractions (<10 µm) of MK9D olivines (21%) and pyroxenes (43%) that were crushed under hot conditions ($T > 300 \,^{\circ}$ C) (Table 2 and Fig. 4). All samples crushed under cold conditions (T < 25 °C), in the tube or the agate mortar, have ³He_c concentrations that are identical with the reference value, within 1σ uncertainties, regardless of the duration and style of crushing (tube or mortar) (Table 2 and Fig. 4). These results are consistent with the observations from online crushing. These data support the conclusion that significant ³He_c loss only affects the smallest fraction and only if the sample is heated to sufficiently high temperatures $(T > 300 \, ^{\circ}\text{C}).$

Furthermore, it is possible to perform a $^3\text{He}_c$ budget calculation weighting the $^3\text{He}_c$ concentration measured in each fraction by the mass of the corresponding sieve fraction. This estimate must be considered with caution because it assumes that the sieving has not changed the size distribution of the sample, while very fine fractions could have been lost during this process. The total $^3\text{He}_c$ losses are $3\pm1\%$ for M9D olivines and $13\pm8\%$ for MK9D pyroxenes, which is consistent with the total loss observed from the online crushing (Table 1). Therefore, the good match between this budget and the data supports the view that the total $^3\text{He}_c$ released by crushing the bulk fraction at 300 °C is mainly controlled by loss affecting the fraction <10 μ m, even when the loss from the bulk fraction is significant (>10%).

3.3. Diffusivity of matrix-sited ³He in crushed olivines

The results of the two diffusion experiments are summarized in Table 3 and shown in Fig. 5 as Arrehnius plots ($\ln D$ vs. $10^4/T$). The regression parameters of the best fit and the resulting diffusion parameters are summarized in Table 4.

Small 3 He excesses were observed relative to linearity during the initial and low-temperature ($T < 100 \,^{\circ}$ C) steps. Because this release is within the blank level and can bias the estimated diffusivity, the points with this initial excess were removed before reducing these diffusion data. After removal of this excess, the data yield a linear correlation within the $\ln D$ vs 1/T space for steps between 250 and 560 $^{\circ}$ C. The retrograde cycling experiment performed on the San Carlos olivine exhibits a slight scatter around the regressed line (Fig. 5), but this remains lower than the inter-sample variability and should not represent a major limitation for the first order assessment that we want to perform here.

These experiments yielded activation energies (E_a) of 127 ± 1 and 127 ± 6 kJ.mol⁻¹ and the frequency factors (given as $\ln D_0$) of -4.65 ± 0.02 and -4.8 ± 0.7 cm² s⁻¹ for the Guadalupe and San Carlos olivines, respectively (Table 4). Similarities and discrepancies with previous determination are discussed below in Section 4.2.

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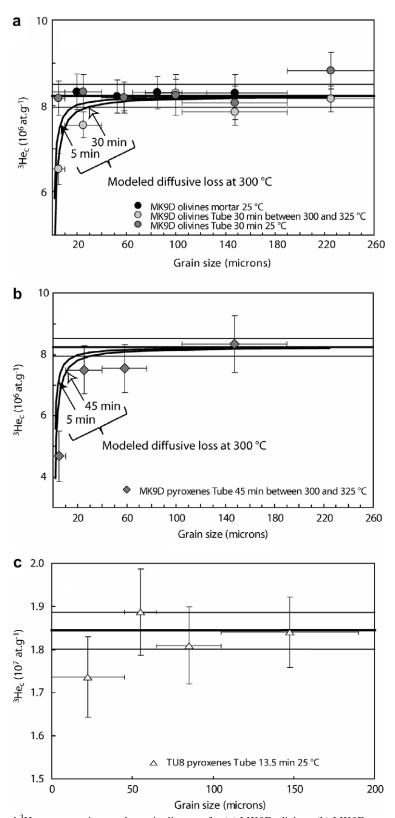


Fig. 4. Plot of the measured 3 He concentration vs. the grain diameter for (a) MK9D olivines, (b) MK9D pyroxenes and (c) TU8 pyroxenes. Horizontal lines are the average and the 1σ envelop of the 3 He_c concentrations determined from uncrushed phenocrysts. Black curves show the modeled diffusive loss for two extreme crushing times at 300 $^{\circ}$ C (5 and 30 min of heating for MK9D olivines; 5 and 45 min of heating for MK9D pyroxenes). This was done because the time of apparition of the small fraction during crushing is uncertain. This modeling was done assuming grains have spherical geometry, using the diffusion coefficient established in the present study (Table 4) and the diffusion equation of (Fechtig and Kalbitzer, 1966).

Helium loss during vacuum crushing

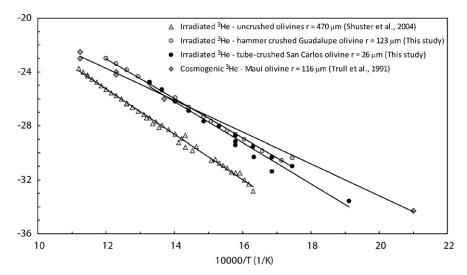


Fig. 5. Arrhenius plot from the diffusion experiment performed on the crushed Guadalupe and San Carlos olivines (this study). Also shown are the data of (Shuster et al., 2004) from irradiated uncrushed Guadalupe olivines and the data of (Trull et al., 1991) from Maui olivines exposed to cosmic rays. The diffusion coefficients obtained from these regressed data are given in Table 4.

Table 4
Diffusion coefficients obtained for matrix-sited ³He in olivines

Reference	He variety	Sample	Crushed/uncrushed	Radius ^a (µm)	R^2	$ln(D_0)$ (ln (cm ² s ⁻¹)	$E_{\rm a}$ (kJ mol ⁻¹)
Shuster et al. (2004)	Irradiated ³ He	Guadalupe olivine– Fo _{75–80}	Uncrushed	470	0.99	-5.1 ± 0.3	140 ± 2
Trull et al. (1991)	Cosmogenic ³ He	Maui olivine-Fo ₈₀	Uncrushed	116	0.99	-8.5 ± 1.8	105 ± 17
This study	Irradiated ³ He	Guadalupe olivine– Fo _{75–80}	Crushed in air three hammers strokes	123	0.99	-4.65 ± 0.02	127 ± 1
This study	Irradiated ³ He	San Carlos olivine– Fo ₉₂	Crushed in vacuo 25 min in tube	26	0.97	-4.8 ± 0.7	127 ± 6

^a Radii are given as equivalent sphere radii.

4. DISCUSSION

4.1. Is the ³He_c crushing loss mainly the result of volume diffusion?

These new experiments show that the ³He_c loss during vacuum crushing is not uniquely controlled by the grain size of the crushed phenocrysts, because even the smallest fraction (<10 µm) did not show significant release under cold conditions (T < 25 °C). This result is in contradiction with the previously proposed mechanism that involved spallation tracks, which are 10- to 20-µm-long matrix damage tracks induced by cosmogenic reactions (Finkel et al., 1978; Yokochi et al., 2005). This hypothesis assumes that any atom of ³He_c located in such a lattice defect would be released when the track encounters a new surface created by crushing. The relative loss induced by this theoretical mechanism is purely a function of the surface/volume ratio, and, is thus only dependent on the grain size. According to Yokochi et al. (2005), this model is not supposed to be temperature-dependent, and predicts that the loss would be at least 30% for fractions smaller than 30 μm , whatever the temperature and the crushing energy. Consequently, on the basis of our new data set, it seems reasonable to conclude that the model involving spallation tracks is invalid.

It appears that the $^3\text{He}_c$ loss is controlled both by (i) the grain size and by (ii) the temperature and duration of the heating. Such behavior is characteristic of volume diffusion. To quantitatively assess this possibility, we modeled the magnitude of pure diffusion-driven loss as a function of grain size and temperature (Fig. 4). This theoretical loss was calculated assuming a spherical geometry of the grains

and using the analytical solution from (Fechtig and Kalbitzer, 1966) with the temperature monitored during the crushing (Table 1) and the new empirical diffusion parameters (D_0 and E_a , Table 4) obtained in this study. Because the time at which the finest grains appears during crushing is uncertain, we considered two scenarios with extreme durations (from 5 to 30 min for olivines and from 5 to 45 min for pyroxenes, see Fig. 4 caption) to provide conservative upper and lower limits on the modeled magnitude of loss. The modeled loss matches the data fairly well, within 1σ analytical uncertainties, and this consistency is thus a good argument in favor of a pure temperature-induced diffusive control of the ³He_c release during crushing. Diffusivity of magmatic helium can be significantly different in olivines and pyroxenes (Trull and Kurz, 1993). However, the diffusion kinetics of radiogenic ⁴He in pyroxene (Lippolt and Weigel, 1988) is similar (within the same order of magnitude) to those obtained in this study from irradiated olivines (Fig. 6). Consequently, because we did not get diffusion data from irradiated clinopyroxene, we decided to use identical diffusion parameters (from irradiated olivines) to model the loss for both mineral species. The good agreement between the modeled and observed loss in MK9D pyroxenes (Fig. 4) suggests that diffusivity must be within the same order of magnitude for both minerals. Hence, this approximation is not a major source of uncertainty.

The hypothesis of purely diffusive control of ³He_c loss was ruled out by previous studies (Moreira and Madureira, 2005; Yokochi et al., 2005; Blard et al., 2006b), on the basis

that, even if the highest empirical diffusion parameters were considered (Trull et al., 1991; Shuster et al., 2004), the calculated loss would be significantly lower than the observed3Hec release. If the new diffusion parameters obtained here are considered and if we assume a similar grain size distribution as the one obtained here, the diffusion-induced loss would be about 0.5% for a bulk sample crushed at 70 °C. This is consistent with the release observed by (Moreira and Madureira, 2005) at this temperature. However, these authors argue that the ³He/²¹Ne ratio measured during crushing is similar to the ratio of the olivine matrix, and this observation does not support a diffusion control of the crushing release given that helium diffusivity is several order of magnitude higher than the neon diffusivity (Shuster and Farley, 2005). Consequently, this observation implies that other mechanisms than diffusion cannot be totally ruled out. The fact that alternative mechanisms may also play a role is supported by the fact that 1.8% of matrix-sited ³He was released by crushing under cold conditions the irradiated San Carlos olivine during 25 min. Previous studies (Yokochi et al., 2005) and (Moreira and Madureira, 2005) proposed that extraction at the grain surface due to lattice distortion might be caused by plastic deformation during crushing. If this alternative mechanism could account for $\sim 1\%$ of the extraction, the experimental results obtained in this study suggest that temperature-enhanced diffusion is the mechanism responsible for the increase of the level of ³He_c extraction above 10% (Table 2 and Fig. 3). Although each of these mechanisms is characterized by a different magnitude of ³He_c release, the coexis-

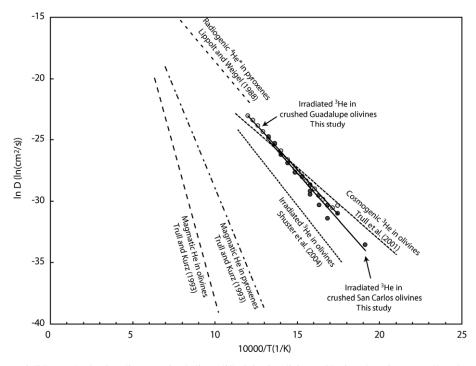


Fig. 6. Comparison of different Arrhenius diagrams for helium diffusivity in olivine. Diffusion data from (Trull and Kurz, 1993) are from inclusion-sited magmatic helium in olivine and pyroxenes. (Lippolt and Weigel, 1988) measured diffusivity of radiogenic ⁴He in augite. Shuster et al. (2004) determined diffusion parameters for matrix-sited ³He in irradiated olivines. Trull et al. (1991) measured the diffusivity of cosmogenic ³He from olivines exposed to cosmic rays during ~500 ka at ~3000 m elevation. Shown data from the present study were obtained from proton-irradiated olivines that were subsequently crushed (Fig. 5 and Table 4).

tence of both is compatible with our observations and those of (Moreira and Madureira, 2005) and (Scarsi, 2000).

Nevertheless, it must be acknowledged that, even if our diffusion parameters are used, volume diffusion remains insufficient to reproduce the magnitude of losses of about 10-25% (from bulk fractions) that were observed by crushing olivines at the CRPG (Nancy) noble gas laboratory (Yokochi et al., 2005; Blard et al., 2006b). Indeed, considering the tube temperature ($T < 100 \,^{\circ}$ C) reported in these studies and the grain size distribution of the crushed samples (Yokochi et al., 2005; Blard et al., 2006b), diffusion would induce a loss <1%, far below the magnitude actually observed. Consequently it seems difficult to state that volume diffusion was the sole factor controlling the loss observed at the CRPG laboratory. Other mechanisms responsible for the loss could also be specific to the CRPG crushing apparatus, such as short-term increase of diffusivity caused by plastic deformation (Yokochi et al., 2005). Another possibility would be to question the accuracy of the temperature determinations. In the studies of (Yokochi et al., 2005; Blard et al., 2006b) temperature was measured on the wall of the crushing tube, several centimeters above the bottom of the tube, which could have resulted in an underestimate of the actual sample temperature (at the bottom of the tube), as a result of a temperature gradient. In contrast, (Moreira and Madureira, 2005) measured the temperature (T = 70 °C) at the contact between the sample and the iron slug using a paint especially designed to record maximum temperature. The low level of ³He_c loss (0.5%) observed by these authors is thus consistent with our data and the conclusion that, when crushing is performed at low temperature (<100 °C), high level of loss (>10%) cannot occur.

4.2. Does crushing affect ³He_c diffusivity?

The diffusion experiments performed in this study using two different crushed, irradiated olivines yielded E_a and D_0 that are not significantly different as those from the previous determinations of (Trull et al., 1991) and (Shuster et al., 2004) for matrix-sited ³He (Table 4 and Fig. 5). However these diffusion coefficients are significantly different from those obtained for magmatic helium (Fig. 6) (Trull and Kurz, 1993). This result is consistent with the previous conclusions of (Trull et al., 1991) and (Trull and Kurz, 1993) stating that, because magmatic helium is most probably trapped within fluid inclusions, this component has a different behavior than matrix-sited helium, leading to very different activation energies and frequency factors (Fig. 6). It should be noted that the D_0 from (Shuster et al., 2004) has been revised here (Table 4) using a new and more accurate estimate of the equivalent sphere radius of the grain analyzed by (Shuster et al., 2004). This new geometry shape correction from (Meesters and Dunai, 2002) yields an equivalent radius of $470\,\mu m$, leading to a revised value of $-5.1 \pm 0.3 \text{ cm}^2 \text{ s}^{-1} \text{ for } \ln D_0 \text{ (Table 4)}.$

In the range of temperature considered here (between 100 and 350 °C), the diffusion parameters obtained from the crushed olivines are similar to the ones obtained in (Trull et al., 1991) and are about 10 times higher than those

determined in (Shuster et al., 2004) (Fig. 5). It has been demonstrated that increasing the density of proton-induced lattice damage can impede the helium diffusivity in some minerals (Shuster and Farley, 2005; Shuster et al., 2006). However, although the olivines used in (Trull et al., 1991) have "natural" concentrations of cosmogenic ³He (several orders of magnitude below the concentrations of the irradiated samples), the diffusivity measured in this sample is similar to the one determined here using irradiated olivines. Consequently, because the concentration of matrix-sited ³He is correlated with the number of radiation tracks, it is improbable that the numerous proton-induced damages have biased the diffusion parameters estimated from the irradiated olivines in this study. Nevertheless, it is not straightforward to propose a mechanism that could have increased the diffusivity of olivines during crushing. Moreover, although the San Carlos and Guadalupe olivines have been crushed under different conditions, both yielded similar diffusivity. Therefore, the influence of crushing-induced fractures within the grains may explain the apparent discrepancy with uncrushed phenocrysts, rather than damage affecting the properties of the mineral lattice. Another possibility is that part (or most) of the discrepancy could be the result of natural variability. Such inter-mineral diffusion variability might result from compositional differences (Farley, 2007).

Finally, it should be noted that, if the diffusivity data previously estimated by (Shuster et al., 2004) or (Trull et al., 1991) had been used to model the release of ³He_c (Fig. 4), the estimated loss would be about one order of magnitude below the observed loss. So, even if crushing has no effect on diffusivity, the newly measured diffusion parameters appear quite relevant and should be considered as an important piece of information for addressing the behavior of ³He_c during crushing.

4.3. Comments on previous and future studies

If we accept that volume diffusion is the sole mechanism able to trigger significant ($\gg 1\%$) ³He_c loss, then the three essential points that should be checked for any crushing apparatus are (i) the actual temperature within the crushed sample, (ii) the duration of crushing and (iii) the grain size distribution of the crushed sample. These considerations are important because they can help assess whether previous determinations of cosmogenic ³He production rates were biased by unexpected crushing loss (Blard et al., 2006b). To exclude this possibility, the most important priority is to determine what temperature is actually reached by the samples within the crushing apparatus. Fig. 7 shows the modeled magnitude of loss at different temperatures, for a 10-min crush. This plot can be used to place constraints on the parameter range (T, time and grain size) that should be used to stay below an acceptable amount of loss during crushing. This calculation used the diffusion equations of (Fechtig and Kalbitzer, 1966) and the diffusion coefficient determined in this study.

Finally, it is important to acknowledge that, in the light of the previous (Scarsi, 2000; Moreira and Madureira, 2005) and the present crushing data, it seems that there

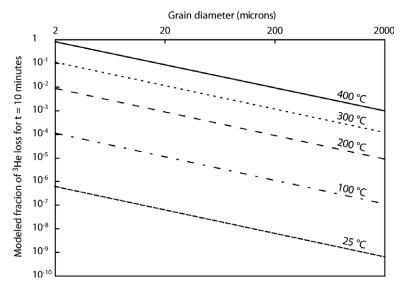


Fig. 7. Modeling of the ³He_c diffusive loss vs. the grain size at variable temperatures during 10 min. This modeling was done assuming grains have a spherical geometry, using the diffusion coefficient established in the present study from the Guadalupe sample (Table 4) and the diffusion equation of (Fechtig and Kalbitzer, 1966).

are some inescapable mechanisms that will yield a loss of about 1% (and of higher magnitude according to Yokochi et al., 2005 and Blard et al., 2006b), even if the temperature is kept low. The level of loss that can be considered as acceptable for a specific study is highly dependent on the sample characteristics and on the nature of the question being addressed. Indeed, given that the uncertainty associated with the scaling of cosmogenic production rates is at least 10%, then \sim 1% of 3 He_c loss can be considered as acceptable for studies interested in determining absolute ages measuring cosmogenic ³He. On the other hand, as demonstrated by Moreira and Madureira (2005) and Yokochi et al. (2005), if accurate measurement of magmatic ³He/⁴He ratio is the main goal of the study, even 0.5% of ³He_c loss may induce a significant bias on the measured ³He/⁴He ratio, this issue being all the more important for samples having a high ${}^{3}\text{He}_{c}/{}^{3}\text{He}_{mag}$ ratio. Similarly, uncontrolled loss of matrix-hosted radiogenic ⁴He can also bias the ³He/⁴He crushing ratio, in particular for samples that have high (U-Th)/³He_{mag} ratio and old closure ages.

5. CONCLUSIONS

Crushing experiments performed on a suite of olivines and pyroxenes with various amounts of matrix-sited $^3{\rm He}$ demonstrated that significant helium extraction (>1%) was manifest only under hot conditions ($T \ge 300$ °C). Moreover, the $^3{\rm He}_c$ neasured within each size fraction showed that the $^3{\rm He}_c$ loss (>20%) only affected the finest fraction (<10 $\mu{\rm m}$). These observations imply that the previously proposed mechanism involving spallation tracks is invalid (Yokochi et al., 2005), given the loss predicted by this model should be >10% for the <10 $\mu{\rm m}$ fraction, even at low temperature. Consequently, temperature appears to be an essential parameter that should be carefully controlled and considered in any studies involving vacuum crushing, such as cosmogenic $^3{\rm He}$ determinations. However, while volume diffusion is the predominant

mechanism responsible for helium loss, other extraction mechanisms are probably responsible for an additional $^3\text{He}_c$ loss, whose magnitude is at least $\sim\!1\%$, even under cold conditions. Moreover, for the specific case of the CRPG laboratory, a significant part of the crushing-loss magnitude observed in this lab (Yokochi et al., 2005; Blard et al., 2006b) is probably not due to volume diffusion. This suggests that the mechanisms responsible of the crushing loss may vary and be specific to the particular crushing apparatus in each laboratory. Because these mechanisms are still poorly understood, this apparently unavoidable loss should be carefully considered and tested.

Finally, the accuracy of cosmogenic ³He measurement may also benefit from alternative strategies that avoid the vacuum crushing step, such as those based on the building of cosmogenic isochrons (Cerling and Craig, 1994; Blard and Pik, 2008).

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