

Effects and bioaccumulation of Cr(III), Cr(VI) and their mixture in the freshwater mussel Corbicula fluminea

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Imad Aharchaou, A. Maul, Marie-Noëlle Pons, Danièle Pauly, Hélène Poirot, et al.. Effects and bioaccumulation of Cr(III), Cr(VI) and their mixture in the freshwater mussel Corbicula fluminea. Chemosphere, 2022, 297, pp.134090. 10.1016/j.chemosphere.2022.134090 . hal-03405490v2

HAL Id: hal-03405490 https://hal.univ-lorraine.fr/hal-03405490v2

Submitted on 28 Oct 2021

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1	Effects and bioaccumulation of Cr(III), Cr(VI) and their mixture in the freshwater mussel
2	Corbicula fluminea
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25 Abstract

26 Chromium has two main oxidation states, Cr(III) and Cr(VI), that can occur simultaneously in 27 natural waters. Current consensus holds that Cr(VI) is of high ecotoxicological concern, but 28 regards Cr(III) as poorly bioavailable and relatively non-toxic. In this work, the effects and 29 bioaccumulation of Cr(III), Cr(VI) and their mixture were studied using the freshwater clam 30 Corbicula fluminea as a model organism. Mixture exposures were carried out using solutions isotopically enriched in ⁵⁰Cr(III) or ⁵³Cr(VI), allowing to quantify the contribution of each redox 31 32 form to total Cr accumulation in the clams. Following exposure to individual redox forms, Cr(III) 33 accumulated preferentially in the digestive glands and Cr(VI) in the gills of C. fluminea. In 34 mixture exposures, both redox forms accumulated mainly in the gills; the concentration of Cr(III) 35 in the digestive glands being much lowered compared with individual exposures. Both oxidation 36 states affected the expression of biomarkers related to energy reserves, cellular damage and 37 mitochondrial functioning, as well as the expression of mRNA for detoxification genes. The 38 observed effects differed between gills and digestive glands. The present study suggests that 39 Cr(III) is a bioavailable and biologically active elemental species deserving more consideration 40 by the ecotoxicological community.

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49 Highligh

- 50 Biomarkers expression responds to Cr redox state, concentration and exposure duration
- 51 First use of isotopically enriched solutions to study Cr(III)/Cr(VI) mixtures
- 52 Both Cr(VI) and Cr(III) accumulate in gills and digestive glands of Corbicula fluminea
- 53 The presence of Cr(VI) changes the accumulation pattern of Cr(III)
- 54 Cr(III) is a biologically active species along with Cr(VI)

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73 **1. Introduction**

Chromium is a widespread environmental contaminant due to its use in several industrial 74 75 processes (WHO, 2009; 2013). In natural environments, Cr mostly occurs in two oxidation states, 76 Cr(III) and Cr(VI), that have markedly different environmental and biological behaviour. 77 Chromium(VI) is a recognized human carcinogen and has a high environmental mobility, 78 bioavailability and (eco)toxicity. These biogeochemical properties follow from chromate ions 79 being the most common chemical species of Cr(VI) (Gorny et al., 2016). Chromate ions are 80 structural analogues of sulphates and can easily cross biological membranes via facilitated 81 transport (Viti et al., 2014). Once inside the cell, Cr(VI) undergoes rapid reduction to Cr(III). The 82 reduction reactions cause oxidative stress damage and produce intracellular Cr(III) that can 83 further react with nucleic acids causing mutagenic and genotoxic effects (Kortenkamp et al., 84 1991; Sawicka et al., 2021; Zhitkovich, 2005). Chromate ions also show limited reactivity 85 towards environmental particles, which usually bear a net negative charge at pH values above 4 86 (Warren and Haack, 2001). Chromate ions can therefore undergo long-distance transport and 87 migrate into groundwater (Guo et al., 2020; Izbicki et al., 2015; Oze et al., 2007).

88 Chromium(III) is putatively less bioavailable and (eco)toxic, but increasing evidence suggests 89 that it can also elicit adverse effects in humans as well as in aquatic and terrestrial organisms 90 (Aharchaou et al., 2018; Park, 2020; Tsuchiyama et al., 2020). Mechanism of Cr(III) uptake are 91 less studied than those for Cr(VI) and may involve Cr(III)-transferrin complexes, endocytosis of 92 Cr(III)-bearing particles or uptake of hydrophobic Cr(III) complexes (Beyersmann and Hartwig, 93 2008; Viti et al., 2014). Effects of Cr(III) have also been reported on the structure and 94 permeability of biological membranes (Filice et al., 2019; Suwalsky et al., 2008). Finally, Cr(III) 95 is particle reactive and tends to adsorb onto suspended particulate matter and colloids eventually 96 accumulating in bed sediments (Dominik et al., 2007; Gustafsson et al., 2014; Liao et al., 2020).

97 Although ecotoxicological studies are biased toward Cr(VI), risk assessment has been 98 performed for both oxidation states (WHO, 2009; 2013). The World Health Organization (WHO) proposes guideline values (Probable No Effect Concentrations – PNEC) of 10 µg L⁻¹ for Cr(III) 99 and 4 µg L⁻¹ for Cr(VI). The Joint Research Centre of the European Commission proposed a 100 PNEC of 1.8 μ g L⁻¹ for Cr(III) based on the result of a chronic test with the teleost zebrafish 101 102 Danio rerio (Gomez Cortes et al., 2020). Several studies also confirmed that both Cr(III) and 103 Cr(VI) can accumulate in marine and freshwater organisms via aqueous and dietary exposure 104 routes (Aharchaou et al., 2017; Baumann and Fisher, 2011; Roditi and Fisher, 1999; Semeniuk et 105 al., 2016; Wang et al., 1997). The net bioaccumulation and biological effects of Cr in real field 106 conditions therefore likely result from the combined uptake of Cr(III) and Cr(VI). The two 107 oxidation states actually occur simultaneously, albeit in variable proportions, in natural field 108 conditions (Ferrari et al., 2019; Gunkel-Grillon et al., 2014; McClain and Maher, 2016; Saputro 109 et al., 2014). To the best of our knowledge, no existing laboratory study has yet examined the 110 effects and bioavailability of Cr during such conditions of simultaneous exposure to Cr(III) and 111 Cr(VI) that organisms often encounter in the field.

112 In principle, it is relatively easy to prepare laboratory exposure media containing both Cr 113 redox forms. On the other hand, it remains far from straightforward to understand the relative 114 contribution of Cr(III) and Cr(VI) to the total Cr body burden and biological responses of the 115 exposed organisms. In aqueous matrices, including ecotoxicological test media, the redox 116 speciation of Cr can be determined using ion Chromatography ICP-MS (Aharchaou et al., 2018; 117 Markiewicz et al., 2015; Seby et al., 2003). However, the redox speciation of Cr in the exposure 118 medium will not be retained in the test organisms, because Cr(VI) is rapidly reduced to Cr(III) in 119 the intracellular environment (Beyersmann and Hartwig, 2008; Sawicka et al., 2021; Zhitkovich, 120 2005). Tracking the fate of Cr(III) and Cr(VI) from exposure solution to biological tissues is however possible using speciated, isotopically enriched solutions (IES). Speciated isotope dilution mass spectrometry has already been used to obtain information on the relative abundance of Cr(III) and Cr(VI) in solid matrices (Martone et al., 2013; Séby and Vacchina, 2018; Unceta et al., 2010). Isotopically enriched solutions have also been used successfully to document the importance of dietary uptake of Cu and Zn by freshwater snails (Croteau et al., 2013; 2017), but similar studies on Cr are still lacking.

Chromium has four naturally occurring stable isotopes (${}^{50}Cr = 4.345\%$, ${}^{52}Cr = 83.789\%$, ${}^{53}Cr$ 127 = 9.501% and ⁵⁴Cr = 2.365%) that are widely studied in relation to cosmochemical, geochemical 128 129 and environmental processes (Cole et al., 2017; Rodríguez-González et al., 2005; Wei et al., 130 2020). Ecotoxicological studies on Cr ecotoxicity usually employ commercially available Cr(III) 131 and Cr(VI) salts that contain all four isotopes at the corresponding natural abundances. However, 132 materials enriched in one specific isotope and in a specific redox form are also available for 133 purchase (Martone et al., 2013; Novotnik et al., 2012). In the absence of redox interconversions 134 between Cr(III) and Cr(VI) in the exposure medium, the fate of each isotope (i.e., redox form) 135 can be followed from the exposure solution to the biological tissue via a simple ICP-MS analysis 136 at the right m/z value. In the present study, specimen of the freshwater bivalve Corbicula fluminea were exposed to solutions amended with Cr(III) or Cr(VI) salts having natural isotopic 137 abundances, or with a mixture of ${}^{50}Cr(III) + {}^{53}Cr(VI)$. Corbicula fluminea is a widespread filter-138 139 feeder that is largely used as a sentinel organism for freshwater biomonitoring of metal 140 accumulation and sublethal effects (de Oliveira and dos Reis Martinez, 2014; Marie et al., 2006; 141 Pereto et al., 2020). The specific objectives of the present study were to determine: (1) the effect of the individual Cr redox forms and of the ⁵⁰Cr(III) + ⁵³Cr(VI) mixture on the expression of 142 143 several biomarkers and mRNA, and (2) the bioaccumulation of Cr in the bivalves' gills and 144 digestive glands.

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146 **2. Materials and methods**

147 2.1. Organism collection and acclimation

148 Individuals of Corbicula fluminea (size 2-4 cm) were collected in the Moselle River (France) at 149 GPS coordinates 49°07'06.5" N, 6°09'39.0" E and kept until the end of sampling in unfiltered site water (pH = 8.0, conductivity = $1823 \ \mu S \ cm^{-1}$, oxygen saturation = 103%). Upon return to the 150 151 laboratory, organisms were kept in aerated water at 15 ± 0.1 °C and acclimated to the test 152 medium (Cristaline® mineral water, spring Aurèle, see composition in Table S1) by replacing 1/3 153 site water with spring water at 24 h intervals over four days (Potet et al., 2016). The viability of 154 organisms was checked at each medium renewal by testing their ability to close the valves after a 155 gentle pressure on the ligament.

156

157 2.2. Exposure set-up

158 A preliminary test was performed to establish suitable exposure conditions for the measurement 159 of sublethal effects and bioaccumulation of Cr. After acclimation (section 2.1), twelve individuals of C. fluminea were transferred to 3 L aquaria containing 1.5 L of Cristaline® mineral water 160 amended with 5, 50, 500, 1000, 2000 and 5000 µg L⁻¹ of Cr(III) or Cr(VI). Trivalent Cr was 161 162 added as CrCl₃ · 6H₂O (Alfa Aesar, purity 99.5 % min, crystalline, CAS 10060-12-5) and hexavalent Cr as K₂Cr₂O₇ (VWR, purity 99.8 % min, CAS 7778-50-9). Both Cr salts had natural 163 164 isotopic abundances (NIA). Exposure medium was continuously aerated by bubbling ambient air 165 in each aquarium. Organisms were kept at 15 °C under artificial daylight (800-1000 Lux) and a 166 $12 \text{ h light} - 12 \text{ h dark photoperiod. Organisms were not fed during the experiments. No mortality$ 167 was observed across the entire concentration range for neither Cr(III) nor Cr(VI) (data not shown) and exposure concentrations for subsequent experiments were selected as described in thenext paragraph.

170 In a first experiment (exp #1) for biomarker and mRNA measurements in the gills and digestive glands, organisms were exposed to 5, 50 and 500 µg L⁻¹ of either Cr(III) or Cr(VI) 171 172 using salts with NIA. Exposures lasted for 24 h or 96 h with daily renewal of exposure medium. 173 Chromium concentrations were selected according to three criteria: absence of mortality, 174 potential for sublethal effects and prospective environmental relevance. Twelve acclimated 175 organisms were used for each concentration and exposure time. Each Cr redox form and exposure 176 time were matched to a batch of control organisms not exposed to Cr. Other experimental 177 conditions were the same as previously described.

178 A second experiment (exp #2) examined the relationships between biomarker expression and 179 Cr bioaccumulation in the gills and digestive glands of C. fluminea. Organisms were exposed to: 50 μ g L⁻¹ of Cr(III) with NIA, 50 μ g L⁻¹ of Cr(VI) with NIA, or a mixture of 50 μ g L⁻¹ of Cr(VI) 180 + 50 µg L⁻¹ of Cr(III) using IES of ⁵³Cr(VI) and ⁵⁰Cr(III). Both IES were obtained from Applied 181 Isotope Technologies - AIT, Pittsburg, PA, USA: batch # SH1332, Lot # CR12222014JC; PN-182 183 20529, isotopic enrichment 97.7 % for ⁵³Cr(VI) and batch # 030129-04-A, Lot # Cr04272010A; PN-2; isotopic enrichment 97.3 % for ⁵⁰Cr(III). All tests lasted for 96 h and were carried out 184 185 under the same conditions as described above with 12 individuals for each scenario plus two 186 control batches of 12 individuals each. The selected concentrations were considered a good 187 compromise between practical constraints (notably the availability of IES and analytical detection 188 in biological tissues) and environmental relevance for contaminated environments (Gunkel-189 Grillon et al., 2014; Hayzoun et al., 2015).

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192 2.3. Chromium measurements in aqueous and biological matrices

193 In exp #1, total and total filterable Cr concentrations were measured at t = 0, 2, 4, 6 and 24 h (n = 194 1 to 3). Filtration was carried out using 0.45 µm membrane filters (Merck Millipore, Millex 195 33mm, PVDF, reference # SLHV033NB). For experiments lasting 96 h, measurements were 196 carried out over a single renewal period of 24 h and considered representative for all exposure 197 periods. Samples were acidified at 1 % v/v with concentrated nitric acid (65%, VWR, Suprapur) 198 and analysed by Flame Atomic Absorption Spectroscopy (FAAS; Perkin-Elmer AAnalyst 100) or 199 Graphite Furnace Flame Atomic Absorption (GFAAS; Varian spectra 880Z+). Measurement accuracy was verified using certified reference materials SPS-SW1 (2.00 \pm 0.02 µg L⁻¹) and 200 WW1 ($200 \pm 1 \text{ µg } \text{L}^{-1}$), both from Spectrapure Standards (Oslo, Norway). 201

In exp #2, Cr concentrations were measured by ICP-MS (iCAPQ, Thermo-Scientific) at t = 0 h and t = 24 h. Measurements were performed at m/z 52 for solutions amended with Cr salts having NIA and at m/z 50 and 53 for the ⁵⁰Cr(III) + ⁵³Cr(VI) mixture. All analyses were performed using Kinetic Energy Discrimination (KED) mode. Because the ICP-MS calibration was performed using solutions with NIA, concentration results for ⁵⁰Cr and ⁵³Cr were corrected to account for the use of IES. A correction for signal drift in the internal standard was also applied using the following formulas:

$$209 \quad {}^{50}\text{Cr}_{\text{corrected}} = ({}^{50}\text{Cr}_{\text{ICP-MSsoftware}} \cdot 4.345) \cdot (100/\text{baseline drift})$$
(Eq. 1)

210
$${}^{53}Cr_{corrected} = ({}^{53}Cr_{ICP-MSsoftware} \cdot 9.500) \cdot (100/baseline drift)$$
 (Eq. 2)

211
$${}^{52}\text{Cr}_{\text{corrected}} = ({}^{52}\text{Cr}_{\text{ICP-MSsoftware}}/\text{baseline drift})$$
 (Eq. 3)

212 Where:

^{xx}Cr_{corrected} are the Cr concentrations corrected for isotopic abundances and for the ICP-MS
baseline drift (see below);

x^xCr _{ICP-MSsoftware} are the Cr concentrations calculated by the ICP-MS software based on normal
 isotopic abundances;

²¹⁷ 'baseline drift' is the ratio between raw counts of the internal standard (45 Sc) for a given sample ²¹⁸ divided by the average raw counts for the calibration standards (n = 7) at the beginning of the ²¹⁹ ICP-MS run. All Cr measurements in exposure solutions were performed on aliquots sampled ²²⁰ from aquaria containing living organisms.

221 Chromium levels in gills and digestive glands of C. fluminea were assayed by ICP-MS at m/z 222 50, 52 and 53 using KED mode after mineralization with 0.5 mL of concentrated nitric acid (65% 223 v/v, Suprapur, VWR) and 0.1 mL of H₂O₂ (30% v/v, Suprapur, VWR). Mineralization was 224 carried out in acid-washed polystyrene tubes placed in a stove at 65 °C overnight. Raw analytical 225 results were corrected as necessary using equations 1 to 3. Gills and digestive glands were 226 recovered by immediately dissecting exposed organisms at the end of the exposure period. 227 Tissues were frozen at -20 °C and then freeze-dried over 24 h (CRIOS-80, Cryotech). The 228 accuracy of the mineralization procedure was checked using the reference material TORT-2 229 (lobster hepatopancreas) from the National Research Council of Canada (NRC-CNRC). The certified value for total Cr was 0.77 ± 0.15 mg kg⁻¹ (uncertainty expressed as 95% confidence 230 231 limits). The ICP-MS detection and quantification limits were calculated as three and ten times the 232 standard deviation of replicate analytical blanks (n = 14).

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234 2.4. Biomarker expression

After each exposure period, gills and digestive glands were pooled from two individuals of the same batch to obtain sufficient material for biomarker measurement and stored at -80 °C pending analysis. A total of four samples per exposure conditions was therefore available for biomarker analysis. Immediately before analysis, samples were thawed on ice and their wet weight 239 determined. Tissues were then transferred to microtubes (2 mL) and manually crushed in 50mM 240 phosphate buffer (98% v/v, pH = 7.6) supplemented with 1% v/v of phenylmethylsulfonyl 241 fluoride (PMSF) and 1% v/v of a 1 mM mixture of L-serine and borate to inhibit proteases 242 (Garaud et al., 2015; Sroda and Cossu-Leguille, 2011). The quantity of added buffer was adjusted 243 to the mass of tissue based on a ratio of 4 mL of buffer to 500 mg of tissue (wet weight). Samples 244 were then centrifuged for 5 min (250 g) at 4 °C. Part of the resulting tissue homogenate was used 245 as such for biomarker analysis. A second aliquot of homogenate was again centrifuged for 20 min 246 (1000 g) at 4 °C. The resulting supernatant was subject to a final centrifugation (50 min, 20000 g, 247 4 °C) to recover the cytosolic fraction for the measurements of enzymatic activities.

248 Tissue homogenates were used to assay markers of cellular damage (LOOH, Lipid 249 Hydroperoxide; ACP, Acid phosphatase), the status of energy reserves (Prot1, proteins; TG, 250 triglycerides), Cholesterol (Chol.), and the activity of the mitochondrial Electron Transport 251 Systems (ETS). Other markers of oxidative stress (TAC, Total antioxidant capacity; GPx-tot, 252 Glutathione peroxidase; GST, Glutathione S-transferase; CAT, Catalase), cellular damage (CSP-253 3, Caspase-3) and energy reserves (Prot2, proteins; LDH, Lactate dehydrogenase) were 254 measured in the cytosolic fraction. All biomarkers were analysed using an automated 255 spectrophotometer (Konelab 20 XTi, Thermo Scientific) according to the methods described in 256 Garaud et al. (2015), Bertrand et al. (2016) and references therein.

257

258 2.5. RNA extraction and RT-qPCR analysis

Gills and digestive glands were excised from 3 individuals per condition after 96 h exposure (exp 41). Non-digestive tissues were carefully removed from digestive glands. Samples were then transferred separately into 1 mL of 4M guanidine thiocyanate and stored at -80° C. Total RNAs were extracted using a RNeasy mini kit (Qiagen, Hilden, Germany) following manufacturer's instructions. Genomic DNA was digested after extraction with DNAse I (Qiagen) and a new total
RNA purification was performed with RNeasy mini kit. Total RNA purity and quantity were
assessed by optical density measurements on a Nanodrop spectrophotometer (A₂₆₀ nm and
absorbance ratio 260/280 and 260/230). RNA integrity was evaluated by capillary electrophoresis
using a Bioanalyzer 2100 (Agilent, CA, USA).

268 RT-qPCR was performed on 8 detoxification genes (GST-pi, CAT, Mn SOD, CuZn SOD, TR, 269 MT, HSP70 and Se-GPX) and PS3 as reference gene using the methodologies described in 270 Koehle-Divo et al. (2019) and the sequences reported in Table S2. cDNA was reverse-transcribed 271 from total RNA (900 ng) using the Superscript IV Reverse Transcriptase with oligo(dT)₁₈ 272 according to the manufacturer recommendations (Invitrogen, Carlsbad, CA, USA). Real-time 273 PCR was performed on a StepOnePlus PCR system using the Fast SYBR® green master mix 274 (Applied Biosystem®, CA, USA). Amplifications were performed with 50 ng of template, or 275 water (no template control), or 50 ng total RNA (no RT control) using 200 to 300 nM of primers 276 in a final volume of 20 µL (Koehle-Divo et al., 2019). The cycling conditions were 20 s at 95°C, 277 followed by 40 cycles of 10 s at 95°C (except for CuZn SOD, HSP70, PS3 and GST-pi: 3 s at 278 95°C) and 30 s at 60° C. A melting curve analysis (65° C–95° C) was performed to ensure the 279 absence of artefacts and the expression of the genes of interest was normalized to that of PS3 using the $2^{-\Delta\Delta Ct}$ method. 280

281

282 *2.6. Statistical methods and data treatment*

Correlation circle graphs among biomarkers (exp #1 and 2) and detoxification genes (exp #1) were obtained by Principal Component Analysis (PCA) using the R-package 'FactoMineR'. Biomarkers response was modelled for both exp #1 and 2. For exp #2, accumulation of 'total chromium' was added to the list of modelled variables. The values, levels and coding of the 287 factors considered in the models for the two experiments are presented in Table S3. The levels mentioned in Table S3 were then arranged according to a $2^2 \times 4$ (exp #1) and a 2^3 (exp #2) 288 289 randomized factorial design with 4 and 3 replicates per combination of the factors in exp #1 and 290 exp #2, respectively. The expression of the 8 mRNAs was also modelled for exp #1, but with a 291 single exposure duration (96h) and three replicates for each factor combination. Parametric 292 statistical methods and regression analysis were performed for biomarkers, mRNAs and Cr 293 accumulation as described below. The homogeneity of variances, associated with all the factor 294 combinations within the replicate groups, was preliminary verified using the modified Levene's 295 test (i.e., Brown-Forsythe's test for homogeneity of variances) after log transformation of the 296 explained variables.

In exp #1, the biomarkers' responses were modelled separately for gills and digestive glandsusing the following log-linear regression model:

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300
$$\text{Log}_{10} \text{ (biomarker)} = \beta_0 + \beta_1 D + \beta_2 Cr + \beta_3 Co + \beta_{12} D^* Cr + \beta_{13} D^* Co + \beta_{23} Cr^* Co + \beta_{123} D^* Cr^* Co$$

301 (Eq. 4a)

302 where, (D) is the exposure duration, (Cr) indicates the chromium redox form, (Co) the exposure 303 concentration and the '*' sign indicates the interactions between factors (Table S3).

For modelling mRNA responses, Eq. 4a simplified to Eq. 4b because the model did not includeexposure duration:

306

307
$$Log_{10} (mRNA) = \beta_0 + \beta_1 Cr + \beta_2 Co + \beta_{12} Cr^* Co$$
 (Eq. 4b)

309 In exp #2, a log-linear regression model was used to describe the response of each biomarker 310 (Eq. 5a) and of total Cr bioaccumulation (Eq. 5b) as a function of the tissue (T) and the 311 presence/absence of trivalent (Cr3) and hexavalent (Cr6) chromium. 312 $Log_{10} (biomarker) = \beta_0 + \beta_1 T + \beta_2 Cr3 + \beta_3 Cr6 + \beta_{12} T^* Cr3 + \beta_{13} T^* Cr6 + \beta_{23} Cr3^* Cr6 + \beta_{13} T^* Cr$ 313 $\beta_{123}T*Cr3*Cr6$ 314 (Eq.5a) 315 $Log_{10} (total Cr) = \beta_0 + \beta_1 T + \beta_2 Cr3 + \beta_3 Cr6 + \beta_{12} T^* Cr3 + \beta_{13} T^* Cr6 + \beta_{23} Cr3^* Cr6 + \beta_{13} T^* Cr6$ 316 β₁₂₃TxCr3*Cr6 317 (Eq.5b) 318 319 The testing of the regression coefficients (*i.e.*, the β s') and the subsequent construction of 320 reduced regression models was performed using a stepwise backward procedure with a 321 significance threshold set at $\alpha = 0.01$ for the statistical tests. All statistical analyses were performed using R software (version 3.0.0). 322 The Cr bioconcentration factors (BCF, in L g⁻¹) were calculated as: 323 324 $BCF = \frac{C_{biota}}{C_{water}}$ 325 (Eq. 6) where C_{biota} and C_{water} are the concentrations measured in biological material (in $\mu g \ g^{\text{-}1})$ and in 326 water (in $\mu g L^{-1}$), respectively. 327 328 Chromium speciation calculations were performed using VisualMinteq version 3.1. No changes were made to the default database and the precipitation of amorphous Cr(OH)₃ was 329

allowed as possible solid species. The average composition of Cristaline® mineral water (Table

S1) was used to describe the characteristics of the aqueous matrix and the Debye-Hückelequation was used for activity correction.

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- 334

335 **3. Results and discussion**

336 *3.1.* Chromium concentrations and behaviour in test solutions

Figure of merits for FAAS and GFAAS were the same as previously reported (Aharchaou et 337 338 al., 2018) and analytical blank values were below detection limits. Measured Cr concentration in certified waters (n = 1) were 1.96 μ g L⁻¹ for SPS-SW1 (98 % accuracy) and 202 μ g L⁻¹ for WW1 339 (101 % accuracy). Detection and quantification limits for analyses by ICP-MS were: ${}^{50}Cr = 0.074$ 340 and 0.247 μ g L⁻¹, ⁵²Cr = 0.037 and 0.124 μ g L⁻¹, ⁵³Cr 0.034 and 0.115 μ g L⁻¹. Analytical blank 341 values were below detection limits for ${}^{50}Cr$ and for ${}^{53}Cr$ at t = 0; between detection and 342 quantification limits for ⁵³Cr at t = 24 and measurable for ⁵²Cr. Results for experiments carried 343 out using Cr salts with NIA (for which ⁵²Cr was used) were therefore corrected for the 344 corresponding average analytical blanks. A 200 μ g L⁻¹ standard solution was repeatedly analysed 345 (n = 5) during the ICP-MS run with average values of $238 \pm 6 \mu g L^{-1}$ at m/z = 52, $197 \pm 4 \mu g L^{-1}$ 346 at m/z = 50, and $196 \pm 6 \ \mu g \ L^{-1}$ at m/z = 53. 347

Total and filterable Cr concentrations at t = 0 were slightly lower than expected for solutions amended with 50 or 500 µg L⁻¹ in exp #1 and within 15% of the expected values in the other cases (Table S4). Over the 24 h water renewal periods, total and filterable Cr levels varied by less than 15–20 % in the case of Cr(VI), but decreased by 70–95 % in the case of Cr(III) (Figure 1, Table S4). Measurements taken at t = 0.5, 2, 4 and 6 h during exp #1 suggest that the concentration decrease began early after Cr(III) addition into the exposure medium (Table S4). The concentrations of the two Cr redox form showed a similar behaviour when added alone or in combination to the exposure medium (Figure 1; Table S4). A dilution error during the use of IES (see section 2.2) resulted in an initial total filterable Cr(III) concentration of $104 \pm 1.9 \ \mu g \ L^{-1}$ instead of the targeted 50 $\ \mu g \ L^{-1}$ value (Figure 1B). Once analytically identified, the higher initial Cr(III) concentration did not prevent the study of the interactions between Cr(III) and Cr(VI) on biomarker expression nor of the simultaneous bioaccumulation of Cr(III) and Cr(VI) by *C*. *fluminea*.





Figure 1. Panel A: temporal evolution of total (NF) and filterable (F, < 0.45 μ m) Cr concentrations in solutions amended with 50 μ g L⁻¹ of either Cr(III) or Cr(VI) with natural isotopic abundances (arithmetic means ± 1 standard deviation, n=3 for each point). Symbols for total and filterable values have been slightly offset for representation purposes, but samples for measurements were collected at the same time. Panel B: filterable Cr concentrations measured in solutions amended with a mixture of ⁵⁰Cr(III) (100 μ g L⁻¹) and ⁵³Cr(VI) (50 μ g L⁻¹) using isotopically enriched solutions. See Table S4 for tabulated data.

The decrease in total and filterable chromium concentrations in all solutions amended Cr(III) agrees with the known tendency of Cr(III) to form insoluble Cr(OH)₃ species at circumneutral pH (Rai et al., 1989). Speciation calculation predicted that approx. 81 % and 98 % of the added

374 Cr(III) would precipitate for initial concentrations of 50 and 500 μ g L⁻¹. According to 375 VisualMinteq calculations, the theoretical maximum soluble Cr(III) concentration in Cristaline® 376 water is about 10 μ g L⁻¹ at a pH of 7.4 (measured at the source); in reasonable agreement with 377 experimentally measured Cr concentrations at 24 hours (Fig. 1 and Table S4). On the other hand, 378 the model does not predict the formation of any precipitate for added Cr(III) concentrations of 5 379 μ g L⁻¹, in disagreement with the experimental results (Table S4).

380 Because VisualMinteq is a thermodynamic model, the agreement between theoretical 381 solubility and experimentally measured concentrations at t = 24 h suggests that the exposure 382 system reached quasi equilibrium within 24 hours. The decrease in filterable Cr concentrations 383 agrees with the hypothesis of a rapid formation of Cr(III) bearing particles in the test medium 384 (Aharchaou et al., 2018). On the other hand, the decrease in total Cr concentrations suggests that 385 other phenomena also occurred during the experiments (e.g., adsorption onto the vessels' walls 386 and on the organisms' shells). These phenomena may also explain the observed concentration decrease in solutions initially amended with 5 μ g L⁻¹ of Cr(III); *i.e.* a concentration below the 387 theoretical solubility limit. 388

389

390 *3.2. Chromium bioaccumulation and bioavailability*

391 Chromium levels in control organisms were above the analytical quantification limits (see 392 section 3.1) only for 52 Cr: 0.74 ± 0.17 µg g⁻¹ for gills (G) and 0.32 ± 0.09 µg g⁻¹ for digestive 393 glands (DG). Results for Cr accumulation were therefore corrected for the corresponding blank 394 values in specimen exposed to solutions prepared using Cr salts with NIA (for which 395 measurements were performed at m/z 52). No blank corrections were necessary for specimen 396 exposed to 50 Cr(III) and 53 Cr(VI) using IES. Duplicate analysis of the reference material TORT2 397 gave blank corrected Cr concentrations (measured at m/z = 52) of 0.643 and 0.740 µg g⁻¹ 398 corresponding to percentage recoveries of 83 and 96 %, respectively.

For specimen exposed to individual Cr redox forms, Cr(III) accumulation was higher in DG than in G (10.2 ± 5.5 vs. $2.98 \pm 1.05 \ \mu g \ g^{-1}$), while the opposite situation was observed for Cr(VI) (2.64 ± 1.11 vs. $12.9 \pm 0.19 \ \mu g \ g^{-1}$) (Fig. 2). In organisms exposed to the Cr(III) and Cr(VI) mixture, Cr accumulation was higher in G than in DG for both redox forms (Mann-Whitney-Wilcoxon test; P=0.029).





Figure 2. Arithmetic means of Cr concentrations accumulated in gills (hatched bars) and digestive glands (full bars) of individuals of *Corbicula fluminea* exposed for 96 hours to solutions amended with either Cr(III) or Cr(VI) salts having natural isotopic abundances or to a solution simultaneously amended with a mixture of 50 Cr(III) and 53 Cr(VI) using isotopically enriched solutions (see Table S4, exp #2, for actual exposure concentrations in each scenario). Errors are expressed as \pm 1 standard deviation. Analytical results for individual samples are given in Table S5. NIA, solutions with Natural Isotopic Abundances; IES, Isotopically Enriched Solutions.

413

414 The fitted bioaccumulation model (Eq. 5b) obtained from experimental data was:

415

416 $Log_{10}(total Cr) = 0.405 - 0.164 T + 0.249 Cr3 + 0.286 Cr6 + 0.092 T*Cr3 - 0.166 T*Cr6 - 0.271$ 417 Cr3*Cr6 (Eq. 7)

418

419 with a high explanatory power and predictive capacity (F test, $P < 10^{-4}$; $r^2 = 0.954$; coefficient of 420 variation of the residuals = 34.11%). Except for the three-way interaction T*Cr3*Cr6 (P=0.03), 421 all factors had a statistically significant effect (P<0.01) on the bioaccumulated total Cr 422 concentration and could be ranked as follows: $Cr6 \ge Cr3*Cr6 \ge Cr3 > T*Cr6 \ge T > T*Cr3$. The 423 results of the fitted model agreed well with the experimentally measured total Cr concentrations 424 (Table 1).

425 The fitted bioaccumulation model (Eq. 7) suggests that both direct and interaction effects 426 control Cr(III) and Cr(VI) bioaccumulation by C. fluminea. The interaction effects between the 427 Cr redox forms and the tissues (e.g., T*Cr3 and T*Cr6) agree with the experimental observation 428 that exposure to Cr(III) or Cr(VI) led to a preferential accumulation of Cr in DG or G, 429 respectively. Equation 7 also suggests the presence of an antagonistic effect between Cr(III) and 430 Cr(VI) (e.g., Cr3*Cr6) with a magnitude (*i.e.*, the value of the coefficient in Eq. 7) being 431 comparable to the direct effects of both redox forms. This antagonistic effect is confirmed by the 432 experimental results (Table 1). Specifically, the accumulation of Cr(VI) decreased by about 50% 433 in both G and DG of organisms exposed to the Cr(III)+Cr(VI) mixture compared with the 434 Cr(VI)-only exposure. The presence of Cr(VI) did not affect Cr(III) levels in G, but lowered 435 Cr(III) accumulation in DG by about 10-fold. This antagonistic effect is unlikely to arise from 436 differences in the uptake of the three Cr isotopes considered in the present study. Indeed, in specimen exposed to Cr(VI) or Cr(III) solutions prepared using salts with NIA, the coefficient of
variation of the measured Cr concentrations obtained at m/z 50, 52 and 53 was between 8 and
12%; *i.e.*, not large enough to account for the observed decreases in Cr(III) and Cr(VI)
bioaccumulation during mixture exposure.

441 Although the cellular uptake mechanisms for Cr(VI) and Cr(III) are known to differ 442 (Beyersmann and Hartwig, 2008; Viti et al., 2014), further research is required to understand the 443 mechanisms behind the observed antagonist effect. Future studies should also clarify if the 444 accumulation of Cr(III) solely reflects absorption of soluble elemental species inside the tissues 445 or if adsorption/absorption of Cr-bearing nanoparticles (see section 3.1) also contributes to the Cr 446 body burdens of C. fluminea. The role of Cr-bearing nanoparticles in Cr(III) bioaccumulation 447 may have a real environmental relevance because polymeric Cr(III) species can occur in 448 environmental samples (Hu et al., 2016). Previous studies on Cr bioaccumulation by other bivalves confirm the bioavailability of Cr(VI) and Cr(III) from dissolved or particulate pathways 449 450 (Parlak et al., 1999; Roditi and Fisher, 1999; Wang et al., 1997). Extending the isotopic approach 451 to simultaneous exposure to dissolved Cr(VI) and particulate-bound Cr(III) would allow to better 452 understand Cr bioaccumulation in real field conditions where simultaneous exposure to Cr(III) 453 and Cr(VI) can occur.

454

455

Table 1. Total chromium concentrations (95% confidence intervals are given in brackets) accumulated in gills and digestive glands for each combination of factors in experiment 2 as estimated by the fitted reduced model described in Eq. 7. The corresponding arithmetic means of measured concentrations (in italics) are reported for comparison purposes (see Table S5). Nominal exposure concentrations were 0 or 50 μ g L⁻¹ except for '\$' that was 100 μ g L⁻¹ (see 461 section 2.2). Concentrations in organisms exposed to either Cr(III) or Cr(VI) have been corrected 462 for the corresponding blank values of control organisms, while no blank corrections were 463 necessary for organisms exposed to the Cr(III)+Cr(VI) mixture (see section 3.2).

			Total accumulated Cr concentration ($\mu g g^{-1}$)					
Cr(III)	Cr(VI)			Gills	Digestive glands			
			Mean	95% CI	Mean	95% CI		
	20	Estimated	0.488	[0.335; 0.710]	0.323	[0.222; 0.471]		
по	110	Measured	0.737		0.319			
	Vac	Estimated	13.6	[9.36; 19.8]	1.96	[1.35; 2.85]		
110	yes	Measured	12.9		2.64			
	20	Estimated	3.51‡	[2.41; 5.1]	5.42	[3.72; 7.88]		
yes	по	Measured	2.98		10.2			
		Estimated	8.06	[5.54; 11.7]	2.70	[1.85; 3.93]		
ves ^{\$}	Ves	Measured	8.80 ^A		2.02 ^A			
yes	yes		2.49 ^B		0.737 ^B			
			6.31 ^C		1.28 ^C			

464

465 A, total Cr = 50 Cr(III) + 53 Cr(VI); B, 50 Cr(III); C, 53 Cr(VI)

[‡] Example of calculation using Eq. 7. Remembering that the actual numerical values of the factors Cr3, Cr6 and T in Eq. 7 correspond to the codes in Table S3, total Cr accumulation in gills (code = -1) for organisms exposed to 50 μ g L⁻¹ of Cr(III) (code = +1) and 0 μ g L⁻¹ of Cr(VI) (code = -1) leads to the following development for Eq. 7: Log₁₀(total Cr)=0.405 - [0.164×(-1)] + [0.249×(+1)] + [0.286×(-1)] + [0.092×(-1) ×(+1)] - [0.166×(-1)×(-1)] - [0.271×(+1)×(-1)] = 0.545; or 3.51 μ g g⁻¹ on a linear scale.

Organisms exposed to Cr(III) showed a higher BCF in DG than in G (0.5 vs. 0.15 L g⁻¹), while 473 the opposite was true for Cr(VI) (0.05 vs. 0.22 L g⁻¹). In exposures to the Cr(III) + Cr(VI) 474 475 mixture, both redox forms showed higher BCFs in G than in DG: 0.08 vs. 0.02 for Cr(III) and 0.14 vs. 0.03 for Cr(VI) (all values in L g⁻¹). Calculations of BCF for Cr(III) were done using the 476 477 geometric means of the measured filterable concentrations at t = 0 and t = 24 h to account for the 478 temporal decrease in Cr concentration (Fig. 1, Table S4). Existing studies on Cr accumulation in 479 bivalves are biased toward Cr(VI) or do not specify Cr speciation in the exposure matrices (Table 480 S6). Potet et al. (2016) exposed two species of dreissenids to Cr(VI) using experimental 481 conditions similar to those adopted in the present study. Despite the differences in the measured 482 Cr burdens, the BCF factors for the two dreissenids (Table S6) are similar to those observed in 483 the present study for C. *fluminea* exposed to Cr(VI). de Oliveira and dos Reis Martinez (2014) 484 exposed specimen of C. fluminea along a contamination gradient in a stream impacted by landfill 485 leachate and observed BCF much higher than those reported in the present study (Table S6). We 486 surmise that, during such in situ exposures, the bulk of accumulated Cr originated from particle 487 ingestion rather than uptake of aqueous Cr species. Uptake of particle-bound Cr(III) has been 488 confirmed using radiolabelled material in laboratory studies (Roditi and Fisher, 1999; Wang et 489 al., 1997). Comparison with other published results is less straightforward because of interspecies 490 differences and variability of exposure conditions (Table S6). Nonetheless, our results clearly 491 highlight the need for more research on the bioavailability of Cr(III) or Cr(III) + Cr(VI) mixtures. 492

493 *3.3. Biomarker responses*

Biomarkers expression could be properly measured except for TAC in gills (both experiments)
and CAT in gills and digestive glands (exp #2 only). The expression of energy related biomarkers
(Prot 1, Prot 2, TG, LDH), Chol and some markers of oxidative stress (TAC, GST and CAT) was

497 higher in digestive glands (DG) than in gills (G) for both experiments and regardless of exposure 498 conditions (Tables S7 and S8). Among the markers of cellular damage, ACP activities were 499 similar in the two tissues, while CSP-3 levels were higher in G than in DG for both experiences. 500 Lipid hydroperoxide (LOOH) behaved differently in exp #1 (DG > G) and exp #2 (G > DG); 501 most likely because of the changes in LOOH levels among the specimen of C. fluminea not 502 expose to Cr between the two experiments (Tables S7 and S8). In some cases, the experimental 503 values were too low for quantification (blank cells in Tables S7 and S8) and the corresponding 504 results were excluded from further statistical analysis.

505 For exp #1, the first two PCA factors explained 58.2% and 66.3% of the inertia for G and DG, 506 respectively. Apart from Chol, Prot1, Prot2, TG and CAT, LDH, the associations between the 507 biomarkers differed between tissues (Fig. 3). Note however that CAT and LDH were negatively 508 correlated in G and positively correlated in DG. ACP and CSP3 were associated with Chol, 509 Prot1, Prot2, TG in G and with CAT and LDH in DG. LOOH was linked to either GST (G) or 510 ETS (DG) and seemed to respond differently to Cr exposure compared with other biomarkers. In 511 DG, the first PCA axis opposed some energy-related biomarkers (Prot, TG) and Chol to those 512 indicative of oxidative or cellular damage (Fig. 3B).

513 Regression analysis did not yield statistically significant models for Chol, LDH, TG and, on a 514 tissue-specific basis Prot 1, Prot 2, GST and LOOH (Table S9). These biomarkers appear 515 unsuitable to detect adverse effects of Cr in G and/or DG of C. fluminea. Regression analysis was 516 not performed for CAT and TAC because the missing values prevented comparisons between 517 tissues and experiments. Statistically significant direct effects of Cr redox form and/or Cr 518 concentration were observable for ACP and CSP3 in gills, while exposure duration had a direct 519 effect on ACP in gills and ETS in both gills and digestive glands (Table S7). The responses of 520 other biomarkers in exp #1 were mostly driven by interaction effects among exposure duration,

521 Cr redox state and Cr exposure concentration (Table S9). In particular, interaction effects 522 involving Cr(VI) or Cr(III) concentration were identified for Prot1, LOOH, and CSP3 in DG and 523 Prot2, ACP, CSP3 and ETS in G (Table S9). In the other cases, interaction effects involved only 524 exposure duration and/or Cr oxidation state.



Figure 3. Correlation circle graphs of the PCA (A: gills; B: digestive glands) for the biomarker activities measured in specimen of *Corbicula fluminea* exposed to Cr(III) and Cr(VI) in experiment 1.

529

Interpreting interaction effects is a complex task and is better illustrated graphically for some biomarkers with the highest r^2 ; e.g. ETS and CSP3 in G and ETS and LOOH in DG (Table S9 and Fig. 4). The activities of ETS in G at 96 h decreased with increasing Cr(III) concentrations and increased with increasing Cr(VI) concentrations, but showed no clear trends at 24 h. The response of CSP3 to Cr(III) showed opposite trends at 24 h and 96 h, while CSP3 levels in organisms exposed to Cr(VI) increased with concentration regardless of exposure duration. In DG, ETS levels peaked in specimen exposed to 50µg L⁻¹ Cr(III) for 96 h, but no clear trends emerged for the other exposure conditions. Similarly, LOOH expression increased monotonically
with Cr(III) concentration for 96 h exposures, but otherwise had variable patterns.



Figure 4. Interaction plots for the responses of ETS and CSP3 in gills (A; top graphs) and of ETS
and LOOH in digestive glands (B; bottom graphs) for different combinations of the Cr redox
forms and the exposure time, as a function of Cr concentration in experiment 1.

Wang et al. (2012) observed that exposure to Cr(VI) increased lipid peroxidation and decreased catalase activity in *C. fluminea*. Effects of Cr(VI) exposure on various biomarkers have been previously reported in *Mytilus galloprovincialis* and *Dreissena* spp. (Barmo et al., 2011; Ciacci et al., 2011; Ciacci et al., 2012; Potet et al., 2016). The present study suggests that Cr(III) can also affect biomarker expression in bivalves and should be considered as a biologically active chromium form deserving more attention in future studies.

For exp #2, neither Cr(III) nor Cr(VI) showed any direct effect on biomarker responses, but differences in biomarker expression existed between tissues. Expression levels of Chol, TG, Prot1, Prot2 and GST were significantly higher ($P < 10^{-4}$) in DG than in G, while the opposite (P < 0.01) was observed for LOOH, CSP3 and ETS (Table S9). The simultaneous exposure to Cr(III) and Cr(VI) did not seem to modify biomarkers responses compared with exposure to individual redox forms of Cr; at least for the selected aqueous exposure levels and the corresponding bioaccumulated concentrations (see section 3.5).

557

558 *3.4. Gene expression*

The expression levels of all mRNAs could be properly quantified except for Se-GPX in G. The first two PCA factors explained 75% and 72% of the inertia for the G and DG, respectively (Fig. 5). The expression patterns of mRNAs differed between the two tissues. For example, the correlations between Mn SOD and CAT as well as TR and CuZn SOD were observed only in G, while Mn SOD and MT were correlated only in DG.

Regression models (Table S10) also pointed to tissue-dependent relationships between mRNA expression and Cr exposure. In the gills, expression of mRNAs for GST-pi, CAT, CuZn SOD, Mn SOD, MT and HSP70 increased with Cr(III) and slightly decreased with Cr(VI) concentrations. In digestive glands, significant regression models were obtained only for TR, MT and HSP70. The expression of TR and HSP70 increased with Cr concentration for both redox states, although the response was higher in the presence of Cr(VI). At the opposite, MT expression decreased with Cr concentration regardless of the element oxidation state (Table S10).





Figure 5. Correlation circle graphs of the PCA (A: gills; B: digestive glands) for the mRNA
expression in specimen of *Corbicula fluminea* exposed to Cr(III) and Cr(VI) in experiment 1.

574

575 Changes in the expression levels of GST-pi, CAT, MnSOD, CuZnSOD, MT, TR and HSP70 576 were driven by both Cr redox state and concentration in at least one tissue, while MT expression 577 in digestive glands was driven only by Cr concentration (Table S10). The effects on mRNA 578 expression are mostly marked in the gills compared to the digestive glands (Table S10). 579 Modifications of GST-pi, CAT, MT and HSP70 mRNA levels have been reported in the gills and 580 digestive glands of Mytilus galloprovincialis exposed to Cr(VI) (Ciacci et al., 2012; Franzellitti et 581 al., 2012), while we are unaware of previous report for Cr(III) effects on mRNA expression in C. 582 fluminea or other bivalves. Ciacci et al. (2012) also observed that the effects of Cr(VI) on mRNA 583 GST-pi, CAT, MT isoforms and HSP70 differed between male (decreased expression) and 584 female (increased expression); a distinction that was not taken into account in the present study.

585

586

587 **5.** Conclusions

588 Both Cr(III) and Cr(VI) can accumulate in gills and digestive glands of the freshwater mussel 589 Corbicula fluminea and affect biomarkers' and mRNA expression. Future research on Cr in 590 bivalves and, likely, other aquatic organisms should therefore consider both redox forms and not 591 focus exclusively on Cr(VI). Exposure to Cr(III)+Cr(VI) mixtures does not markedly modify the 592 accumulation of Cr(III) and Cr(VI) in gills compared with exposures to individual redox forms. 593 On the other hand, the presence of Cr(VI) seems to decrease Cr(III) accumulation in digestive 594 glands. Typical laboratory-based scenarios may therefore miss possible interactions between Cr(VI) and Cr(III) when exposing organisms to individual Cr redox forms. This caveat applies to 595 596 the study of Cr uptake, bioaccumulation and ecotoxicity in both laboratory and real field 597 conditions. Studies under more diversified exposure scenarios, including exposure via dietary 598 pathways, are required to confirm the general validity of this hypothesis.

The regression approach suggests that biomarkers for the activity of mitochondrial electron transport systems (ETS) and cellular damage (ACP and CSP-3) are the most suitable to study the effect of Cr on *C. fluminea*. However, the biomarker response varies depending on the selected tissue and must also consider the complex interactions between Cr oxidation state, Cr concentration and exposure duration. In the case of antioxidant genes, regression analysis suggests that CAT and MnSOD mRNA levels are the most relevant to study in the gills. For digestive glands, regression for mRNA levels of TR and HSP70 were the most significant. In all

606	cases, regression analysis suggest that mRNA levels respond to both Cr(III) and Cr(VI), further							
607	supporting the recommendation for more comparative studies on both Cr oxidation states.							
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610	Acknowledgements							
611	This work has been performed as a part of the PhD thesis of Imad Aharchaou who received							
612	financial support from the French Research Ministry. The research was partially supported by the							
613	French National Research Agency through the national program "Investissements d'avenir" with							
614	the reference "ANR-10-LABX-0021/ LABEX RESSOURCES21", excellence action 86.							
615								
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617	CRediT author statement							
618	IA: Conceptualization, Investigation, Validation, Writing (Original Draft)							
619	AM: Formal analysis, Visualization, Writing (Original Draft)							
620	M-N P: Resources, Validation, Writing (Review & Editing)							
621	DP: Investigation							
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624	FR: Investigation, Methodology, Writing (Original Draft)							
625	PR: Investigation							
626	MB: Investigation							
627	EB: Conceptualization, Supervision, Validation, Funding acquisition, Writing (Original Draft)							
628	DAL V: Conceptualization, Supervision, Project Administration, Funding acquisition,							
629	Validation, Visualization, Writing (Original Draft)							

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Supporting information for

811	Effects and bioaccumulation of Cr(III), Cr(VI) and their mixture in the freshwater mussel
812	Corbicula fluminea
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- 858 Table S1: Composition of Cristaline® mineral water (spring Aurèle) used as a test medium for
- experiments on biomarker activities in and Cr accumulation by specimen of *Corbicula fluminea*. All values are in mg L^{-1} .

	Element	Concentration
	Са	106
	Mg	4.2
	Na	3.5
	K	1.5
	HCO ₃	272
	Cl	3.8
	SO ₄	50
	NO ₃	1
	Si	8.3
	F	0.9
	рН	7.2
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Table S2. Primer sequences used in RT-qPCR analysis.

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	Accession			Amplicon	Concentration		
	Number			length	used for qPCR		
Gene Name	(GenBank)	Forward (5'-3')	Reverse (5'-3')	(bp)	(nM)	GO biological process	Ref.
						Response to	
Catalase	EF446609	CCTTCAGTGCGTAGTCTGG	CCTGTTCTCTGACCGTGG	196	200	oxidative stress	1
Cu-Zn Superoxide						Response to	
dismutase	KF218347.1	GAATGTTGTGGCTGATGAGG	GGATCGTTCTCCTGTCAATG	76	250	oxidative stress	2
						Response to	
Se-GPx	EF547366.1	TATGCCCTGTTCCCGTGTTC	CTGCCATCTGGACCAATCAG	148	200	oxidative stress	3
	AN					Response to reactive	
GST pi	AY885667.1	CTGGGCGACCGAATACAAATC	GATGGCATTGGACTGGGCTA	90	250	oxygen species	3
						Cellular response to	
						stress, heat, protein	
HSP70	KJ461738.1	AAGAAGGTTGGTGGTGAACG	TCCCAAATGTGTGTCTCCTG	121	250	refolding	4
						Cellular metal ion	
						homeostasis, cellular	
Metallothionein						response to oxidative	
(isoform 1)	EF185126	CGGCTATCTCCCGCGA	AGCTTTTACCAGAACCAAACAGT	210	200	stress	1
Mn Superoxide						Response to	
dismutase	EF446611	CTGTAGGTAGTAAGCGTGCTCC	GGATCATTTGACGAGATGAAGGC	200	200	oxidative stress	1
Ribosomal protein						Translation (Used as	
S3	AJ517687	CAGTGTGAGTCCCTGAGATACAAG	AAACGGTTGATAGGACGCAAG	154	250	housekeeping gene)	5
Thioredoxin						Cell redox	
reductase	KF275126	AATGCGGTGGCTTCCTACA	TCTGCTGGTCAAATCCTCGC	89	200	homeostasis	3

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⁵Navarro A, Faria M, Barata C, Piña B. 2011. Transcriptional response of stress genes to metal exposure in zebra mussel larvae and

adults. Environ Pollut. 159, 100-7.

891 Table S3. Factors, levels and coding values used for regression modelling in the two experiments.

Tissue was not considered as a factor in experiment 1 because data for gills and digestive glands were treated separately, while only one exposure duration (96 h) was used in experiment 2 (see text, section 2.2, for details). In the column 'Factor', Cr3 and Cr6 are the chosen variable names for the corresponding Cr redox species.

	Factor	Level	Coding
	Exposure duration (D)	24h 96h	-1 1
Experiment 1	Chromium redox form (Cr)	Cr3 Cr6	-1 1
	Exposure concentration (Co)	0 μg L ⁻¹ 5 μg L ⁻¹ 50 μg L ⁻¹ 500 μg L ⁻¹	0 5 50 500
	Tissue (T)	Gills Digestive glands	-1 1
Experiment 2	Cr3	0 μg L ⁻¹ 50 μg L ⁻¹	-1 1
	Cr6	0 μg L ⁻¹ 50 μg L ⁻¹	-1 1

910 Table S4. Total and total filterable ($< 0.45 \,\mu$ m) Cr concentrations in aliquots of Cristaline® water amended with different levels of either Cr(III) 911 or Cr(VI) at different incubation times (in hours).

Experiment 1. All values are in μ g L⁻¹. Concentrations at t=0, 2, 6 and 24 hours are arithmetic means \pm one standard deviations of triplicate measurements (duplicate measurements for t = 2). Concentrations at t = 0.5, 1 and 4 hours were measured only in one replicate. Bdl, below detection limit.

	Expected concentrations	Measured concentrations						
		t = 0	t = 0.5	t = 1	t = 2	t = 4	t = 6	t = 24
Cr(III) control unfiltered	0	0.19 ± 0.12	0.07	bdl	bdl	bdl	0.19 ± 0.27	1.0 ± 1.4
Cr(III) control <0.45 µm	0	0.07 ± 0.11	bdl	bdl	0.35 ± 0.49	bdl	0.17 ± 0.28	0.97 ± 1.27
Cr(VI) control unfiltered	0	0.07 ± 0.06	bdl	bdl	bdl	bdl	0.08 ± 0.10	0.42 ± 0.40
Cr(VI) control <0.45 µm	0	0.10 ± 0.17	bdl	bdl	bdl	bdl	0.07 ± 0.06	0.24 ± 0.33
Cr(III) unfiltered	5	4.59 ± 0.27	3.85	3.03	2.26 ± 0.23	1.41	1.64 ± 0.46	1.18 ± 0.29
Cr(III) filtered 0.45 µm	5	4.44 ± 0.41	3.73	4.09	2.33 ± 0.18	1.37	1.54 ± 0.45	0.68 ± 0.46
Cr(VI) unfiltered	5	4.32 ± 0.36	3.99	3.93	4.22 ± 0.67	3.97	4.32 ± 0.48	3.64 ± 0.44
Cr(VI) filtered 0.45 µm	5	4.93 ± 0.35	4.24	4.32	4.48 ± 0.44	4.24	4.50 ± 0.42	4.13 ± 0.40
Cr(III) unfiltered	50	31.3 ± 4.04	26	24	22.0 ± 2.8	15	17.3 ± 2.9	8.33 ± 2.08
Cr(III) filtered 0.45 µm	50	30.7 ± 1.53	29	24	22.0 ± 1.4	13	14.7 ± 2.5	6.00 ± 1.00
Cr(VI) unfiltered	50	36.0 ± 5.00	31	32	33.5 ± 4.9	30	32.7 ± 3.2	29.3 ± 3.1
Cr(VI) filtered 0.45 µm	50	34.3 ± 2.52	31	33	31.5 ± 4.9	31	33.7 ± 3.2	30.3 ± 2.5
Cr(III) unfiltered	500	434 ± 86.0	309	281	228 ± 2	136	124 ± 24	246 ± 210
Cr(III) filtered 0.45 µm	500	362 ± 25.4	304	274	148 ± 16	83	55.7 ± 9.7	18 ± 5.6
Cr(VI) unfiltered	500	363 ± 26.8	354	351	363 ± 30	327	361 ± 26	312 ± 32
Cr(VI) filtered 0.45 µm	500	378 ± 20.8	361	351	364 ± 28	336	360 ± 28	318 ± 28

Experiment 2. NIA, solutions with natural Cr isotopic abundances; IES, isotopically enriched solutions. Concentration in control exposures are given for ⁵²Cr only considering that only the concentration of this isotope was above the ICP-MS detection limit (see main text, section 3.1 for details). All values are arithmetic means ± 1 standard deviations, n = 3, t is in hours.

	Expected concentration	Measured c	oncentration
	1	t = 0	t = 24
⁵² Cr control unfiltered	0	0.84 ± 1.29	0.91 ± 1.33
⁵² Cr control filtered 0.45 μm	0	0.43 ± 0.43	0.37 ± 0.49
⁵² Cr(III) unfiltered – NIA	50	56.4 ± 2.98	16.1 ± 3.59
52 Cr(III) filtered 0.45 μ m – NIA	50	56.2 ± 0.93	7.35 ± 0.81
⁵² Cr(VI) unfiltered – NIA	50	56.0 ± 3.89	52.6 ± 1.81
52 Cr(VI) filtered 0.45 μ m – NIA	50	58.5 ± 1.76	52.7 ± 2.81
⁵⁰ Cr(III) unfiltered – IES	100	108 ± 6.2	26.7 ± 4.92
50 Cr(III) filtered 0.45 μ m – IES	100	104 ± 1.9	9.89 ± 1.98
⁵³ Cr(VI) unfiltered – IES	50	45.0 ± 1.86	42.0 ± 0.32
53 Cr(VI) filtered 0.45 μ m – IES	50	45.0 ± 0.15	$42.0\pm~1.13$

Table S5. Chromium concentrations accumulated in the gills (G) and digestive glands (DG) of individual specimen of Corbicula fluminea

exposed for 96 hours to solutions amended with either Cr(III) or Cr(VI) salts having natural isotopic abundances (NIA) or to a solution simultaneously amended with a mixture of ⁵⁰Cr(III) and ⁵³Cr(VI) using isotopically enriched solutions (IES). n.m., isotope not measured.

Cr species/isotopes	Exposure Concentration (µg L ⁻¹)	Tissue	⁵² Cr (µg g ⁻¹)	⁵⁰ Cr(III) (µg g ⁻¹)	⁵³ Cr(VI) (µg g ⁻¹)
			2.034	3.192	2.611
		Cilla	4.381	5.069	4.615
		GIIIS	2.332	3.105	2.727
	50		3.161	3.469	3.222
CI(III) = INIA	50		4.998	5.097	4.257
		DG	6.191	7.455	6.398
			13.250	13.984	11.140
			16.293	19.656	16.797
		Gille	13.058	12.445	11.856
		UIIIS	12.783	12.783 11.099	
Cr(VI) –NIA	50		2.490	2.318	2.234
		DG	3.812	3.533	3.399
			1.613	1.656	1.508
			n.m.	2.528	8.038
		Cilla	n.m.	2.923	7.692
		UIIIS	n.m.	1.891	5.582
${}^{50}Cr(III) + {}^{53}Cr(VI)$ IFS	100 + 50		n.m.	2.616	3.911
CI(III) + CI(VI) - IES	100 + 50		n.m.	0.969	1.134
		DG	n.m.	0.232	1.437
		DO	n.m.	1.394	2.002
			n.m.	0.354	0.539

Table S6. Comparison of Cr bioaccumulation ($\mu g g^{-1}$ dry weight) in bivalves under different exposure conditions from published literature studies. \$, exposure temperature = 12°C ; #, exposure temperature = 17°C ; & = $\mu g g^{-1}$ wet weight ; BCF = Bioconcentration factor.

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Exposure conditions							Bioaccumulation			
Organism	Acclimation (d)	Cr form	Cr (µg L ⁻¹)	Duration (d)	Test medium	Tissue	Cr (µg g ⁻¹)	BCF (L g ⁻¹)		
			0.001			DC	1.5	1500		
		Cr(VI)	0.01	— 7	Filtered seawater	DG	2.5	250		
Mytilus galloprovincialis ¹	3		0.05			Soft tissue (no DG)	2.1	54		
			0.001				1	1000		
			0.01				1.2	120		
			0.05				2.2	44		
			5	4\$			0.5 ± 0.2	0.1		
Dreissena			15		Cristaline®	Whole organism (no shell)	1.9 ± 0.7	0.13		
polymorpha ²			5	4#			0.7 ± 0.1	0.14		
	4		15	4"			1.3 ± 0.7	0.09		
	4	Cr(VI)	5	4\$	mineral		1 ± 0.2	0.2		
Dreissena r.			15	4°	water		0.9 ± 0.4	0.06		
bugensis ²			5	.#	-	-	1.2 ± 0.3	0.24		
0			15	4#			1.2 ± 0.3	0.08		
			0.5			Whole	1.4 ± 0.2	2.8		
Mytilus	7	Cr(VI)	5	28	Artificial seawater	organism	2.5 ± 0.4	0.5		
galloprovincialis	,		50			(no shell)	15.4 ± 6.1	0.31		

Table S6 (continu	ied)							
Organism	Acclimation (d)	Cr form	Cr (µg L ⁻¹)	Duration (d)	Test medium	Tissue	Cr (µg g ⁻¹)	BCF (L g^{-1})
				1			3400*	3400
				2		X 71 1 -	8400	4200
Caelatura	15	Unspecified	20000	5	Uneposified	whole	16800	3360
companyoi ⁴	15	Unspecified	20000	10	Unspecified	(no shell)	28200	2820
				15		(no snen)	42400	2827
				20			65000	3250
			88			G	27.47 ± 18.65	0.31
Crassostrea	23	unspecified –	00	- 4	Artificial seawater	DG	13.89 ± 5.69	0.16
virginica ⁵			144			G	19.41 ± 2.87	0.13
						DG	28.19 ± 12.85	0.2
Crassostrea		Cr(VI) –	5			Whole organism (no shell)	3.12 ± 0.45	0.62
virginica ⁶	1 month		10	-12 weeks	Artificial		5.63 ± 1.15	0.56
Mytilus adulis ⁶	1 monui		5	- 12 weeks	seawater		4.83 ± 1.32	0.97
Mynnas eauns			10			(no shen)	9.41 ± 3.37	0.941
				1			8.24 - 22.59	$2.3 - 21^{A}$
			0.70 3.58	5		G	17.51 - 31.44	$4.9 - 45^{A}$
			0.70 - 5.58	15		0	29.83 - 34.32	$8.3 - 49^{A}$
Corbicula	0	Cr total –		30	In situ		22.75	$6.3 - 32^{A}$
fluminea ⁷	U			1	(river)		4.05 – 9. 05	$1.1 - 13^{A}$
			0.70 - 3.58	5		DG	3.28 - 14.81	$0.9 - 21^{A}$
				15		DG	4.80 - 17.31	$1.3 - 25^{A}$
				30			3.51 - 18.22	$1.0 - 26^{A}$

945 A, the BCF range covers the ratios (minimum accumulated concentration / maximum exposure concentration) and (maximum accumulated
 946 concentration / minimum exposure concentration).

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- 962 Landfill Leachate Contamination. Bulletin of Environmental Contamination and Toxicology. 2014;93(2):149-53.
- 963

Table S7. Arithmetic means ± 1 standard deviation (n = 4; n = 3 for \$) for the values of cellular biomarkers measured in gills and digestive glands of the freshwater bivalve *Corbicula fluminea* exposed to 0, 5, 50 and 500 µg L⁻¹ of Cr(III) or Cr(VI) during 24 or 96 h (exp #1). w.w., wet weight.

			Gills				Digestiv	e glands		
Biomarker	Exposure (Type)	Exposure (Duration)	0 μg L ⁻¹	5 μg L ⁻¹	50 μg L ⁻¹	500 μg L ⁻¹	0 μg L ⁻¹	5 μg L ⁻¹	50 μg L ⁻¹	500 µg L ⁻¹
	Cr(III)	24 h	19.1 ± 3.5	18.7 ± 3.5	17.3 ± 4.9	18 ± 4.6	38.7 ± 1.5	36.0 ± 6.4	37.1 ± 6.3	41.4 ± 4.8
Protein content 1	Cr(VI)		16.1 ± 4.8	18.9 ± 2.8	18.7 ± 1.3	18.1 ± 1	35.9 ± 7.2	44.1 ± 7	37.9 ± 4.8	31.0 ± 5.3
(nomogenate) mg/g w.w.	Cr(III)	96 h	20.4 ± 1.3	18.1 ± 1.2	16.9 ± 4.5	13.4 ± 2.4	39.8 ± 2.5	37.2 ± 5.2	34.8 ± 7.6	33.3 ± 10
0,0	Cr(VI)		14.9 ± 2.0	15.7 ± 1.1	16.6 ± 1.2	16.2 ± 2.6	36.8 ± 5.8	Digestive glands 5 μ g L ⁻¹ 50 μ g L ⁻¹ 5 36.0 ± 6.4 37.1 ± 6.3 4 2 44.1 ± 7 37.9 ± 4.8 3 5 37.2 ± 5.2 34.8 ± 7.6 3 8 34.8 ± 4.2 34.7 ± 5.8 3 5 25.5 ± 3.5 26.9 ± 4.7 3 3 31.5 ± 5.9 25.7 ± 2.6 2 4 26.3 ± 2.8 25.4 ± 6.4 2 2 24.7 ± 2.4 24.3 ± 4 2 2 24.7 ± 2.4 24.3 ± 4 2 2 24.7 ± 0.2 1.5 ± 0.1 1.5 ± 0.1 4 1.6 ± 0.2 1.5 ± 0.1 1.5 ± 0.2 5 5.2 ± 1.8 4.1 ± 0.9 4 4 1.6 ± 0.2 1.5 ± 0.2 1.5 ± 0.2 5 5.2 ± 1.2 4.8 ± 0.9 4 4.8 ± 0.7 4.7 ± 0.7 4 4.8 ± 0.7 4.7 ± 0.7 4 1.1 ± 0.1 1.3 ± 0.2 1.5 ± 0.2 4.	39.8 ± 3.5	
	Cr(III)	241	9.7 ± 1.8	11.9 ± 0.8	12.3 ± 1.4	11.8 ± 0.3	27.8 ± 2.5	25.5 ± 3.5	26.9 ± 4.7	30.4 ± 1.9
Protein content 2	Cr(VI)	24 n	12.6 ± 1.8	12.9 ± 1.4	12.9 ± 1.2	12.6 ± 1.1	28.2 ± 5.8	31.5 ± 5.9	25.7 ± 2.6	23.6 ± 5.1
(cytosolic fraction) mg/g w.w.	Cr(III)	0.6 1	15.1 ± 0.6	12.4 ± 1	10.7 ± 2.3	8.7 ± 1.4	29.3 ± 2.4	26.3 ± 2.8	25.4 ± 6.4	24.9 ± 6.8
0,0	Cr(VI)	96 N	10 ± 0.7	10.9 ± 0.8	11.1 ± 0.8	11.3 ± 1.6	25.6 ± 2.2	24.7 ± 2.4	24.3 ± 4	25.8 ± 2.3
	Cr(III)		1.0 ± 0.3	0.7 ± 0.2	0.6 ± 0.2	0.6 ± 0.1	1.5 ± 0.2	1.4 ± 0.3	1.4 ± 0.4	1.6 ± 0.3
Cholesterol	Cr(VI)	24 n	0.6 ± 0.2	0.7 ± 0.1	0.7 ± 0.1	0.7 ± 0.1	1.5 ± 0.3	1.7 ± 0.2	1.5 ± 0.1	1.5 ± 0.2
mg/g w.w.	Cr(III)		0.7 ± 0.1	0.7 ± 0.1	0.6 ± 0.2	0.5 ±0.1	1.6 ± 0.2	1.5 ± 0.3	1.5 ± 0.4	1.5 ± 0.3
	Cr(VI)	96 N	0.6 ± 0.1	0.6 ± 0.01	0.6 ± 0.01	0.6 ± 0.1	1.6 ± 0.3	1.6 ± 0.2	1.5 ± 0.2	1.6 ± 0.2
	Cr(III)	24.5	1.3 ± 0.4	1.2 ± 0.2	1.1 ± 0.4	1.2 ± 0.5	4.5 ± 0.6	5.2 ± 1.8	4.1 ± 0.9	4.7 ± 0.6
Triglycerides	Cr(VI)	24 N	1 ± 0.5	1.2 ± 0.2	1 ± 0.1	0.9 ± 0.2	3.8 ± 0.8	6.1 ± 0.8	6.1 ± 1	4.5 ± 1.1
mg/g w.w.	Cr(III)	oc h	1.2 ± 0.1	1.2 ± 0.1	1 ± 0.3	0.8 ± 0.2	5.5 ± 0.2	5.2 ± 1.2	4.8 ± 0.9	4.9 ± 1.5
	Cr(VI)	96 N	0.9 ± 0.1	0.9 ± 0.1	0.8 ± 0.1	0.9 ± 0.2	5.0 ± 1.1	4.8 ± 0.7	4.7 ± 0.7	4.0 ± 0.4
	Cr(III)	24 h	2.3 ± 0.1	2.4 ± 0.3	2.3 ± 0.3	2.3 ± 0.2	1.1 ± 0.2	1.1 ± 0.1	1.3 ± 0.2	1.4 ± 0.2
ETS	Cr(VI)	24 N	2.4 ± 0.1	2.2 ± 0.1	2.3 ± 0.1	2.2 ± 0.1	1.8 ± 0.4	1.5 ± 0.2	1.7 ± 0.2	1.8 ± 0.4
mmol O2/g prot/h	Cr(III)	06 h	2.1 ± 0.1	1.9 ± 0.2	1.8 ± 0.4	1.4 ± 0.3	1.6 ± 0.1	1.8 ± 0.2	3.6 ± 1.4	2.2 ± 0.4
	Cr(VI)	90 11	1.4 ± 0.1	2.3 ± 0.1	2.2 ± 0.2	2.2 ± 0.3	1.8 ± 0.3	1.7 ± 0.2	1.9 ± 0.3	1.9 ± 0.3

			Gills				Digestive glands				
Biomarker	Exposure (Type)	Exposure (Duration)	0 μg L ⁻¹	5 μg L ⁻¹	50 μg L ⁻¹	500 μg L ⁻¹	0 μg L ⁻¹	5 µg L ⁻¹	50 μg L ⁻¹	500 μg L ⁻¹	
	Cr(III)	24 h	41.3 ± 2.5	37.7 ± 3.3	33.4 ± 4.5	30.4 ± 4.9	27.0 ± 2.5	30.2 ± 2.0	24.9 ± 4.4	22.9 ± 1.1	
LOOH	Cr(VI)	24 N	34.4 ± 3.3	35.5 ± 2.5	37.1 ± 4.4	36.3 ± 2.9	24.9 ± 3.0	20.9 ± 1.5	23.8 ± 2.1	24.3 ± 4.8	
µmol/g prot	Cr(III)	0C h	33.1 ± 2.1	38.7 ± 2.0	42.0 ± 10.6	37.0 ± 3.1	21.0 ± 1.8	21.4 ± 1.6	24.0 ± 2.3	27.2 ± 5.7	
	Cr(VI)	96 N	31.2 ± 2.9	40.7 ± 3.6	38.0 ± 2.8	39.4 ± 2.2	26.5 ± 2.7	Digestive glands5 μ g L ⁻¹ 50 μ g L ⁻¹ 500 30.2 ± 2.0 24.9 ± 4.4 22.9 20.9 ± 1.5 23.8 ± 2.1 24.3 21.4 ± 1.6 24.0 ± 2.3 27.2 24.8 ± 3.6 26.7 ± 2.1 21.0 89 ± 28 120 ± 41 99 124 ± 15 132 ± 4 125 111 ± 33 104 ± 38 113 87 ± 16 122 ± 37 88 116 ± 30 157 ± 78 121 133 ± 32 107 ± 48 175 94 ± 73 130 ± 77 93 138 ± 50 156 ± 79 112 887 ± 208 955 ± 611 583 632 ± 144 746.7 ± 43 1068 643 ± 205 612 ± 260 605 701 ± 173 841 ± 337 695 6.3 ± 4.2 7.6 ± 7.5 7.1 10 ± 5.2 8.5 ± 6.2 19 6.8 ± 7.7 4.8 ± 7.1 1.7 2.6 ± 2.6 5.8 ± 6.4 1.7 89 ± 25 115 ± 39^5 105 82 ± 16 78 ± 43 99 52 ± 22 50 ± 27^5 62 67 ± 37 67 ± 21 73	21.0 ± 1.0		
	Cr(III)	24 h	107 ± 17	116 ± 9	107 ± 16	108 ± 10	81 ± 37	89 ± 28	120 ± 41	99 ± 27	
ACP μmol/g prot/h	Cr(VI)	24 N	117 ± 7	113 ± 8	113 ± 8	104 ± 6	132 ± 47	124 ± 15	132 ± 4	125 ± 23	
	Cr(III)	06 h	115 ± 5	106 ± 7	106 ± 7	92 ± 9	103 ± 31	111 ± 33	104 ± 38	113 ± 29	
	Cr(VI)	90 11	95 ± 7	102 ± 8	105 ± 1	95 ± 6	84 ± 24	87 ± 16	122 ± 37	88 ± 17	
LDH μM NADPH/g prot/h	Cr(III)	24 h	75 ± 13	100 ± 12	106 ± 32	108 ± 33	129 ± 27	116 ± 30	157 ± 78	121 ± 50	
	Cr(VI)	24 N	114 ± 13	108 ± 17	98 ± 18	103 ± 18	145 ± 39	133 ± 32	107 ± 48	175 ± 40	
	Cr(III)	0C h	96 ± 10	90 ± 12	111 ± 58	183 ± 32	129 ± 71	94 ± 73	130 ± 77	93 ± 34	
	Cr(VI)	96 N	179 ± 32	85 ± 3	75 ± 16	120 ± 16	90 ± 27	129 ± 71 94 ± 73 130 ± 77 93 90 ± 27 138 ± 50 156 ± 79 112 708 ± 190 887 ± 208 955 ± 611 583	112 ± 35		
	Cr(III)	24 h	391 ± 46	397 ± 67	384 ± 170	367 ± 114	708 ± 190	887 ± 208	955 ± 611	583 ± 146	
GST	Cr(VI)	24 N	440 ± 44	366 ± 25	301 ± 47	303 ± 35	710 ± 221	632 ± 144	746.7 ± 43	1068 ± 205	
g prot/min	Cr(III)	0C h	249 ± 11	277 ± 75	387 ± 106	329 ± 65	607 ± 173	ug L-15 μ g L-15 (c) 1 ± 2.5 30.2 ± 2.0 24 1 ± 3.0 20.9 ± 1.5 23 0 ± 3.0 20.9 ± 1.5 23 0 ± 1.8 21.4 ± 1.6 24 5 ± 2.7 24.8 ± 3.6 26 ± 37 89 ± 28 12 2 ± 47 124 ± 15 1 3 ± 31 111 ± 33 10 ± 24 87 ± 16 12 2 ± 27 116 ± 30 19 5 ± 39 133 ± 32 10 2 ± 77 94 ± 73 13 ± 27 138 ± 50 19 ± 190 887 ± 208 95 ± 221 632 ± 144 74 ± 173 643 ± 205 61 5 ± 275 701 ± 173 84 ± 3.3 6.3 ± 4.2 7. ± 1.6 10 ± 5.2 8. ± 7.9 6.8 ± 7.7 4. ± 2.7 2.6 ± 2.6 5. ± 15 89 ± 25 11 ± 46 82 ± 16 7 ± 22 52 ± 22 50 ± 33 67 ± 37 6	612 ± 260	605 ± 80	
	Cr(VI)	90 11	348 ± 44	597 ± 87	443 ± 57	441 ± 165	558 ± 275	$5 \ \mu g \ L^{-1}$ $50 \ \mu g \ L^{-1}$ 30.2 ± 2.0 24.9 ± 4.4 20.9 ± 1.5 23.8 ± 2.1 21.4 ± 1.6 24.0 ± 2.3 24.8 ± 3.6 26.7 ± 2.1 89 ± 28 120 ± 41 124 ± 15 132 ± 4 111 ± 33 104 ± 38 87 ± 16 122 ± 37 116 ± 30 157 ± 78 133 ± 32 107 ± 48 94 ± 73 130 ± 77 138 ± 50 156 ± 79 887 ± 208 955 ± 611 632 ± 144 746.7 ± 43 643 ± 205 612 ± 260 701 ± 173 841 ± 337 6.3 ± 4.2 7.6 ± 7.5 10 ± 5.2 8.5 ± 6.2 6.8 ± 7.7 4.8 ± 7.1 2.6 ± 2.6 5.8 ± 6.4 89 ± 25 $115 \pm 39^{\$}$ 82 ± 16 78 ± 43 52 ± 22 $50 \pm 27^{\$}$ 67 ± 37 67 ± 21	695 ± 174		
	Cr(III)	24 h	6.1 ± 2.0	10.2 ± 5	8.0 ± 5.5	18.7 ± 6.5	4.9 ± 3.3	6.3 ± 4.2	7.6 ± 7.5	7.1 ± 5.7	
CSP3	Cr(VI)	24 N	4.0 ± 0.9	6.3 ± 0.5	5.9 ± 1.4	10.5 ± 5.2	2.9 ± 1.6	10 ± 5.2	8.5 ± 6.2	19 ± 8.4	
µmol pNA/g prot/h	Cr(III)	06 h	7.2 ± 1.7	6.3 ± 2.7	4.1 ± 0.8	3.0 ± 1.2	9.4 ± 7.9	6.8 ± 7.7	4.8 ± 7.1	1.7 ± 1.3	
	Cr(VI)	90 11	3.1 ± 0.6	4.7 ± 1.1	5.3 ± 1.8	9.5 ± 3.5	1.2 ± 2.7	2.6 ± 2.6	5.8 ± 6.4	1.2 ± 1	
	Cr(III)	24 h	51 ± 5.8	47 ± 2.8	42 ± 7.7	43 ± 5.5	72 ± 15	89 ± 25	115 ± 39 ^{\$}	105 ± 63	
Catalase (CAT)	Cr(VI)	24 n	41 ± 1.3	40 ± 6.9	37 ± 4.2	39 ± 3.5	61 ± 46	82 ± 16	78 ± 43	99 ± 39	
prot/min	Cr(III)	0C h	41 ± 4.2	43 ± 2.7	42 ± 17	34 ± 11	63 ± 22	52 ± 22	50 ± 27 ^{\$}	62 ± 36	
• •	Cr(VI)	96 N	36 ± 13	49 ± 8.9	45 ± 4.3	46 ± 9.3	39 ± 33	67 ± 37	67 ± 21	73 ± 28	

			Gills				Digestive glands			
Biomarker	Exposure (Type)	Exposure (Duration)	0 μg L ⁻¹	5 µg L ⁻¹	50 μg L ⁻¹	500 μg L ⁻¹	0 μg L ⁻¹	5 µg L ⁻¹	50 μg L ⁻¹	500 µg L ⁻¹
	Cr(III)	24 h	-	-	-	-	25 ± 3.7	25 ± 3.9	29 ± 8.6	18 ± 3.0
TAC	Cr(VI)	24 n	-	-	-	-	23 ± 6.7	27 ± 3.9	27 ± 3.7	37 ± 5.3
µmolTeq/g prot/min	Cr(III)	96 h	-	-	-	-	27 ± 9.6	24 ± 9.7	22 ± 8.8	19 ± 5.0
	Cr(VI)		-	-	-	-	14 ± 5.6	22 ± 4.6	27 ± 9.7	20 ± 3.1
	Cr(III)	24 h	-	-	-	-	-	-	-	-
GPx	Cr(VI)	24 n	-	-	-	-	-	-	-	-
µmolTeq/g prot/min	Cr(III)	0C h	-	-	-	-	-	-	-	-
	Cr(VI)	90 N	-	-	-	-	-	-	-	-

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Biomarker	Exposure (type)	Gills	Digestive glands	Biomarker	Exposure (type)	Gills	Digestive glands
	Control	11.6 ± 3.8	25.6 ± 4		Control	89 ± 18	85 ± 15
Protein content (homogenate)	⁵⁰ Cr(III) + ⁵³ Cr(VI)	14.9 ± 2.3	28.9 ± 4.2	АСР	⁵⁰ Cr(III) + ⁵³ Cr(VI)	89 ± 17	76 ± 20
mg/g w.w.	Cr(III)	15.6 ± 2.2	29.6 ± 1.8	µmol/g prot/h	Cr(III)	78 ± 14	80 ± 13
	Cr(VI)	11.5 ± 2.5	32.5 ± 6		Cr(VI)	77 ± 18	83 ± 22
	Control	8.6 ± 0.8	19.7 ± 2.7		Control	135 ± 35	182 ± 88
Protein content (cytosolic fraction)	⁵⁰ Cr(III) + ⁵³ Cr(VI)	10 ± 1.8	20.7 ± 2.4	LDH	⁵⁰ Cr(III) + ⁵³ Cr(VI)	107 ± 35	121 ± 35
mg/g w.w.	Cr(III)	9.9 ± 1.7	21.7 ± 0.7	µmol NADPH/g prot/h	Cr(III)	95 ± 29	137 ± 65
	Cr(VI)	9.6 ± 0.3	22 ± 3.8		Cr(VI)	110 ± 59	170 ± 121
	Control	0.48 ± 0.16	1.03 ± 0.25		Control	-	31.3 ± 6.5
Cholesterol	⁵⁰ Cr(III) + ⁵³ Cr(VI)	0.61 ± 0.1	1.23 ± 0.24	TAC	⁵⁰ Cr(III) + ⁵³ Cr(VI)	-	25.5 ± 3.9
mg/g w.w.	Cr(III)	0.61 ± 0.1	1.20 ± 0.17	µmol Teq/g prot/min	Cr(III)	-	28.9 ± 5.2
	Cr(VI)	0.48 ± 0.12	1.29 ± 0.34		Cr(VI)	-	53.9 ± 40.7
	Control	0.45 ± 0.24	3.59 ± 1.38		Control	343 ± 126	982 ± 345
Triglycerides	⁵⁰ Cr(III) + ⁵³ Cr(VI)	0.45 ± 0.1	4.22 ± 1.53	GST umol CDNB/	⁵⁰ Cr(III) + ⁵³ Cr(VI)	428 ± 178	1001 ± 333
mg/g w.w.	Cr(III)	0.58 ± 0.1	3.78 ± 0.83	g prot/min	Cr(III)	$415\ \pm 77$	972 ± 184
	Cr(VI)	0.43 ± 0.2	4.08 ± 1.28		Cr(VI)	461 ± 126	949 ± 173
	Control	2.8 ± 0.1	1.8 ± 0.1		Control	13 ± 5.9	7 ± 3.6
ETS	⁵⁰ Cr(III) + ⁵³ Cr(VI)	3.1 ± 0.1	1.8 ± 0.4	CSP3	⁵⁰ Cr(III) + ⁵³ Cr(VI)	13.7 ± 1.6	3.8 ± 2.6
mmol O ₂ /g prot/h	Cr(III)	2.8 ± 0.2	1.7 ± 0.2	µmol pNA/g prot/h	Cr(III)	19.8 ± 5.8	5.5 ± 3.6
	Cr(VI)	2.9 ± 0.3	1.6 ± 0.3		Cr(VI)	4.3 ± 1.7	5.8 ± 4.7
	Control	64.3 ± 9.7	22.4 ± 0.9				
LOOH	⁵⁰ Cr(III) + ⁵³ Cr(VI)	55.9 ± 5.8	23.2 ± 1.3				
µmol/g prot	Cr(III)	61.9 ± 3	22.6 ± 1.4				
	Cr(VI)	57.4 ± 1.8	24.6 ± 3.1				

982Table S8. Activities of cellular biomarkers (arithmetic means ± 1 standard deviation; n =3) measured in gills and digestive glands of the983freshwater bivalve *Corbicula fluminea* in exp# 2. CAT and GPx could not be quantified and were excluded from the table. w.w., wet weight.

Table S9. Regression analysis of the biomarkers' expression (after Log_{10} transformation) as a function of the different explanatory variables: exposure duration (D), Cr redox form (Cr), exposure concentration (Co), tissue (T) and their interactions (*). A probability threshold of 0.01 was used for the construction of all fitted models. r²: coefficient of determination.

		Experiment 1	Experiment 2				
Biomarker	Tissue	Fitted model	F-test		Fitted model	F-test	
		Log10 (biomarker)	P-value	r^2	Log10 (biomarker)	P-value	r^2
Chol	G		0.209		0.104 ± 0.171 T	< 10 ⁻⁴	0 772
Choi	DG		0.881		- 0.104 + 0.171 1	< 10	0.772
I DH	G		0.010			0.866	
LDII	DG		0.559			0.800	
TG	G		0.107		0.118 ± 0.459 T	< 10-4	0.927
10	DG		0.510		0.110 + 0.459 1	< 10	0.927
Prot1	G		0.059		1.287 ± 0.173 T	< 10-4	0.800
11011	DG	1.563 + 1.096 · 10 ⁻⁴ D*Cr*Co	0.003	0.129	1.207 + 0.175 1	< 10	0.000
	G	1.061 - 0.035 D*Cr - 1.008 · 10 ⁻⁴ D*Co +	< 10 ⁻⁴	0.302			
Prot2		+ 1.195 · 10 ⁻⁴ D*Cr*Co			1.148 + 0.173 T	< 10 ⁻⁴	0.924
	DG		0.180				
GST	G	2.558 + 0.049 D*Cr	0.001	0.161	2.786 ± 0.191 T	< 10 ⁻⁴	0 703
001	DG		0.055		2.700 + 0.171 1	< 10	0.705
LOOH	G		0.114		1 570 - 0 205 T	< 10 ⁻⁴	0.968
LOON	DG	1.384 + 0.038 D*Cr - 1.416 · 10 ⁻⁴ D*Cr*Co	< 10 ⁻⁴	0.320	1.570 0.205 1	< 10	0.900
ACP	G	2.036 - 0.018 D - 0.753 · 10 ⁻⁴ Co	< 10 ⁻⁴	0.290		0.962	
nei	DG	2.009 - 0.046 D*Cr	0.006	0.118		0.902	
	G	$0.725 - 0.069$ Cr + $3.724 \cdot 10^{-4}$ Co - $4.168 \cdot 10^{-4}$ D*Co +	< 10 ⁻⁴	0.460			
CSP3		$+ 3.911 \cdot 10^{-4} \text{ D*Cr*Co}$	-		0.848 - 0.180T	0.005	0.310
	DG	0.550 - 9.121·10 ⁻⁴ D*Co	< 10 ⁻³	0.175			
	G	$0.316 - 0.042 D + 0.974 \cdot 10^{-4} Cr^*Co + 1.094 \cdot 10^{-4}$	< 10 ⁻⁴	0.439			
ETS		D*Cr*Co			0.347 - 0.115 T	< 10 ⁻⁴	0.844
	DG	0.224 + 0.072 D - 0.051 D*Cr	< 10 ⁻⁴	0.440			

991 Table S10. Regression analysis of the mRNA expression (after Log_{10} transformation) as a 992 function of the Cr redox form (Cr) and the exposure concentration (Co) after 96 h of exposure 993 (experiment 1). A probability threshold of 0.01 was used for the construction of all fitted models. 994 r^2 : coefficient of determination. G, gills; DG, digestive glands.

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mRNA	Tissue	Fitted model	F-test	
		Log ₁₀ (mRNA)	P-value	\mathbf{r}^2
GST pi	G	0.207 - 2.671 10 ⁻⁴ Cr*Co ^(a)	0.012	0.255
	DG		0.706	
GPv	G	Not performed ^(b)		
UI X	DG		0.405	
САТ	G	-1.482 - 6.294 10 ⁻⁴ Cr*Co	< 10 ⁻³	0.475
	DG		0.183	
CuZn	G	-0.516 - 10.148 10 ⁻⁴ Cr*Co	< 0.01	0.276
SOD	DG		0.090	
Mn SOD	G	-2.146 - 4.477 10 ⁻⁴ Cr*Co	< 10 ⁻³	0.472
MII SOD	DG		0.201	
тр	G		0.116	
IK	DG	-1.407 + 0.074 Cr + 3.966 10 ⁻⁴ Co	0.002	0.439
MT	G	0.423 - 3.556 10 ⁻⁴ Cr*Co	0.016	0.237
171 1	DG	0.801 - 2.862 10 ⁻⁴ Co	0.011	0.259
USD70	G	0.458 - 4.447 10 ⁻⁴ Cr*Co	< 0.01	0.268
HSP/0	DG	0.120 + 0.091 Cr + 5.077 10 ⁻⁴ Co	<10-4	0.668

(a) See also table S2 for the coding of Cr redox forms. The coding for Cr(III) is '-1', meaning that mRNA expression increases with concentration because the minus sign of the coefficient for 'Cr*Co' and the minus sign for Cr(III) factor give a + sign The opposite situation occurs for Cr(VI).

999 (b) Insufficient GPx level.

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