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1 **The diagenetic history of the giant Lacq**
2 **gas field, witness to the Apto-Albian**
3 **rifting and the Pyrenean orogeny,**
4 **revealed by fluid and basin modeling**
5

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11 Highlights

- 12 • Combining several equations of state allows the P-T modeling of H₂S-bearing fluids.
- 13 • The rifting and the orogenic stages influenced the diagenesis of the Deep Lacq reservoir.
- 14 • Hydrocarbon maturation, dolomitization and calcite precipitation were controlled by the
15 rifting event.
- 16 • The reservoir was infilled with gas during the Pyrenean orogeny
- 17 • TSR did not occur *in-situ*.

18
19

20 Abstract

21 The link between fluid circulation schemes and basin histories remain one of the most valuable
22 tools to understand the diagenetic evolution of petroleum reservoir. This study proposes a
23 diagenetic model for the Deep Lacq reservoir constrained by petrography, geochemistry, fluid
24 inclusions studies and basin modeling analyses. Drill cores penetrating the 3000 meters deep Late
25 Jurassic to Early Cretaceous “Deep Lacq” reservoirs, in the Aquitaine basin SW France, were used
26 to investigate how the geodynamic evolution of the basin influenced the diagenesis of petroleum
27 reservoirs. This basin experienced a phase of rifting during the Lower Cretaceous followed by
28 inversion and compression during the Paleogene. Petrographic and geochemical results indicate
29 that early diagenesis involved bacterial activity and early dolomitization. Fluid thermodynamic
30 modeling, coupled with the reconstruction of the basin history provided insight into the timing of
31 diagenetic fluid circulations. Burial diagenesis involved a rift-related dolomitization episode linked
32 mainly to the recrystallization of earlier dolomites and the circulation of deep hot fluids connected
33 to the Triassic evaporites. The dolomitizing fluids circulated at temperatures close to 136-144°C
34 and pressures of 338-386 bars, during the Aptian (120-116Ma). During the post-rift uplift and its
35 subsequent thermal re-equilibration, multiple fluid pulses precipitated coarse blocky calcites. These
36 Ca-rich fluids circulated between 109 and 97 Ma, at high-pressure conditions at around of 704-766
37 bars and with temperatures of 146-156°C. The circulation of another fluid pulse that resulted in the
38 precipitation of anhydrite cements is linked to the onset of the Pyrenean orogeny. No evidence that
39 support *in-situ* TSR reactions were found, which suggest migration of TSR-related H₂S from a
40 kitchen in the deeper parts of the basin. This multi-modeling approach show how the detailed

41 thermodynamic analyses of fluid inclusions can add valuable constraints on a diagenetic model
42 and eventually link it to the larger-scale basin's history.

43 Keywords: Lacq petroleum system, Aquitaine basin, fluid inclusion modeling, basin modeling,
44 sulfur-rich fluid system, TSR.

45 1 Introduction

46 The circulation of diagenetic fluids in petroleum reservoirs can be strongly influenced by the
47 geologic history of the basin (Beaudoin et al., 2014, 2015; Salardon et al., 2017; Renard et al.,
48 2019; Elias Bahnan et al., 2020). In the Aquitaine foreland basin, north of the Pyrenees Mountains
49 (SW France), several hydrocarbon fields are located in Jurassic and Cretaceous reservoirs (Biteau
50 et al., 2006). The giant Lacq field is part of the Lacq petroleum system with productive reservoirs
51 extending over 320 km², a gas column reaching 3100 m and reserves in the order of 8.9 trillion
52 cubic feet (Biteau et al., 2006). The exploitation of this gas field was a pioneering achievement at
53 the time of its discovery in 1951. The high pressure (675 bar) and temperature (135°C) with a
54 dangerous gas composition (15,2% H₂S, 9,7% CO₂; 69,2% C₁, 5,3% C₂₊) (Le-vot et al., 1996)
55 made it necessary to build an innovative desulfurization plant in 1957, still operational till present
56 day.

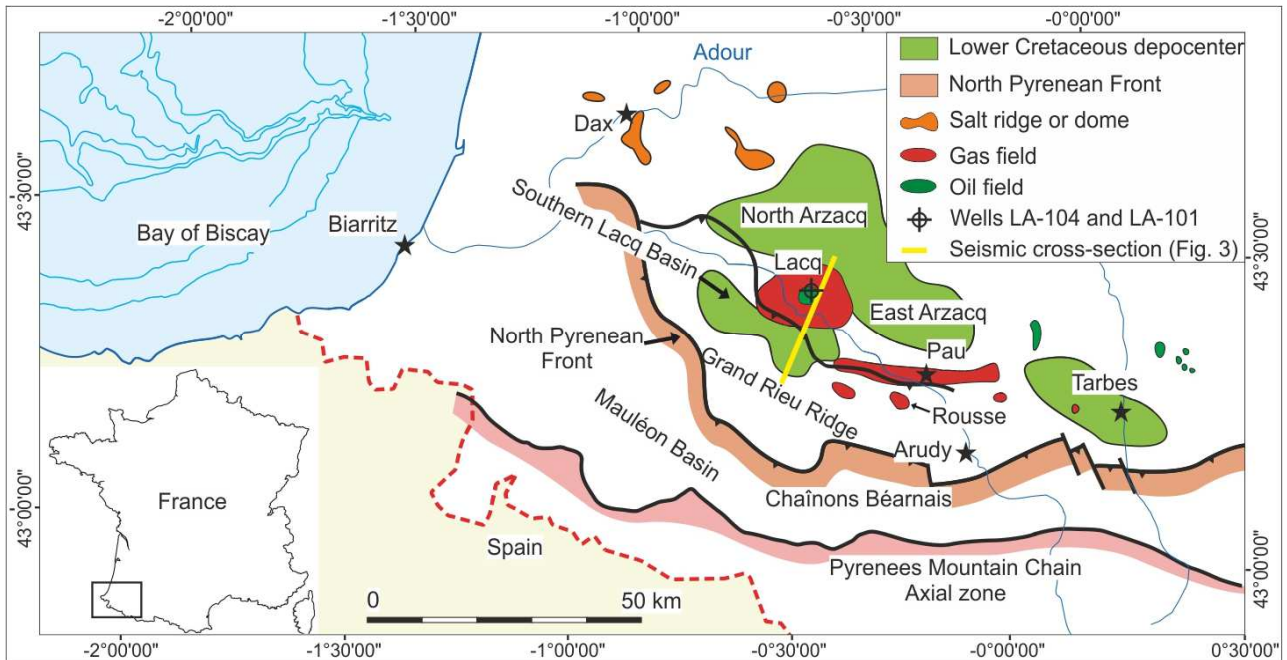
57 The anticlinal structure of Deep Lacq is the result of the superposition of Jurassic rifting and Early
58 Cretaceous rifting events, later inverted during the Late Cretaceous and the Paleogene (Sibuet et
59 al., 2004; Jammes et al., 2010, 2009; Clerc and Lagabrielle, 2014; Mouthereau et al., 2014; Corre
60 et al., 2016; Teixell et al., 2016; Vacherat et al., 2016). These events had significant control on the
61 types, sources, chemistries and circulation mechanisms of the diagenetic fluids in the Aquitaine
62 basin, as previously documented in the Rouse field (Renard et al., 2019), in the Upper Lacq oil
63 reservoir (Elias Bahnan et al., 2020) and in the Chaînons Béarnais (Incerpi et al., 2020; Motte et
64 al., 2021; Salardon et al., 2017). However, despite its importance as the most strategic reservoir in
65 France during the 1970s (Biteau et al., 2006), no models were yet proposed to explain the
66 diagenesis of Deep Lacq.

67 This study focuses on the Kimmeridgian-Barremian Deep Lacq gas field and aims to present a
68 diagenetic model describing the different types, sources and circulation schemes of diagenetic
69 fluids at different phases of the complex geological history that the Aquitaine basin has

70 experienced. The diagenetic events described in this research are placed in a geodynamic
71 perspective that allows a better understanding of fluid activity during a complete Wilson cycle.
72 Starting from a tectonic quiescence state and proceeding to early rifting, hyperextension, post-rift
73 subsidence and ending with orogenesis, fluid-rock interactions are examined at each step. The
74 case study presented in this work can be used as an example to compare diagenetic models of
75 petroleum reservoirs in basins that experienced a similar geodynamic history as the Aquitaine.
76 Examples include the Dengying reservoirs of the SW Sichuan basin (Wu et al., 2016), the Jurassic
77 carbonates of Mount Lebanon (Nader, 2003), the Mobile Bay Jurassic Norphlet Formation of the
78 Gulf of Mexico (Mankiewicz et al., 2009), the Níspero deeply buried Lower Cretaceous carbonate
79 reservoir in the Gulf of Mexico (Bourdet et al., 2010), the Devonian Southesk-Cairn carbonate
80 complex in Alberta, Canada (Machel and Buschkuehle, 2008), etc... The significance of this work
81 is the use of multi-scale modeling approaches to extrapolate μm -scale fluid inclusions to basin-
82 scale geodynamic events of the Aquitaine basin.

83 2 Geology and stratigraphy

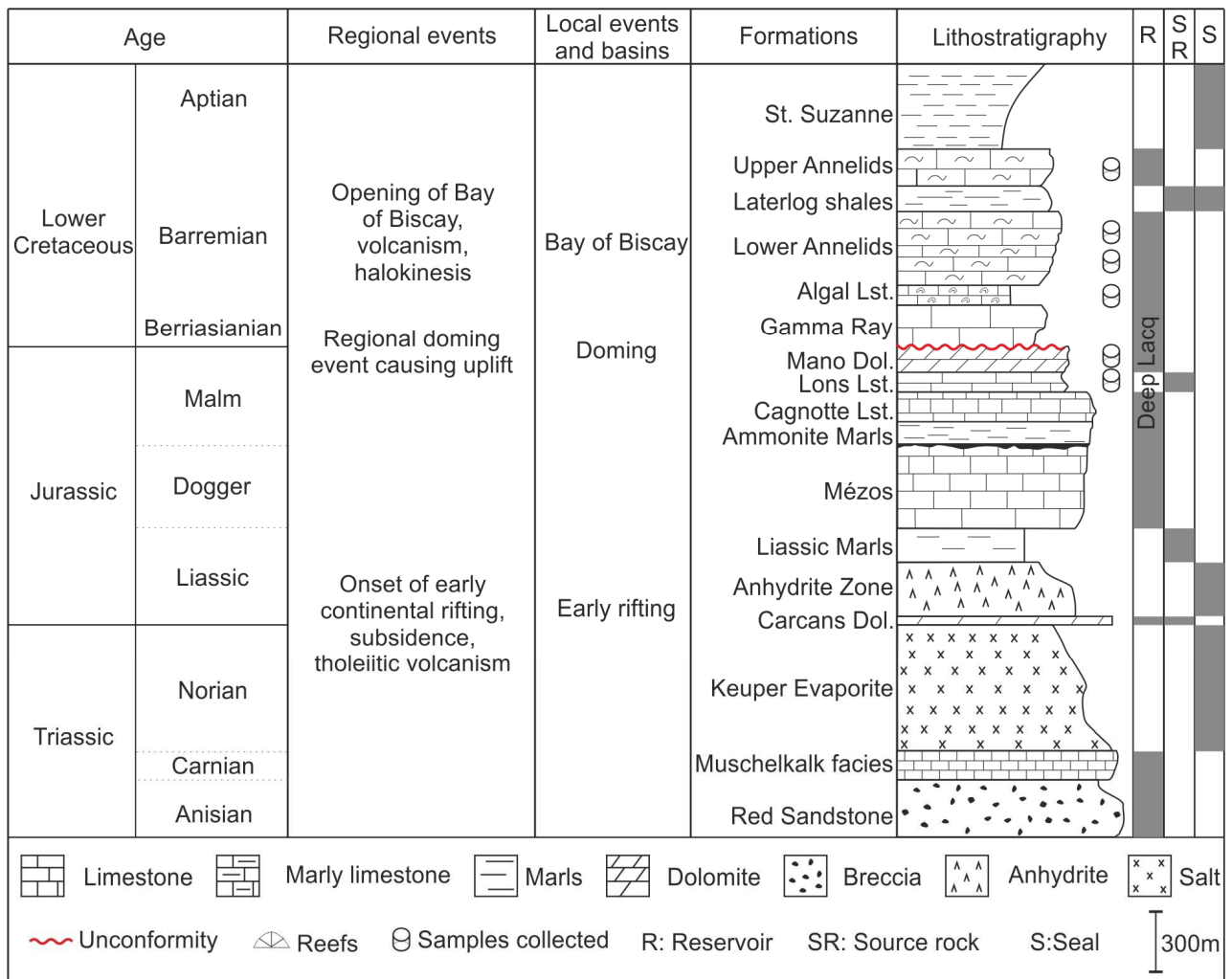
84 The Deep Lacq reservoir is part of the Lacq petroleum system in the Arzacq sub-basin, south of
85 the Aquitaine basin (Fig. 1). The Upper Lacq reservoir (Elias Bahnan et al., 2020) is separated
86 from Deep Lacq by the Sainte Suzanne Marls. Since the target of this work is Deep Lacq, only the
87 Jurassic to Early Cretaceous formations will be discussed. The genesis of the Aquitaine basin and
88 the complete stratigraphic column of Lacq have been extensively described and detailed in a
89 previous work (Elias Bahnan et al., 2020).



90
 91 Figure 1. Structural map of the SW Aquitaine basin, showing the main structural elements, Cretaceous rift
 92 basins and hydrocarbon fields (modified after Biteau et al., 2006 and Elias Bahnan et al., 2020).

93 A simplified lithostratigraphic column with the main geodynamic events is compiled in figure 2. The
 94 basal fluvial sandstones (Red sandstone), limestones and dolomites (Muschelkalk) and evaporites
 95 (Keuper) constitute the Triassic formations (Fig. 2) (Lucas, 1985; Fréchengues et al., 1990; Le
 96 Marrec et al., 1995). The Carcans Dolomites is the first Jurassic formation followed by the Liassic
 97 anhydride unit (Curnelle and Dubois, 1986). The Mézos formation is a clay-rich peloidal limestone
 98 that was deposited during the mid-Jurassic (Le-Marrec et al., 1995). The Kimmeridgian is
 99 represented by the Cagnotte and Lons formations. The latter is composed of organic-rich marly
 100 limestones and dolomites and constitutes an important source rock for the Lacq petroleum system
 101 (Connan and Lacrampe-Couloume, 1993). The Late Jurassic Mano dolomite formation, an
 102 important reservoir in the Deep Lacq petroleum system, overlies the major source rock of the Lons
 103 formation (Fig. 2).

104 The Barremian deposits are subdivided into four formations, from bottom to top (Fig. 2): Algal
 105 Limestones, Lower Annelids Limestones, Laterlog shales, and Upper Annelids Limestone
 106 Formations (Le Marrec et al., 1995). All formations between the Mézos and the Upper Annelids
 107 (inclusive) constitute the main reservoirs of Deep Lacq (Biteau et al., 2006). Rift-related
 108 subsidence during the hyperextension created enough accommodation space for the deposition of
 109 nearly 600m of silty marls of the Sainte-Suzanne formation, which serve as an important seal for
 110 the Deep Lacq reservoirs (Biteau et al., 2006).



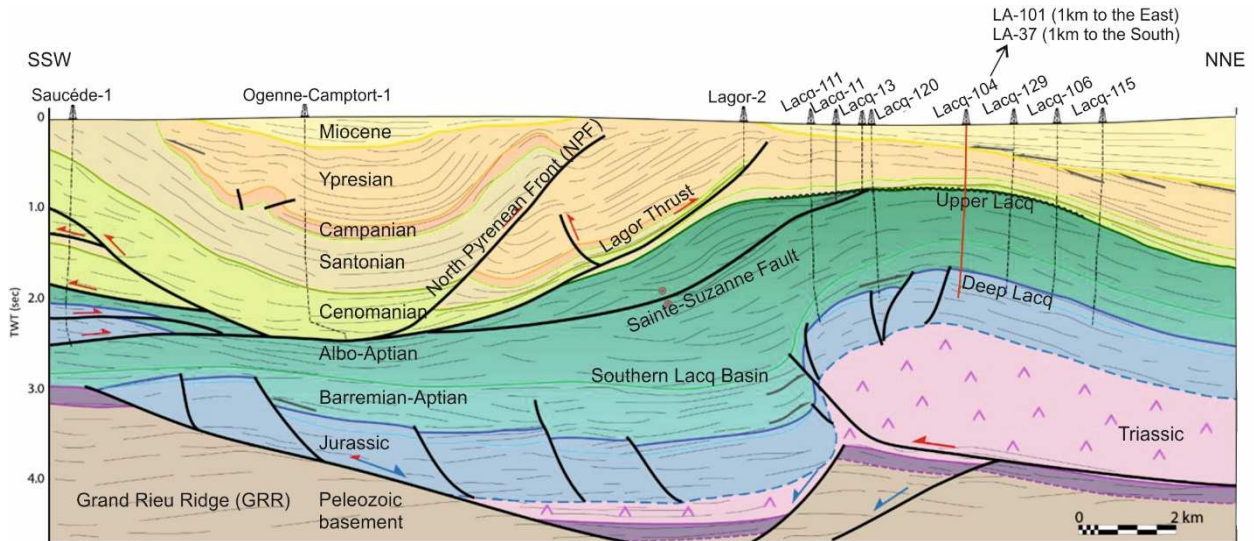
111

112 Figure 2. Simplified stratigraphic column of the Deep Lacq reservoir, source rocks and seals. The summary
 113 of the main tectonic events is also added. (Modified from Elias Bahnan et al., 2020).

114 3 Material and methods

115 3.1 Sampling

116 The studied core samples of Deep Lacq were obtained from wells LA-101 (Latitude: 43.420648°;
 117 longitude: -0.619980°) and LA-104 (Latitude: 43.420421°; longitude: -0.638811°) (Figs. 1 and 3).
 118 Additional public data on these wells can be found at the French national geological survey
 119 website (BRGM: infoterre.brgm.fr). Following the same strategy as that described by Elias Bahnan
 120 et al. (2020) for Upper Lacq, these wells were selected after a preliminary evaluation of nearly 600
 121 thin sections provided from the archives of TOTAL. Targeted formations include the Lons, Mano,
 122 Algal and Annelids formations (Fig. 2) and the selected samples contain abundant coarse cement
 123 phases, fractures, veins, stylolites and cross-cutting relationships observable at the macroscopic
 124 scale.



125
 126 Figure 3. Interpreted seismic line passing NNE-SSW across the Lacq structure in the Arzaq basin (See Fig.
 127 1 for location). No depth-velocity model was available to convert the TWT scale to meters. Instead, drilling
 128 reports indicate that well LA-104 (red line) reaches a true vertical depth (TVD) of 4346m (Elias Bahnan et al.,
 129 2020).

130 3.2 Petrography

131 Cement phases, fractures, veins, stylolites and cross-cutting relationships were recognized by
 132 using optical microscopy, cathodoluminescence and fluorescence. The microscope used for optical
 133 observations was a Zeiss® AX10 with a maximum magnification of x1000. A Cathodyne OPEA
 134 cold cathode instrument (400 mA and 13–15 kV gun current) in the GeoRessources Laboratory
 135 (Nancy, France) was used to perform cold cathodoluminescence (CCL) observations. Ultraviolet
 136 (UV) micro-fluorescence observations were made by using a Zeiss® AX10 optical microscope
 137 equipped with an HBO-100 epi-fluorescence source filtered at 365 ± 5 nm and an LP400 nm
 138 emission filter.

139 3.3 Isotope Geochemistry

140 3.3.1 Oxygen and carbon isotopes ($\delta^{18}\text{O}$ and $\delta^{13}\text{C}$)

141 For $\delta^{18}\text{O}$ and $\delta^{13}\text{C}$ analysis, 24 powder samples were collected with a DREMEL micro-drill from
 142 large veins to avoid mixing of several generations of cements during sampling. A Thermo-Scientific
 143 MAT253 isotope ratio mass spectrometer at CRPG laboratory (Nancy, France) was used for
 144 analysis. Measurements were calibrated using international standards IAEA CO-1, IAEA CO-8 and
 145 NBS 19. Values are reported in per mil (‰) relative to V-PDB (Vienna Pee Dee Belemnite).
 146 Reproducibility was checked by replicate analysis of standards and was $\pm 0.07\%$ (1σ) for oxygen
 147 isotopes and $\pm 0.01\%$ (1σ) for carbon isotopes. The oxygen isotopic compositions of the parent
 148 fluids ($\delta^{18}\text{O}_{\text{water}}$) were calculated by using the trapping temperatures obtained from fluid inclusion

149 modeling and the isotope fractionation equations for carbonates of O'Neil et al.(1969) for calcite
150 and Horita (2014) for dolomite.

151 3.3.2 Multiple sulfur isotopes ($\delta^{34}\text{S}$, $\Delta^{33}\text{S}$)

152 Multiple sulfur isotopes analysis ($\delta^{34}\text{S}$, $\Delta^{33}\text{S}$) was performed on two anhydrite samples drilled from
153 coarse anhydrite cements from the Upper Annelids formation. Sulfur from anhydrite was first
154 extracted by the acid attack technique using a Strongly Reducing Hydriodic Hypophosphorous acid
155 solution (STRIP; Kitayama et al., 2017) to release sulfur as H_2S which then reacted with silver
156 nitrate (AgNO_3) solution to precipitate silver sulfide (Ag_2S). The recovered Ag_2S was then reacted
157 with F_2 to produce SF_6 which was purified by cryogenic traps and gas chromatography. SF_6 was
158 then introduced into a ThermoFinnigan MAT 253 dual-inlet gas-source mass spectrometer. These
159 analyses were carried out at the Institut de Physique du Globe de Paris (IPGP, France). All data
160 are reported in the classical $\delta^x\text{S}$ notation (with $x = 33$ or 34) and expressed in ‰ with respect to the
161 Vienna Cañon Diablo Troilite (V-CDT) standard. Capital delta (Δ) notation is used to represent $\delta^{33}\text{S}$
162 deviations from a reference line (Hulston and Thode, 1965) defined as:

$$163 \quad \Delta^{33}\text{S} = \delta^{33}\text{S} - 0.515 \times \delta^{34}\text{S} \quad (1)$$

164 Standard deviation is better than 0.1‰ and 0.01‰ for $\delta^{34}\text{S}$ and $\Delta^{33}\text{S}$, respectively.

165 3.3.3 U/Pb isotopes

166 To obtain absolute ages on diagenetic phases using U/Pb isotopes, the Laser Ablation-Inductively
167 Coupled Mass Spectrometry (LA-ICPMS) technique was used on dolomite and calcite cements.
168 The analyses were done at the GeoRessources Lab in Nancy, France. Laser sampling was
169 performed with a 193 nm GeoLas Pro ArF Excimer laser equipped with beam homogenization
170 optics. The ablation beam was focused at the sample surface with a constant fluency of 10 J/cm^2 .
171 The repetition rate was stabilized at a frequency of 5 Hz. Helium was then used as a carrier gas to
172 transport the aerosols, released from the targeted sample spot, from the ablation cell to the
173 spectrometer. These aerosols were analyzed by an Agilent 7500c Quadrupole ICP-MS equipped
174 with an Octopole Reaction System and an enhanced sensitivity lens (Cs type, Agilent). The
175 certified glass standards NIST610, NIST612 and NIST614 SRM were used as reference materials
176 (concentrations from Pearce et al. (1997) for the calibration of the LA-ICP-MS.

177 3.4 Fluid inclusion microthermometry

178 Microthermometric measurements were carried out on a Linkam THMS600 heating-cooling stage
179 connected to an Olympus BX51 microscope, at the GeoRessources Lab (Nancy, France). Glass-
180 mounted rock slices of nearly 120 μm thickness were prepared at the litho-preparation unit of the
181 GeoRessources Lab, Nancy. The finished slices were then soaked in acetone for 24 hours to
182 dissolve the glue and unmount the rock samples from the glasses. It is important to note the all
183 samples were prepared under running cold water to avoid pre-heating the samples before
184 microthermometry. Synthetic H_2O pure fluid inclusion standard and a natural aqueous fluid
185 inclusion standard (Ta 33.2) were used for calibration. The phase transition temperatures (i.e.
186 homogenization into the liquid phase) have an accuracy of about $\pm 0.2^\circ\text{C}$, and the obtained data
187 were calibrated accordingly. When the inclusions were too small (less than 5 μm) the glass cover
188 was removed to allow better visibility but the error range increased to up to $+1.5^\circ\text{C}$. Individual rock
189 slices were petrographically observed before being cut down to small chips. To avoid heating other
190 fluid inclusions, the chips were broken down to small sizes to allow one field of view per sample to
191 be studied. Individual fluid inclusions were manually drawn and traced on a copybook to make it
192 easier to trace their location at different depths within the field of view. Homogenization
193 temperatures were recorded in an ascending order after gradually heating the chip at a rate of
194 $2^\circ\text{C}/\text{minute}$. At each temperature increase increment, all selected fluid inclusions were observed
195 and checked for the homogenization of the vapor phase. When the gas phase, the bubble, was too
196 small to observe, the cycling procedure was employed to check for homogenization. After
197 homogenization recording was done, freezing cycles were carried out to record the final ice melting
198 temperatures (T_{m}). T_{m} was converted to equivalent mass % NaCl following the equation of
199 Bodnar (1993). In total, 207 aqueous fluid inclusions and 167 oil inclusions were analyzed,
200 distributed over 3 dolomite samples, 4 calcite samples and 3 anhydrite samples. The choice of
201 these samples was done after a preliminary scan of nearly 80 thin sections, after which those of
202 representative cross-cutting relationships between the diagenetic phases, as well as coarse crystal
203 phases were selected for the FI study.

204 3.5 Raman spectroscopy

205 Raman spectroscopy was used to analyze the gases dissolved in the aqueous fluid inclusions. In
206 samples where the melting temperature was not easily measured, the salinity of the aqueous
207 phase was determined by calculating the area under the Raman band of the water stretching
208 vibrations, which is influenced by the presence of chloride ions (Caumon et al., 2015). Analyses
209 were performed with a LabRAM HR spectrometer (Horiba Jobin Yvon) equipped with a 600 $\text{g}\cdot\text{mm}^{-1}$
210 grating and an Edge filter. The confocal aperture was 200 μm and the slit aperture was 200 μm .
211 The excitation beam was provided by a Stabilite 2017 Arp laser at 457.9 nm to 514.5 nm with a
212 power of 200 mW. The acquisition time was 20 seconds x10 accumulations to optimize the signal-
213 to-noise ratio. Quantification of the CH_4 was done following the best fit, second-order polynomial
214 curve obtained from the calibrations of Caumon et al. (2015). The uncertainty on the calculated
215 CH_4 molality value resulting from the calibration function ($\pm 1\sigma$) is less than $\pm 2\%$ (Caumon et al.,
216 2014). The area ratios were calculated by using the integral tool in LabSpec. The spectra with
217 peaks between 2550 and 2600 cm^{-1} were analyzed to quantify the concentration of HS^- (2570 cm^{-1})
218 and H_2S (2590 cm^{-1}). To quantify the ratios of these species, the Raman spectra were treated
219 using the *Opus (Bruker)* software. After baseline correction, the spectra were decomposed by
220 using the Levenberg-Marquardt algorithm to differentiate the HS^- and H_2S species. Concentration
221 of sulphur aqueous species were calculated following the outline of Jacquemet and Pironon
222 (2006).

223 3.6 Fluid modeling

224 Aqueous inclusions thermodynamic (AIT) modeling is a calculation procedure that relies on the
225 homogenization temperature and the chemistry of the aqueous fluid phases to draw the pressure-
226 temperature phase diagrams using an Equation of State (EoS). Similarly, petroleum inclusions
227 thermodynamic (PIT) modeling allows the construction of phase diagrams of petroleum inclusions.
228 The intersection of isochores of aqueous and oil inclusions, known as the double-isochore
229 technique (Bourdet et al., 2008) indicates the trapping P-T conditions of fluids. Given limitation in
230 detecting the exact quantity of H_2S dissolved in the oil inclusions, 2 compositions of oil inclusions
231 are modeled, such that the first has 0% H_2S while the second has 20% H_2S . These two
232 compositions allow the modeling of the minimum and maximum possible compositions. To date, no
233 EoS exist that can model the phase evolution of a fluid system composed of H_2O - NaCl - CH_4 - H_2S -
234 HS^- at temperatures less than 200°C. Hence, a workflow is developed that can take into account

235 the impact of H₂S on the homogenization pressure (P_h) and which can be extrapolated to the P-T
236 domain of the studied fluid inclusions. After obtaining the composition of the aqueous phases using
237 Raman spectroscopy (section 3.5), the EoS of Duan and Mao (2006) was used, which is suitable
238 to model fluids containing CH₄-H₂O-NaCl at temperatures between 0-275°C pressures of 1-2000
239 bars. With this EoS, the homogenization pressure (P_{h1}) was obtained (step 1 – table 1). The
240 obtained P_{h1} was then added to the P_h obtained from the EoS of Zhang and Frantz (1987) hereby
241 labeled P_{hz-f} (step 2 – table 1). This EoS is commonly used to draw isochores between 0-700°C
242 and pressures up to 3000 bars for the H₂O-NaCl fluid system. Adding P_{h1} to P_{hz-f} gives P_{h2} and
243 allows the construction of the isochores of CH₄-bearing fluids without the impact of H₂S (step 3 –
244 table 1). To model the influence of H₂S on the isochores, Loner 9 software
245 (<http://fluids.unileoben.ac.at/Computer.html>), which relies on the EoSs of (Duan et al.,1992a,b;
246 1996) was used. These EoSs allow the modeling of H₂O-CO₂-CH₄-H₂S fluids systems but at
247 temperatures higher than the supercritical and pressures up to 25000 bars. Since the validity
248 domain is beyond the temperature range of the fluid inclusions, simulations were made at T=250°C
249 which rendered a valid solution in the thermodynamic modeling of the Loner 9 software. The
250 purpose of this calculation step is to analyze the difference in homogenization pressures with and
251 without H₂S. This step gave P_{h3} that shows the homogenization pressure of an H₂S-bearing fluid at
252 250°C (step 4 – table 1). However, since the EoS of Duan and Muao (2006) was used to draw the
253 isochore of fluids without H₂S and to be consists when comparing different isochores with or
254 without H₂S, the EoS of Duan and Muao (2006) was again used to calculate another
255 homogenization pressure P_{h4} at T=250°C (step 5 – table 1). P_{h4} is similar to the P_{h1} with the
256 exception of the higher temperatures. The purpose of calculating both P_{h3} and P_{h4} is to compare
257 how H₂S changes the P_h of a fluid that homogenized at T=250°C. ΔP_h is then the difference
258 between P_{h3} and P_{h4} such that $\Delta P_h = P_{h3} - P_{h4}$ (step 6 – table 1). To be able to model the fluids at
259 their real homogenization temperatures and with the impact of H₂S, i.e. with ΔP_h , P_{h4} is calculated
260 as $P_{h4} = P_{h1} + \Delta P_h$ (step 7 – table 1). Finally, a second isochore is constructed by adding P_{h4} to the
261 P_{hz-f} (step 8 – table 1). These calculation steps are summarized in table 1.

262 Table 1. Workflow showing the calculation steps and their purposes for the AIT modeling.

Step	Parameters	T (°C)	EoS used	Validity range	Purpose	Gases present
1	P_{h1}	T_h	Duan and Muao (2006)	T=0-275°C P=1-2000 bars	Impact of CH ₄ on P_h	CH ₄
2	P_{hz-f}	T_h	Zhang and Frantz	T=0-700°C	Construction of isochores	None

			(1957)	P=0-3000 bars		
3	$P_{h2} = P_{h1} + P_{hz-f}$	T_h	Zhang and Frantz (1957) + P_{h1}		Construction of isochores	CH ₄
4	P_{h3}	250°C	Duan et al. (1992a,b; 1996)	T>supercritical P=0-25000 bars	Impact of CH ₄ +H ₂ S on P_h	CH ₄ +H ₂ S
5	P_{h4}	250°C	Duan and Muao (2006)	T=0-275°C P=1-2000 bars	Impact of CH ₄ on P_h	CH ₄
6	$\Delta P_h (P_{h3} - P_{h4})$	250°C			Impact of H ₂ S on P_h	CH ₄ +H ₂ S
7	$P_{h4} = P_{h1} + \Delta P_h$	T_h			Impact of CH ₄ +H ₂ S on P_h at T_h	CH ₄ +H ₂ S
8	$P_{h5} = P_{hz-f} + P_{h4}$	T_h	Zhang and Frantz (1957) + P_{h4}		Construction of isochores	CH ₄ +H ₂ S

263

264 The modeling workflow assumes that the difference in P_h remains constant when extrapolating
265 from 250°C to the measured T_h . This assumption is validated by calculating the homogenization
266 pressures again at 300°C, which showed constant difference between the two data sets. With this
267 consistency, it is assumed that the part of the isopleth at 250°C is a straight line and the
268 extrapolation from 250°C to the measured T_h remains along this straight line. Therefore, after
269 adding H₂S to the fluid system, the new isopleth is expected to be parallel and hence adding P_{h1} to
270 ΔP_h can be justified.

271 Petroleum inclusions thermodynamic modeling (PIT) allows the construction of the P-T phase
272 diagram of a petroleum inclusion. This was done by obtaining a 3D volumetric reconstruction of oil
273 inclusions to determine the bubble point curves and eventually the phase diagrams. These 3D
274 images were obtained by compiling stacks of images captured using a Confocal Scanning Laser
275 Microscope (CSLM) fitted with a Biorad Rainbow system adapted to a Nikon inverted microscope
276 at the GeoRessources Lab (Nancy, France). Emission source was an Argon laser at 488 nm and a
277 diode laser at 405 nm. Observations were done through an oil-immersed 60x objective lens. This
278 method allows volume estimation of the cavity of a petroleum inclusion with an accuracy greater
279 than 95%. The x-y resolution was 0.2 μm and the resolution along the z-axis was approximately
280 0.5 μm . The studied chips were glued on a glass plate using paraffin to avoid any movement
281 during acquisition. For a higher accuracy, the radii of the gas phases in oil inclusions were
282 measured prior to CSLM measurements by using optical light microscopes. To construct the phase
283 diagrams, the PIT modeling software developed by Thiéry et al. (2002) was used. To model the
284 impact of H₂S on the isochores of oil inclusions, and given the limitations in accurately quantifying
285 H₂S dissolved in oil inclusions using infrared spectroscopy, two isochores of each oil inclusion are

286 modeled; one containing 0% H₂S and the other 20% H₂S. These two compositions represent
287 geologically reasonable end-members and can help in placing a range of trapping P-T conditions.

288 4 Basin Modeling

289 The geodynamic history of the Lacq petroleum system was modeled using Petromod 1D software.
290 A synthetic well combining the entire stratigraphy of wells LA-101 and LA-104 was prepared from
291 drilling reports and used as the main input source for the model. Boundary conditions were set by
292 using the paleowater depths (PWD), sediment-water interface temperatures (SWIT) and basal heat
293 flow (HF) as discussed by Elias Bahnan et al. (2020). Vitrinite data were collected from TOTAL
294 archives and were used to calibrate and validate the thermal history of the model.

295 5 Results

296 5.1 Petrography of diagenetic phases

297 Precursor limestones consist of chalky mudstones, wackestones to bioclastic packstones (Fig. 4-A
298 and B). Bioclastic material consists of microforaminifera, shells of mollusks, sponges, echinoderms
299 and annelids, with sizes reaching up to 2 mm.

300 5.1.1 Micritization

301 Microborings filled by micrite growing from the walls of the bioclasts inwards are observed (Fig. 4-
302 A). This phenomenon, in some cases, is widespread enough to engulf the entire bioclast and
303 hence disrupts the original structure of the grain.

304 5.1.2 Pyrite

305 Clusters and agglomerates of rounded, to sub-rounded framboidal pyrite crystals are observed on
306 shells and bioclastic fragments and inside bioclastic micropores (Fig. 4-B). Individual crystal sizes
307 range between few μm up to nearly 40 μm . Under reflected light microscopy, a golden yellowish
308 color, typical of pyrites, is observed (Fig. 4-B).

309 5.1.3 Dolomite 1

310 This dolomite phase is represented by rhombohedral dolomite crystals with average crystal sizes
311 between 30 and 80 μm (Fig. 4-C). CCL microscopy reveal a dark core and a bright yellow to

312 orange luminescent external rim. The core of some crystals is dissolved thus creating intra-
313 crystalline secondary porosity.

314 5.1.4 Dissolution

315 Moldic porosity, later infilled by other cements (Fig. 4-D), provide indirect evidence that dissolution
316 may have occurred before cementation. This dissolution phase is also evidenced by the corroded
317 rims and dissolved cores of the Dol. 1 phase (Fig. 4-C).

318 5.1.5 Calcite 1

319 This is a diagenetic phase limited to the Annelids Limestone Formations. It appears as zoned
320 dog-tooth to columnar calcite cements lining the walls of moldic pores (Fig. 4-D and E). Under
321 CCL, zoning evolve from a dull-orange/brown core to progressively brighter orange to yellow colors
322 (Fig. 4-E). Crystal sizes range between 5 and 30 μm . Zoning is well-developed and show no
323 evidence of recrystallizations. No aqueous or oil inclusions are observed in this phase.

324 5.1.6 Calcite 2

325 This second calcite phase appears as a yellow-orange overgrowth engulfing the dog-tooth calcites
326 of the Annelids Limestone Formations. The crystals are coarse ($\sim 300\mu\text{m}$), equant and have sub-
327 rounded terminations which helps to distinguish them from other calcite phases. Under CCL, no
328 zoning is observed and crystals are characterized by uniform shades of yellow to dull orange
329 colors with no evidence of recrystallization (Fig. 4-E). When this phase develops in open vuggy
330 pores, crystal-sizes can reach up to 650 μm . No aqueous or oil inclusions were observed in this
331 phase.

332 5.1.7 Mechanical compaction

333 This episode is represented by the collapse of the rounded bioclasts and shell fragments. Figure 4-
334 F shows an example of an annelid stem that has undergone mechanical compaction as witnessed
335 by the damage of its walls.

336 5.1.8 Stylolite 1

337 This is a minor compaction phase represented by low-amplitude stylolites with a horizontal
338 orientation parallel to the stratigraphy. These stylolites are cross-cut by calcite/dolomite-infilled
339 fracture A1 (Fig. 5-A).

340 5.1.9 Fracture A1

341 Fractures A1 cross-cut the host rocks and stylolites 1 (Fig. 5-A) and are characterized by straight
342 walls with thicknesses between 5µm up to 3cm. Infilling cements are composed of saddle dolomite
343 and calcite.

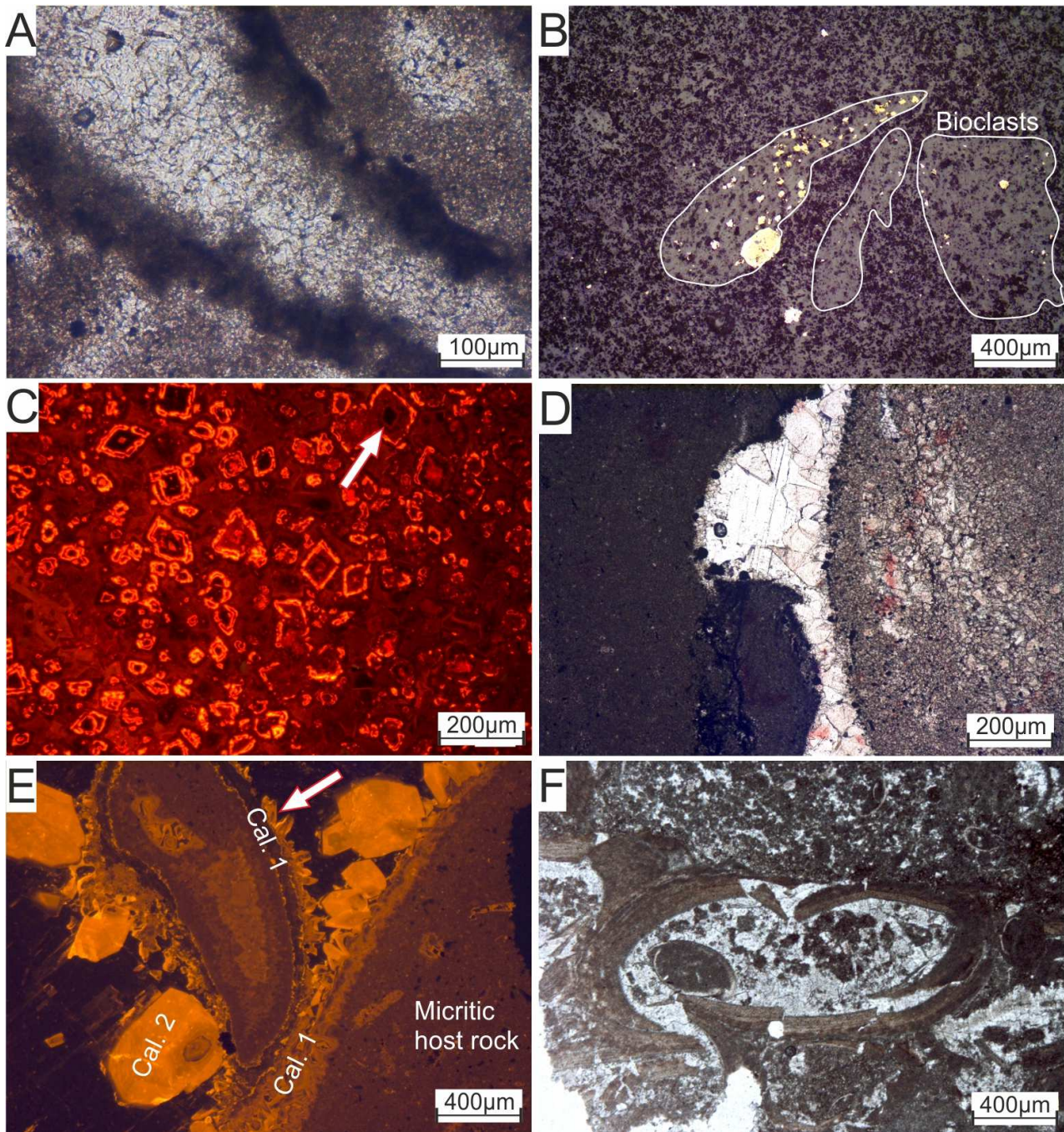
344 5.1.10 Dolomite 2

345 In the fracture set A, very coarse dolomite crystals, reaching up to 3mm in size, with curved faces
346 are observed (Fig. 5-B). They have turbid cores and growth zones rich in fluid inclusions. Oil
347 inclusions are also identified in the growth zones. CCL microscopy reveal a zoned core with
348 luminescence alternating between different shades of red. In some cases, the dolomite 2 phase,
349 known as saddle dolomites, is partially replaced by a calcite phase (Fig. 5-B).

350 5.1.11 Fracture A2

351 The occurrence of host-rock fragments inside fractures A (Fig. 5-B) sealed by the calcite 3
352 cements suggests a fracturing mechanism during or prior to the cementation of the calcite 3.

353



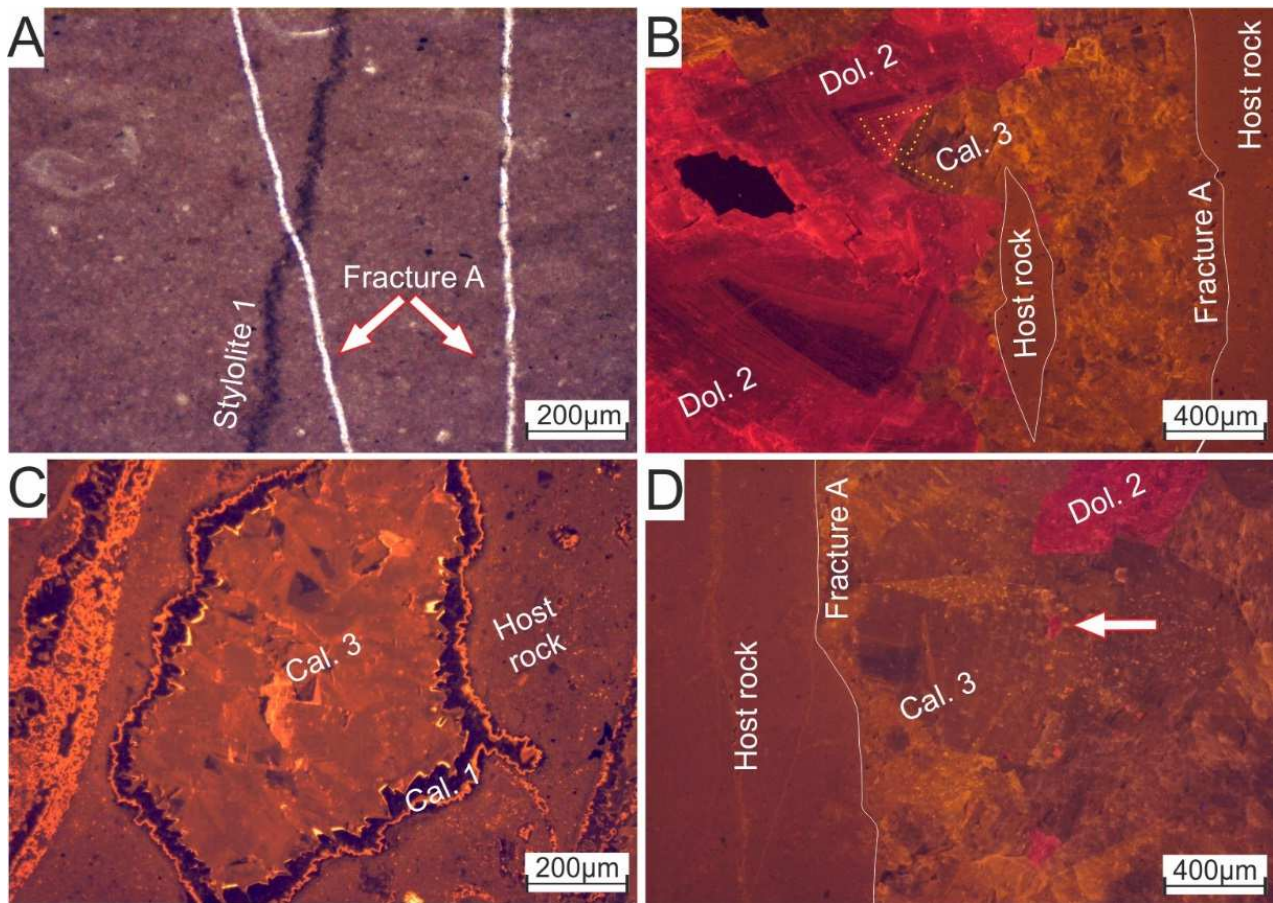
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355 Figure 4. A: Plane polarized light (PPL) image of a shell fragment affected by destructive micritization. The
 356 shell is later replaced by other calcite phases. B: reflected light image showing ghosts of bioclasts with local
 357 concentrations of rounded to sub-rounded pyrites. C: CCL image of partially dissolved rhombic dolomite
 358 crystals with a brightly luminescent external rim. Arrow point to a dissolved core within a crystal. D: shell
 359 mold cemented by calcite and anhydrite cements. This shell was initially dissolved and moldic porosity was
 360 filled by younger cement phases. E: CCL image of the zoned “dog-tooth” calcite 1 phase precipitated at the
 361 contact with the host rock and covered by the yellow-luminescent calcite 2 phase. F: Broken annelid tubes
 362 as evidence of mechanical compaction.

363 5.1.12 Calcite 3

364 This dominant cement phase occludes most of the fracture and vuggy porosity (Fig. 5-C) and
 365 significantly reduces reservoir quality. These calcites are drusy with crystal-sizes reaching up to 8
 366 mm (Fig. 5-D). Lack of uniform CCL zoning and the heterogeneous brown to dull orange colors

367 suggest possible recrystallization. Relics of saddle dolomites are frequently observed in Calcite 3
368 (arrows in fig. 5-D). This phase is rich in oil and aqueous inclusions.



369

370 Figure 5. A: PPL photomicrograph of stylolite 1 crosscut by fracture set A. B: CCL image of the cross-cutting
371 relationships between fracture A, dolomite 2 and calcite 3 phases. Fragments of host rocks in the fracture
372 are indicators of fracture reopening mechanism. Dolomite 2 is partially replaced by calcite 3 (see the yellow
373 dotted line between Dol. 2 and Cal. 3). C: CCL image of a vuggy pore now filled by Cal. 1 and Cal. 3 calcites.
374 D: CCL image of coarse calcite 3 crystals. Arrows indicate remnants (relics) of Dol. 2 after replacement. Cal.
375 3 shows heterogeneous shades of yellow to dull orange luminescence.

376 5.1.13 Fracture B

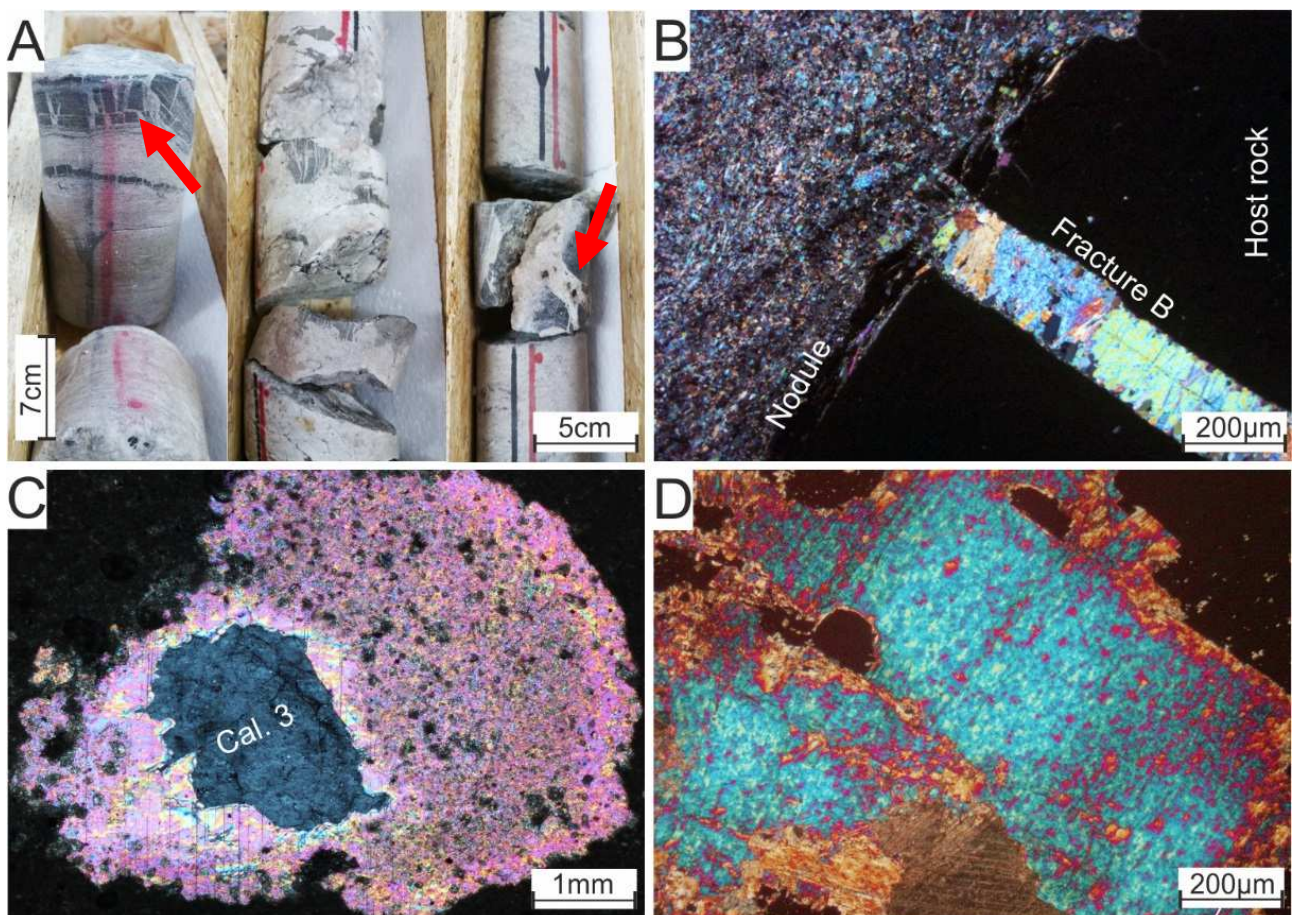
377 Fracture set B postdates calcite 3 and is filled by anhydrite. The width of fractures can reach up to
378 1cm (arrows in Fig. 6-A) and appear mostly as vertical fractures when viewed at the macroscopic
379 scale.

380 5.1.14 Anhydrite

381 Anhydrite was observed inside the vertically oriented fracture set B in wells LA-101 and LA-104
382 (Fig. 6-A) and tend to form massive nodules and concretions that replace the initial fabrics of the
383 host lithology. The thickness of anhydrite layers, observed in the studied cores (Fig. 6-A) can reach
384 up to 50cm. Stratigraphically, the massive anhydrites lie immediately under the Sainte-Suzanne
385 marls where they partially replaced the Upper Annelids Limestone Formation, and reach around 15

386 to 20 meters in total thickness. This phase also occurs as pore-blocking cements in all the studied
387 formations. Under plane polarized light (PPL), anhydrites appear clear white to creamy, while
388 under cross-polarized light (CPL), an array of bright second order colors were observed (Fig. 6-B,
389 C and D). This is the last diagenetic cement that replaces younger calcite and dolomite phases. No
390 luminescence was detected under CCL. Similarly, under PPL, crossed-polarized light (CPL) and
391 CCL, no evidence of gypsum precursor was observed. Several morphologies of anhydrites have
392 been identified.

- 393 1. Coarse and tabular crystals with sizes reaching up to 2 mm (Fig. 6-B and D), often showing
394 evidences of calcite and dolomite replacement (Fig. 6-C), and rich in oil and aqueous
395 inclusions.
- 396 2. Fine fibrous crystals filling nodules inside the host rocks. Fracture set B is often observed
397 connected to these nodules (Fig. 6-B). Individual crystal size is no more than 50 μm (Fig. 6-B).
398 These are too small to observe aqueous or oil inclusions.



399
400 Figure 6. A: Photographs of well LA-101 and LA-104 with massive anhydrites few meters below the Sainte
401 Suzanne Formation. Arrows point to cm-scale veins. B: CPL image of an anhydrite-filled fracture C in a host
402 micrite matrix in contact with an anhydrite nodule. C: CPL image of an anhydrite with partial replacement of a
403 calcite 3 crystal. D: CPL image of coarse anhydrite crystals with bright 2nd order birefringence.

404 5.1.15 Stylolite 2

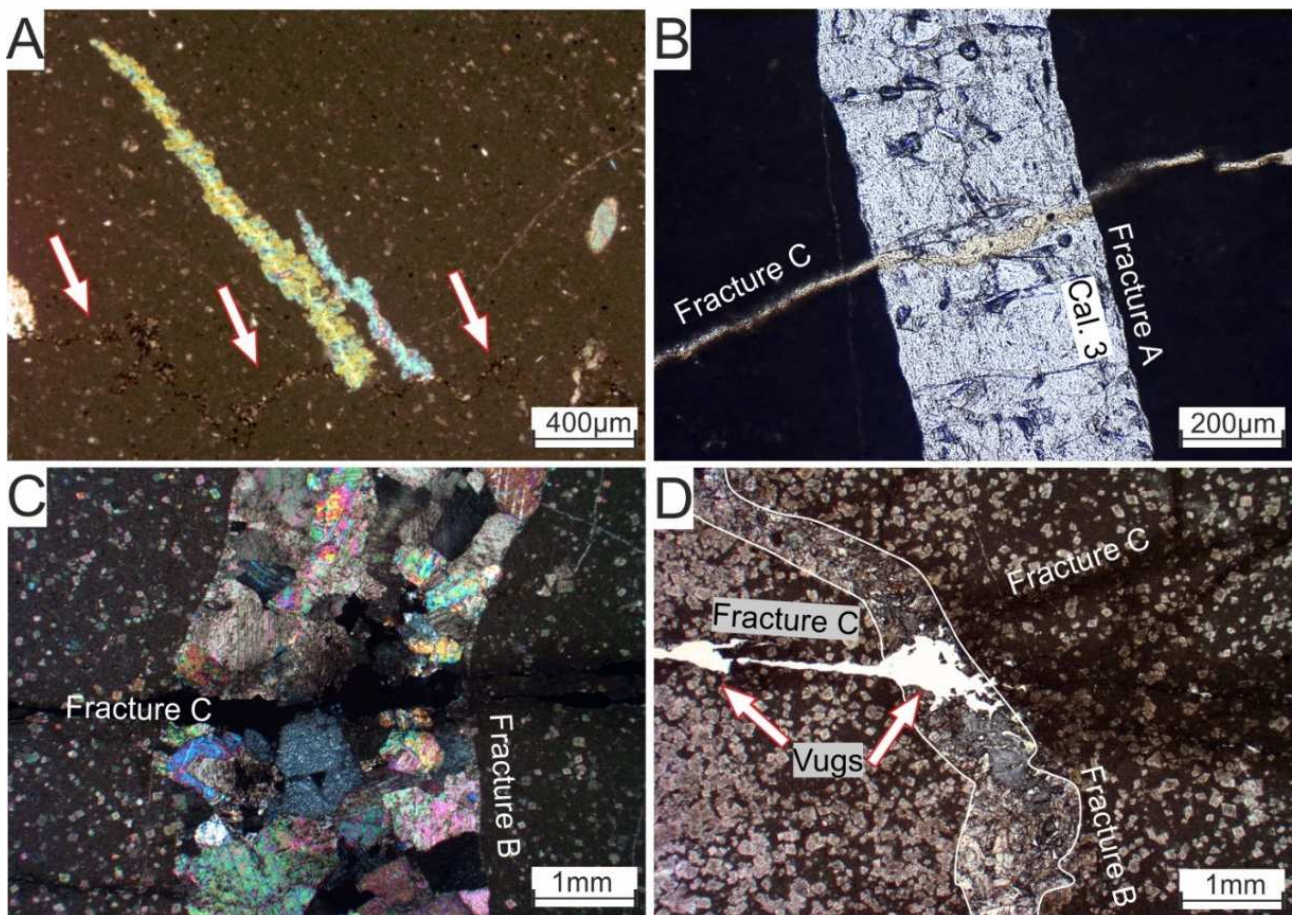
405 These stylolites have amplitudes in the range of 200 to 600 μ m, are mostly oblique relative to the
406 vertical wells and crosscut anhydrites (Fig. 7-A).

407 5.1.16 Fracture C

408 This last fracturing phase affected all previous phases and generated fracture porosity in the
409 reservoir of Deep Lacq. The fracture set C crosscuts calcite-filled fracture A (Fig. 7-B) and
410 anhydrite-filled fracture B (Fig. 7-C).

411 5.1.17 Dissolution

412 A final phase of dissolution is represented by solution-enhanced fractures that enlarged fracture
413 set C (Fig. 7-C) where in some cases it created vuggy porosities (Fig. 7-D).

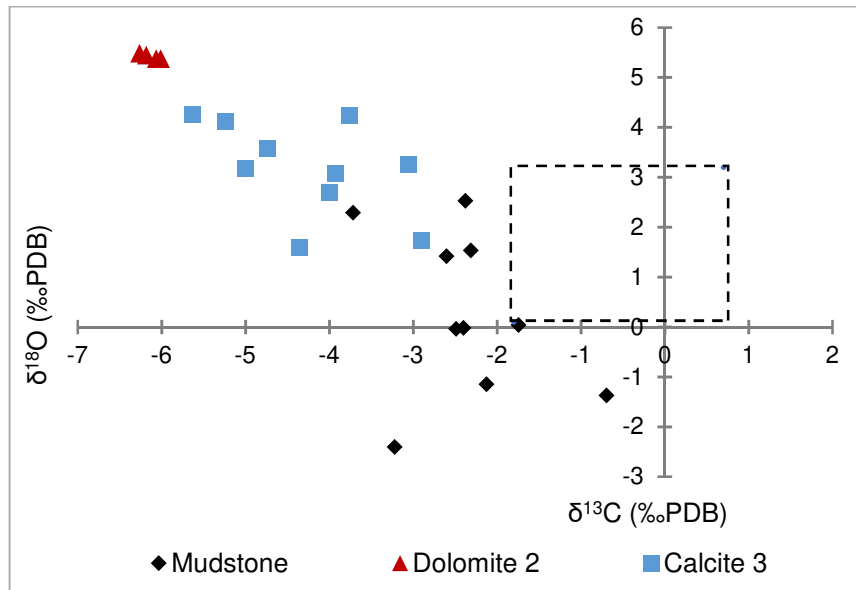


414

415 Figure 7. A: CPL photomicrograph of anhydrite affected by stylolite 2 (arrows). B: PPL image of a fracture A
416 filled by calcite (Cal. 3) crosscut by an open fracture C. C: Anhydrite-infilled fracture B crosscut by fracture C.
417 D: Fracture B crosscut by a fracture C with a solution-enhanced vuggy-porosity.

418 5.2 Isotope Geochemistry

419 $\delta^{18}\text{O}$ and $\delta^{13}\text{C}$ isotopes were obtained from the micritic host rocks (n=10), Dol. 2 (n=4) and Cal.3
 420 (n=10) cements (Fig. 8) of the Lons, Algal and the Annelids formations. $\delta^{13}\text{C}$ of the host black
 421 mudstones are in the range of -2.4‰ to +2.5‰ PDB and $\delta^{18}\text{O}$ between -3.7‰ to -0.6‰ PDB. Dol.
 422 2 have $\delta^{13}\text{C}$ values between +5.5 and +5.6 ‰ PDB and a depleted $\delta^{18}\text{O}$ between -6 and -6.2 ‰
 423 PDB. $\delta^{13}\text{C}$ of calcite 3 is between +1.6 and +4.3 ‰ PDB while $\delta^{18}\text{O}$ varies between -2.8 and -5.6
 424 ‰ PDB. Error ranges for individual data points are within the symbols plotted in figure 8 and are
 425 between ± 0.01 to ± 0.07 (1σ) from the reported values.



426

427 Figure 8. Cross-plot of $\delta^{18}\text{O}$ and $\delta^{13}\text{C}$ isotopes for Deep Lacq. Dashed box represents the Late Jurassic
 428 seawater signature (Veizer et al., 1999).

429 The two analyzed anhydrites from the Annelids formation were sampled from massive anhydrites
 430 at 3514.5 m (well LA-104) and 3619.5 m (well LA-101) and they respectively show multiple sulfur
 431 isotopes signatures at: $\delta^{34}\text{S} = 19.12\text{‰} \pm 0.006$, $\Delta^{33}\text{S} = 0.023\text{‰} \pm 0.009$ and $\delta^{34}\text{S} = 18.97\text{‰} \pm 0.005$,
 432 $\Delta^{33}\text{S} = 0.017\text{‰} \pm 0.011$. The error range represents 1σ for dual-inlet mass-spectrometry
 433 measurements. These values are consistent with signatures observed in the Keuper carbonate
 434 associated sulfates (CAS) at $\delta^{34}\text{S} = 20.6\text{‰} \pm 5.4$ and $\Delta^{33}\text{S} = 0.000\text{‰} \pm 0.025$ (Wu et al., 2010 –
 435 2014).

436 The detected amounts of U and Pb in calcite and dolomite samples ranged between 0.001 and a
 437 0.004 ppm. Of the 52 measurements, 18 were below the detection limits of the spectrometer.
 438 These quantities of U and Pb are too low to meet the minimum pre-requisites for U/Pb
 439 geochronology and hence proceeding with this step was not possible.

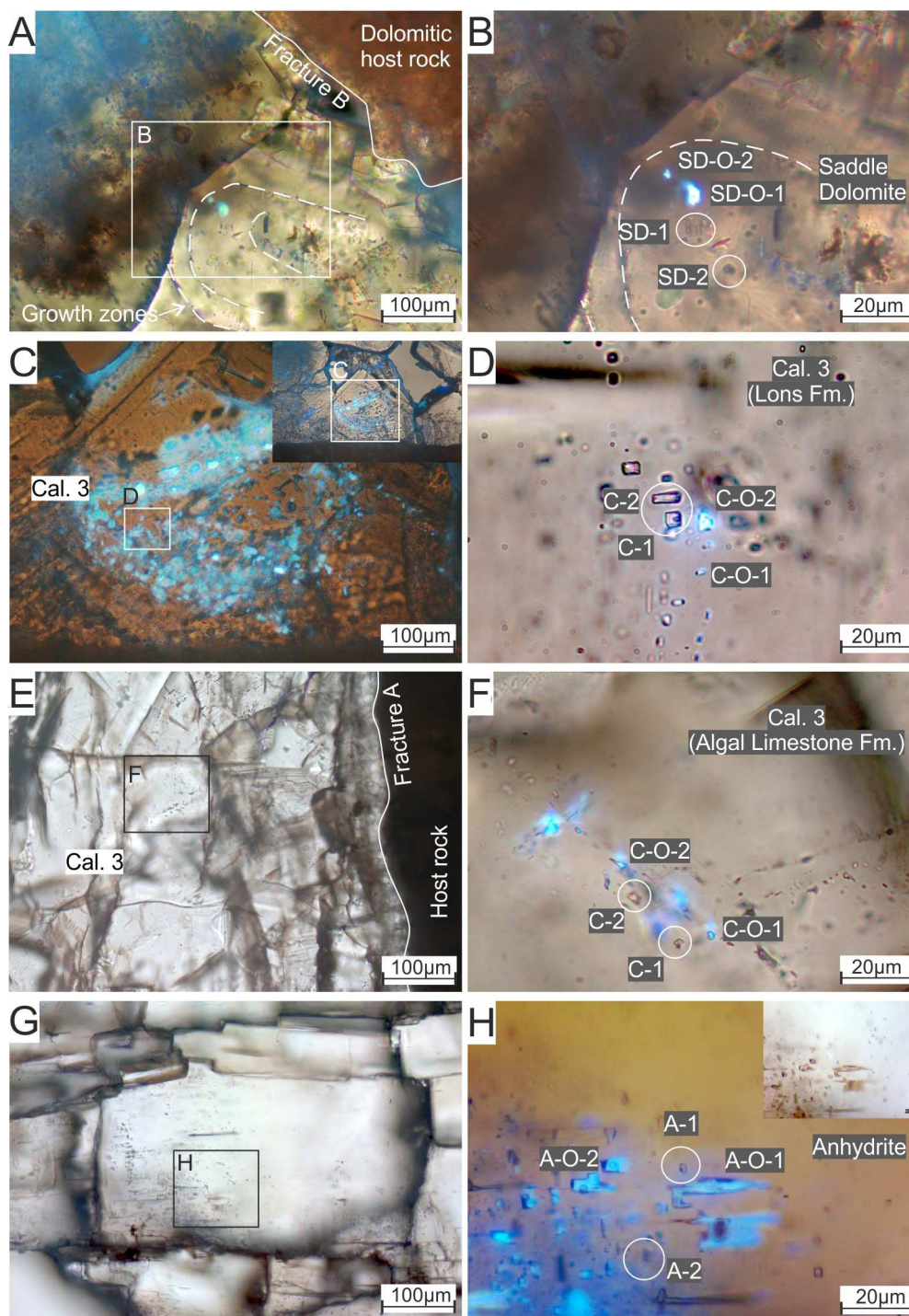
440 5.3 Fluid inclusions

441 5.3.1 Petrography

442 The Dol. 2, Cal. 3 and Anhydrite are the coarsest cement phases in Deep Lacq with fluid inclusions
443 of suitable sizes allowing proper observations and analyses.

444 In Dol. 2 crystals, two-phase aqueous inclusions and two-phase oil inclusions are observed along
445 growth zones (Fig. 9-A and B). The aqueous inclusions have mostly irregular to oval shapes with
446 sizes ranging between 3 to 8 μm . These aqueous inclusions are located close to two-phase light
447 brown oil inclusions which, when viewed under UV light, display a strongly fluorescent blue to
448 white color (SD-O-1-2 in Fig. 9-B). The liquid/vapor ratio noted as Fv%, of these studied inclusions
449 was visually assessed and approximated to nearly 8%. Fv% is constant across all studied FIs in
450 the dolomite phase.

451 Special attention was given to check for possible necking down or post-entrapment alteration of
452 the fluid inclusions. A necked-down inclusion would typically appear as a split-up inclusion
453 between a single-phase aqueous inclusion and a smaller inclusion dominated by the vapor phase
454 (Diamond and Tarantola, 2015). No petrographic evidence was found of such inclusion or possible
455 intermediate inclusions in the process of being necked down.



456

457 Figure 9. A: Coarse dolomite 2 crystal with oil and aqueous inclusions. The growth zones are traced in the
 458 dashed white lines B: The oil inclusions SD-O-1 and SD-O-2 show a strongly fluorescent blue to white color.
 459 White ovals contain two-phase aqueous inclusions SD-1 and SD-2. C: Aqueous and blue-fluorescent oil
 460 inclusion in Cal 3 of the Lons Limestone formation. The assemblage where the aqueous and oil inclusions
 461 are found terminate before the edge of the crystal and does not indicate continue to other crystals. D: Blue
 462 fluorescent oil inclusions (C-O-1 and C-O-2) and aqueous inclusions (C-1 and C-2). E: Fluid inclusions in
 463 Cal. 3 of the Algal Formation. F: Blue fluorescent oil inclusions (C-O-1 and C-O-2) and aqueous inclusions
 464 (C-1 and C-2). G: Photomicrograph of an anhydrite crystal with a cluster of aqueous and oil inclusions. Each
 465 anhydrite crystal has its own cluster of inclusions, none with clusters extending to other crystals. H: UV light
 466 showing the strong blue color of oil inclusions (A-O-1 and A-O-2) adjacent to the two-phase aqueous
 467 inclusions (A-1 and A-2). Notice the brown PPL color of the oil inclusions, different from those trapped in the
 468 saddle dolomites and calcite phases .

469 In the Cal. 3 cement, two-phase aqueous inclusions are identified in clusters limited to individual
 470 crystals (Fig. 9-C, D, E and F) These aqueous inclusions have inverted geometries, rectangular,
 471 oval and irregular shapes with sizes ranging from 5 up to 70 μm . The oil inclusions have a strongly

472 fluorescent blue to white colors under UV light (Fig. 9-D and F). The Fv% of the aqueous phases
473 was consistent and ranged between 9-10% across all observed inclusions in the calcites

474 Two-phase primary aqueous and oil inclusions are observed in the coarse anhydrite crystals. The
475 aqueous inclusions have cubic to rectangular shapes with relatively small sizes in the order of 5-15
476 μm (Fig. 9-E). Oil inclusions in anhydrites are two-phased, with brown to orange color under PPL
477 and strong blue fluorescence under UV (Fig. 9-H). They have elongated rectangular to inverted
478 geometric shapes. These inclusions are observed in clusters within crystals. The clusters are
479 confined to individual crystals and do not extend to nearby cements. Similar to the case of the FIs
480 trapped in the dolomite and the calcite phase, the Fv% of all studied aqueous inclusions was
481 consistent and approximated to around 10%.

482 5.3.2 Microthermometry

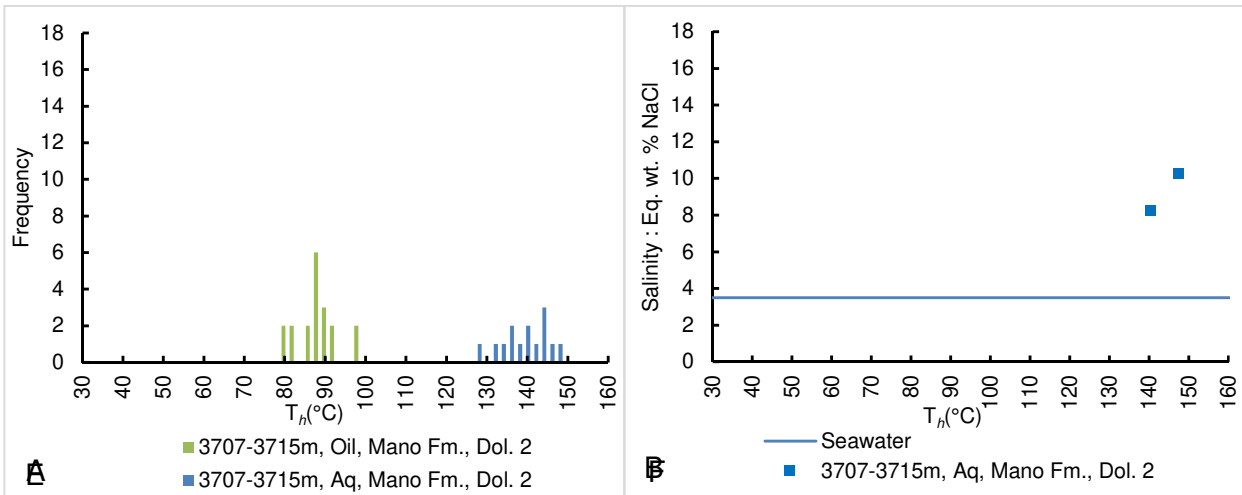
483 The studied sample of Dol. 2 was retrieved from the Mano Dolomite Formation at a depth between
484 3707-3715 m. The aqueous inclusions homogenized into the liquid phase at a T_h range of 132-
485 148°C with a mode T_h at 144°C (Fig. 10-A). The oil inclusions also homogenized into the liquid
486 phase at a T_h range of 80-98°C with a mode T_h at 88°C (Fig. 10-A). After freezing down to -100°C,
487 phase changes were observed and the final melting ice temperatures were detected. Only two
488 reliable T_{m_i} were recorded between -6.8 and -5.2°C that correspond to salinities of 8.2 and 10.2
489 eq. wt.% NaCl respectively (Fig. 10-B).

490 Three samples of Cal. 3 were studied from the Lons Formation (4116m), the Annelids formation
491 (3488 m) and the Algal limestone formation (3627 m). In the Lons formation, one population of two-
492 phase aqueous is observed with a T_h range of 106-110°C with a T_h mode at 108°C (Fig. 10). This
493 population was observed in clusters and its association to growth zones was not possible with
494 confidence, as seen in figure 9-C. These aqueous fluid inclusions have salinities between 12 and
495 13 wt.% NaCl and are present with a dominant oil inclusion phase that homogenized at a T_h range
496 of 50-66°C with a mode T_h at 56°C.

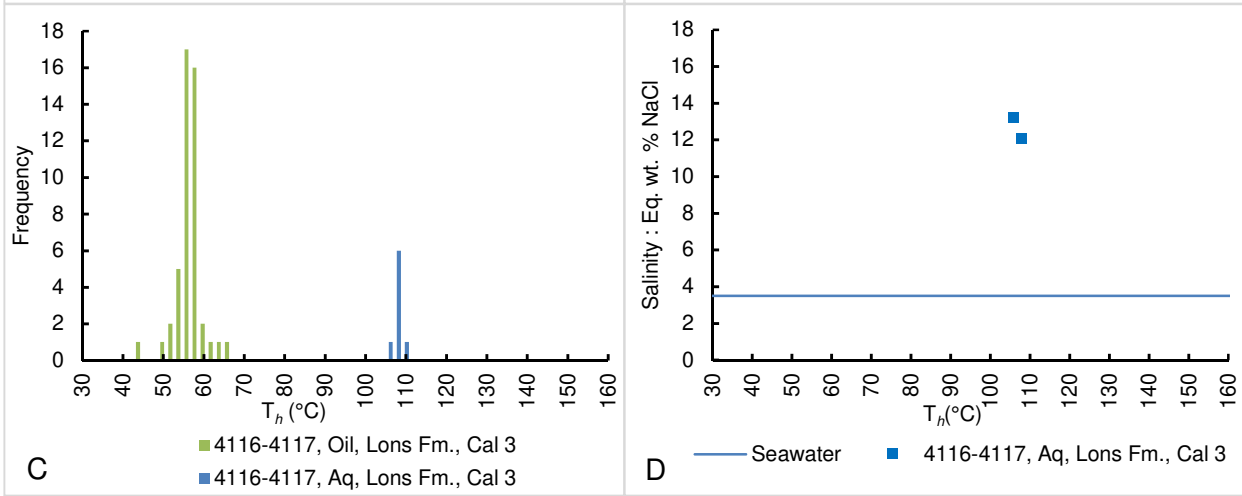
497 In the Cal. 3 samples of the Algal Limestone formation, two populations of two-phase aqueous
498 inclusions are present. The first was observed along growth zones and homogenized into the liquid
499 phase at a T_h range of 118-142°C with a mode at 126°C (Fig. 10-E). The population of two-phase
500 oil inclusions is petrographically associated with this primary FI population. The second population

501 was observed along clusters or groupings whose association to growth zones could not be robustly
 502 established. This second population also homogenized into the liquid phase but at a lower T_h range
 503 of 94-104°C with a mode of 100°C. Both populations of aqueous inclusions have similar salinities
 504 measured from both the final ice melting temperatures and the Raman spectra. T_{m_i} was recorded
 505 between -7 and -9 °C that corresponds to a range of salinities of 11- to 13 eq. wt.% NaCl
 506 respectively (Fig. 10-F).

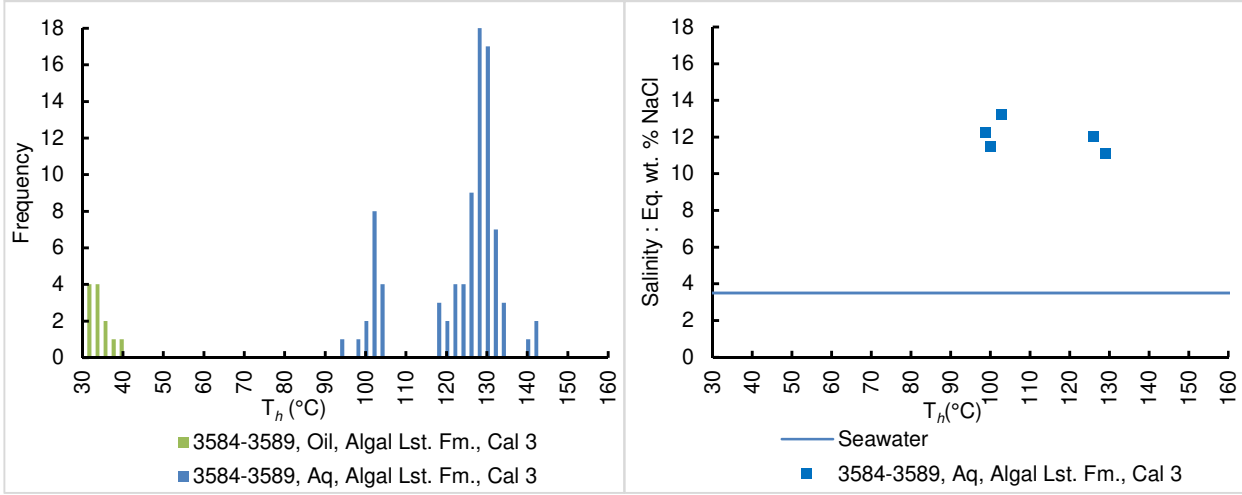
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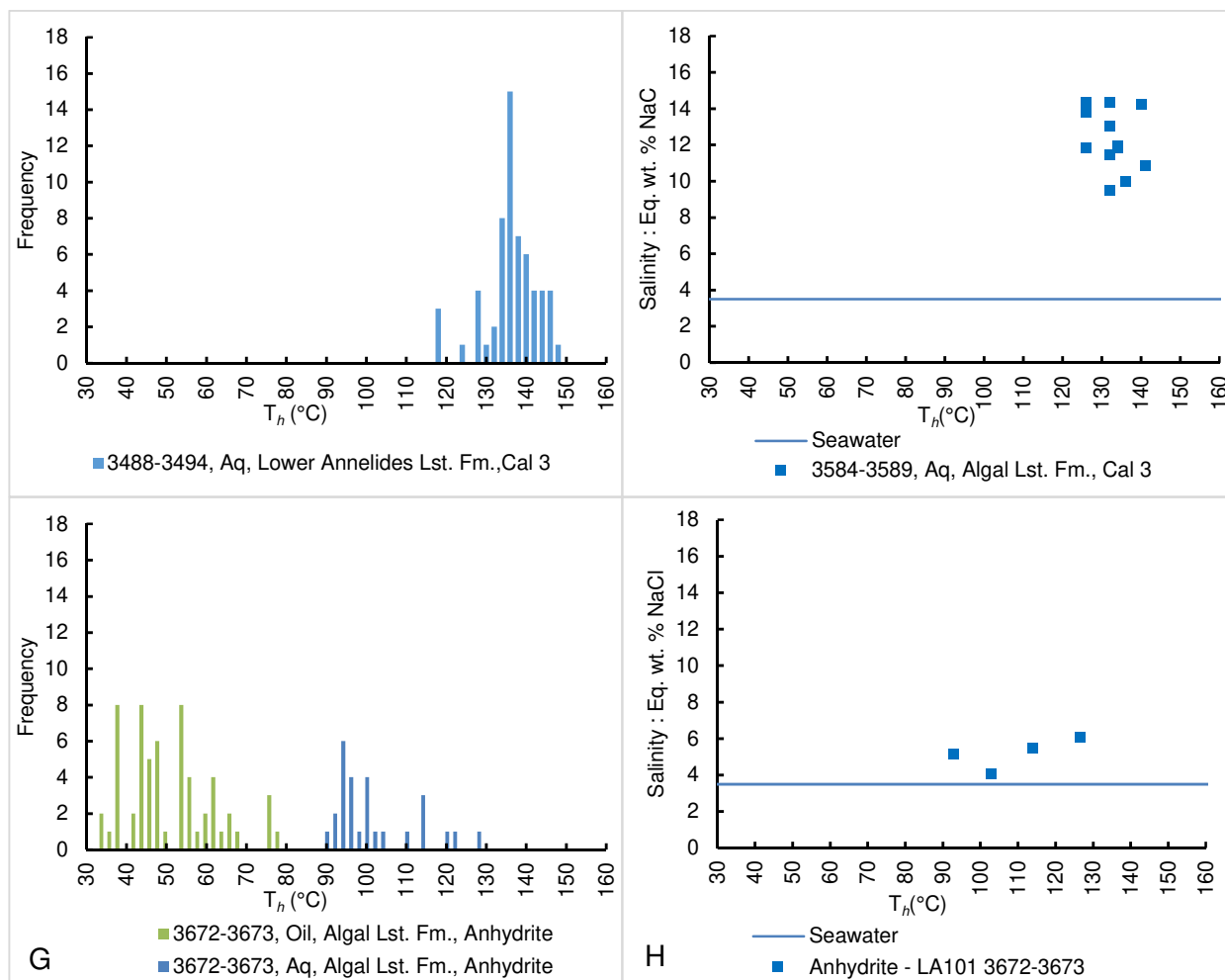
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513 Figure 10. A: Histogram showing homogenization temperatures of aqueous and oil inclusions in the Dol.2 at
 514 a depth of 3707-3715 m in the Mano Dolomite Formation. B: T_h vs. salinity cross-plot of the studied aqueous
 515 FIs in Dol.2. C: Histogram showing homogenization temperatures of aqueous and oil inclusions in the Cal. 3
 516 phase in the Lons formations. D: T_h vs. salinity cross-plot of the studied aqueous FIs in the Cal. 3 phase
 517 of the Lons formations. E: Histogram showing homogenization temperatures of aqueous and oil inclusions in the
 518 Cal. 3 phase in the Algal Limestone formation. F: T_h vs. salinity cross-plot of the studied aqueous FIs in the
 519 Cal. 3 phase of the Algal Limestone formation. G: Histogram showing homogenization temperatures of
 520 aqueous and oil inclusions in the Cal. 3 phase in the Annelids formation. F: T_h vs. salinity cross-plot of the
 521 studied aqueous FIs in the Cal. 3 phase of the Annelids. I: Histogram showing homogenization temperatures
 522 of aqueous and oil inclusions in the anhydrite phase at a depth of 3672-3673m in the Algal Limestone
 523 formation. K: T_h vs. salinity cross-plot of the studied aqueous FIs in the anhydrites.

524 The Cal. 3 sample from the Annelids formation showed only one population of two-phase aqueous
 525 inclusions with no traces of oil inclusions. This ^K population of fluids homogenized into the liquid
 526 phase at a T_h range of 118-148°C with a mode at 136°C (Fig. 10-G). These fluids also have final
 527 ice melting temperatures between -6 and -10°C corresponding to salinity range between 9.5 and
 528 14 wt.% NaCl (Fig. 10-H).

529 The studied anhydrite sample was retrieved from the Algal limestone at a depth of 3672-3673m.
 530 The primary two-phase aqueous inclusions homogenized at a T_h range of 90-128°C with the mode
 531 at 94°C (Fig. 10-I). The oil inclusions homogenized into the liquid phase at a T_h range of 32-78°C
 532 with a mode T_h at 56°C. T_{m_i} could not be recorded with confidence due to repetitive decrepitation

533 and/or darkening of the fluid inclusions. Instead, salinities were obtained from the Raman spectra
534 of the liquid phase which indicated 4 to 6 eq. wt.% NaCl (Fig. 10-J).

535 All measured homogenization temperatures were repeated 3 times by cycling procedures to
536 validate the reproducibility of the recorded T_h . The reproducibility of the all measured T_h was
537 confined to a margin of error of no more than $\pm 1^\circ\text{C}$, which is within the temperature increase of
538 $+2^\circ\text{C}/\text{min}$ used during homogenization.

539 During the freezing cycle of all salinity measurements, and upon gradual heating up from -100°C ,
540 eutectic phase changes were observed at nearly -21.5°C to -22°C , a clear indication of the H_2O -
541 NaCl fluid system (Roedder, 1990).

542 5.3.3 Raman Spectroscopy

543 Raman spectra on the liquid and gas phases of the aqueous FIs in dolomite 2, calcite 3 and
544 anhydrite were analyzed to qualify and quantify the dissolved gases. Measurements were made at
545 the homogenization temperatures in the liquid phase. The detected species include CH_4 and H_2S
546 that are listed in table 2.

547 Table 2. Summary of the average molar fractions of H_2O , H_2S and CH_4 of the aqueous phase in the fluid
548 inclusions. NaCl equivalent mass % were computed according to Bodnar (1993). H_2S values were
549 determined following Jacquemet et al. (2005) while the CH_4 calculated from the Raman spectra following the
550 outline of Caumon et al. (2014).
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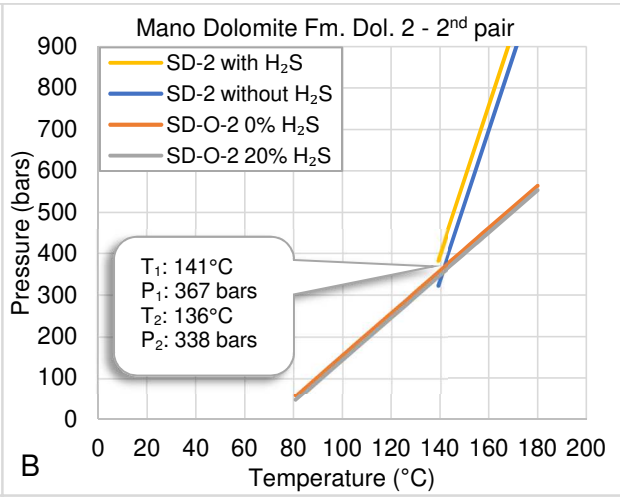
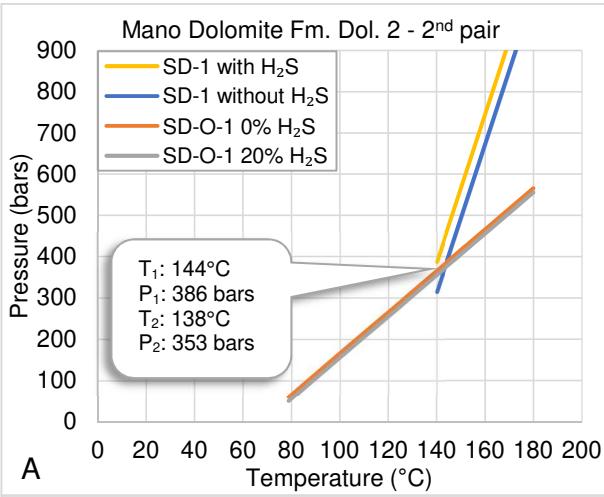
Aq. FIs.	T_h ($^\circ\text{C}$)	Eq. mass %	Mole fractions (%)		
		NaCl	H_2O	H_2S	CH_4
SD-1	139	7.9	94.79	4.90	0.31
SD-2	140	8.24	94.41	5.34	0.25
C-1 (Lons)	107	13	97.38	2.26	0.35
C-1 (Lons)	106	12.7	97.18	2.41	0.40
C-1 (Algal)	126	12	98.06	1.71	0.23
C-2 (Algal)	129	11.1	97.45	2.35	0.19
A-1	94	5.3	98.76	0.97	0.27
A-2	96	5.1	98.81	0.85	0.34

552

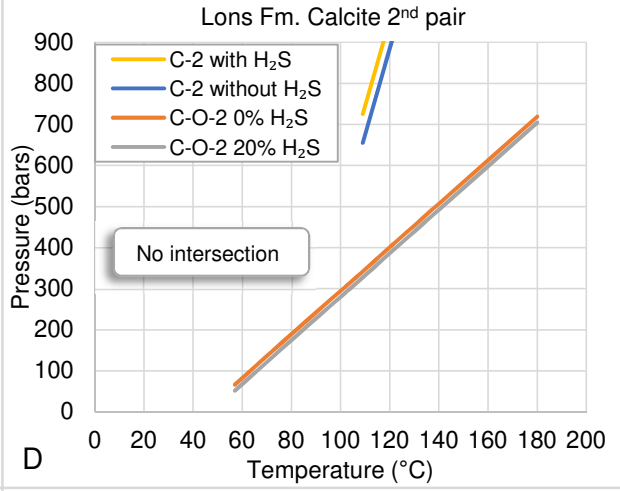
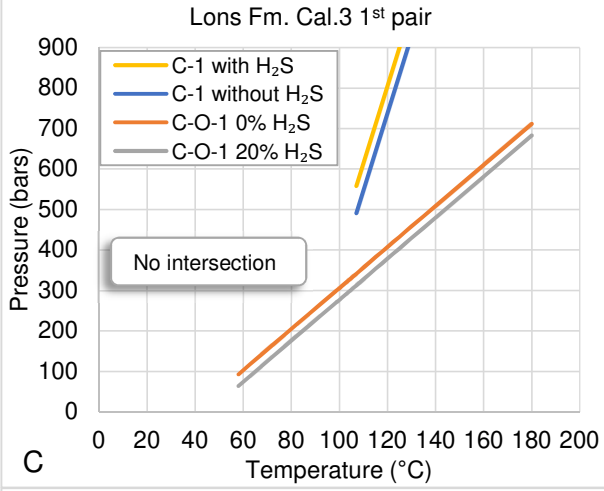
553 5.3.4 Fluid Modeling

554 Aqueous and oil inclusions present as pairs in the same assemblage in a mineral can be tested for
555 AIT-PIT modeling. For this purpose, pairs of aqueous and oil inclusions were selected from Dol. 2,
556 Cal.3 and Anhydrite cement phases whose isochores are plotted in figure 11. The calculation steps
557 are highlighted in section 3.6. As seen in figure 11, the addition of 20% H_2S into the composition of
558 oil inclusions tend to suppress and lower the homogenization pressures.

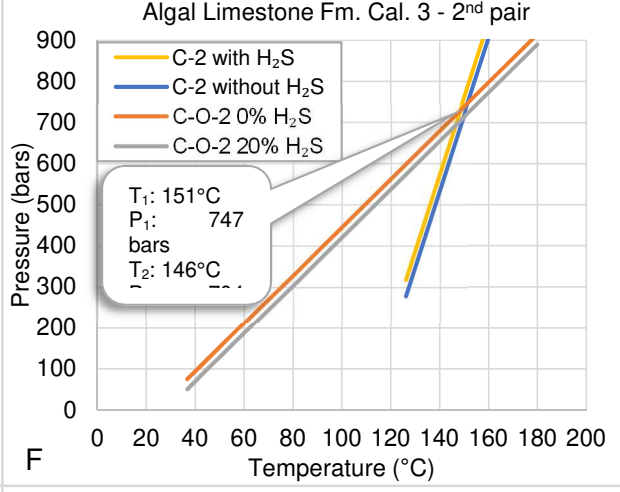
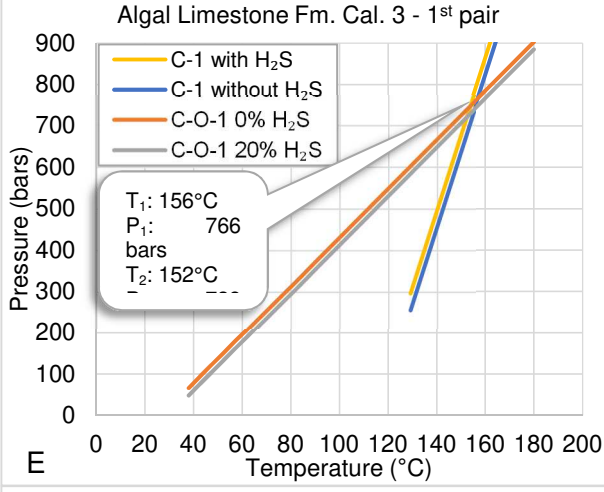
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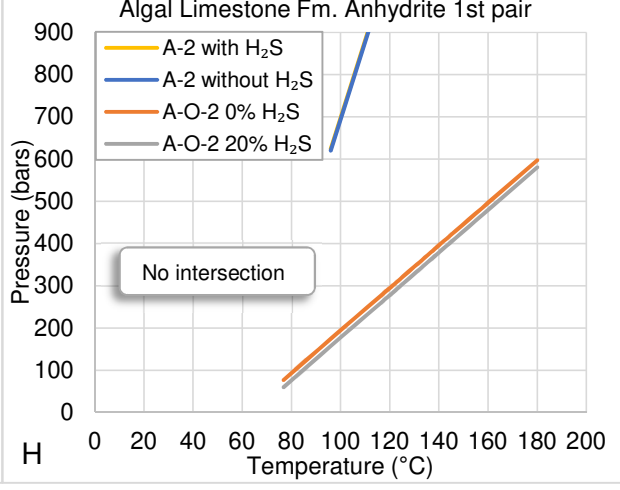
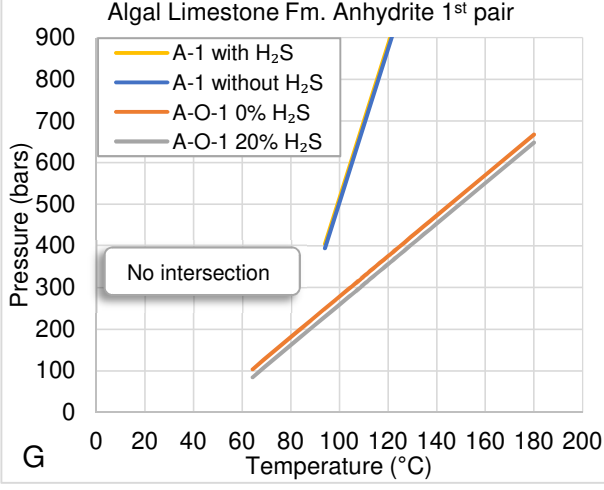
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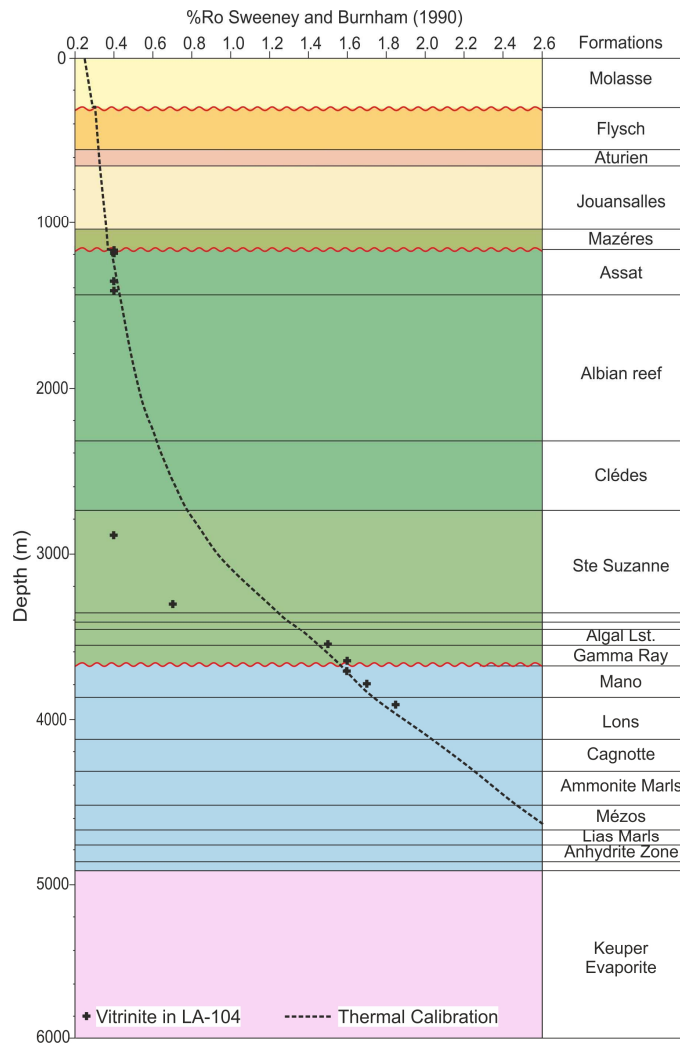
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563 Figure 11. Plots of the aqueous and oil inclusions isochores that are petrographically located within the same
564 assemblages. Location of the selected fluid inclusions are found in figure 9. Plots of A and B show the
565 isochores of aqueous and oil inclusions in the Dol. 2 of the Mano formation. C and D show the isochores of
566 aqueous and oil inclusions of the Cal. 3 phase in the Lons formation Limestone formation. The two sets of
567 isochores do not intersect and hence the trapping P-T conditions could not be determined. E and F represent
568 the isochores of aqueous and oil inclusions in the Cal. 3 phase of the Algal Limestone formation. The two
569 sets of isochores intersect and the trapping P-T conditions are noted in the bubbled boxes. G and H show
570 the two sets of the aqueous and oil inclusions isochores of the anhydrite phase in the Algal Limestone
571 formation, which do not intersect.

572 5.4 Basin Modeling

573 Taking into account realistic amounts of erosion that fit with the geodynamic history of the area, as
574 well as the proper basal heat flows scenarios that fit with the previously published works (e.g. Le
575 Marrec et al., 1995; Renard et al., 2019); the thermal basin model was calibrated as presented in
576 figure 12. The high maturities of vitrinites were reached by increasing the basal heat flow during
577 the Late Jurassic and Albo-Aptian rifting episodes. The amount of eroded material fit with the
578 onlaps and toplaps geometries at the base Cretaceous, base Cenomanian and base Paleocene
579 unconformities (Fig. 3) with 400m eroded at the top of Flysch formation, 200m at the top of the
580 Assat marls and 50 meters at the top of the Mano Formation. During the climax of the Aptian
581 rifting, the heat flow reached 130mW/m^2 (Renard et al., 2019). Previous works showed that this
582 thermal peak did not cool quickly but took at least 20 to 30 Ma before decreasing again (Clerc et
583 al., 2015; Salardon et al., 2017; Angrand et al., 2018). This was taken into account such that the
584 peak basal heat flow reached 110 mW/m^2 at 120 Ma and took 20 Ma to decrease to 85 mW/m^2 .
585 The present-day heat flow of 45 mW/m^2 is equivalent to a geothermal gradient in Lacq close to
586 30°C/km (Le Marrec et al., 1995 and Renard et al., 2019).



587
588
589

Figure 12. Thermal maturity calibration of the basin model.

590 6 Discussions

591 6.1 Fluid modeling

592 Figure 9 gives strong petrographic evidence of primary aqueous and oil inclusions entrapment in
 593 the studied phases. The location of fluids between growth zones or in clusters limited to individual
 594 crystals indicate a primary origin of the inclusions. Also, microthermometry results provide
 595 additional supporting evidence that can complement petrographic observations. Figure 10 shows
 596 different homogenization temperatures and salinities between the different cement phases. This
 597 difference indicates distinct fluid events, that trapped inclusions between growth zones and in
 598 clusters within individual crystals, and not as secondary fluid events that would cross-cut different
 599 cements with identical temperatures and salinities. The risk of having post-entrapment alteration
 600 and necking down was also checked. For this purpose, several lines of evidence are available to
 601 elucidate such risks or doubts. Firstly, no petrographic evidence were found to support any necking
 602 down or shape-modification of the studied fluid inclusions. There are no single-phase aqueous

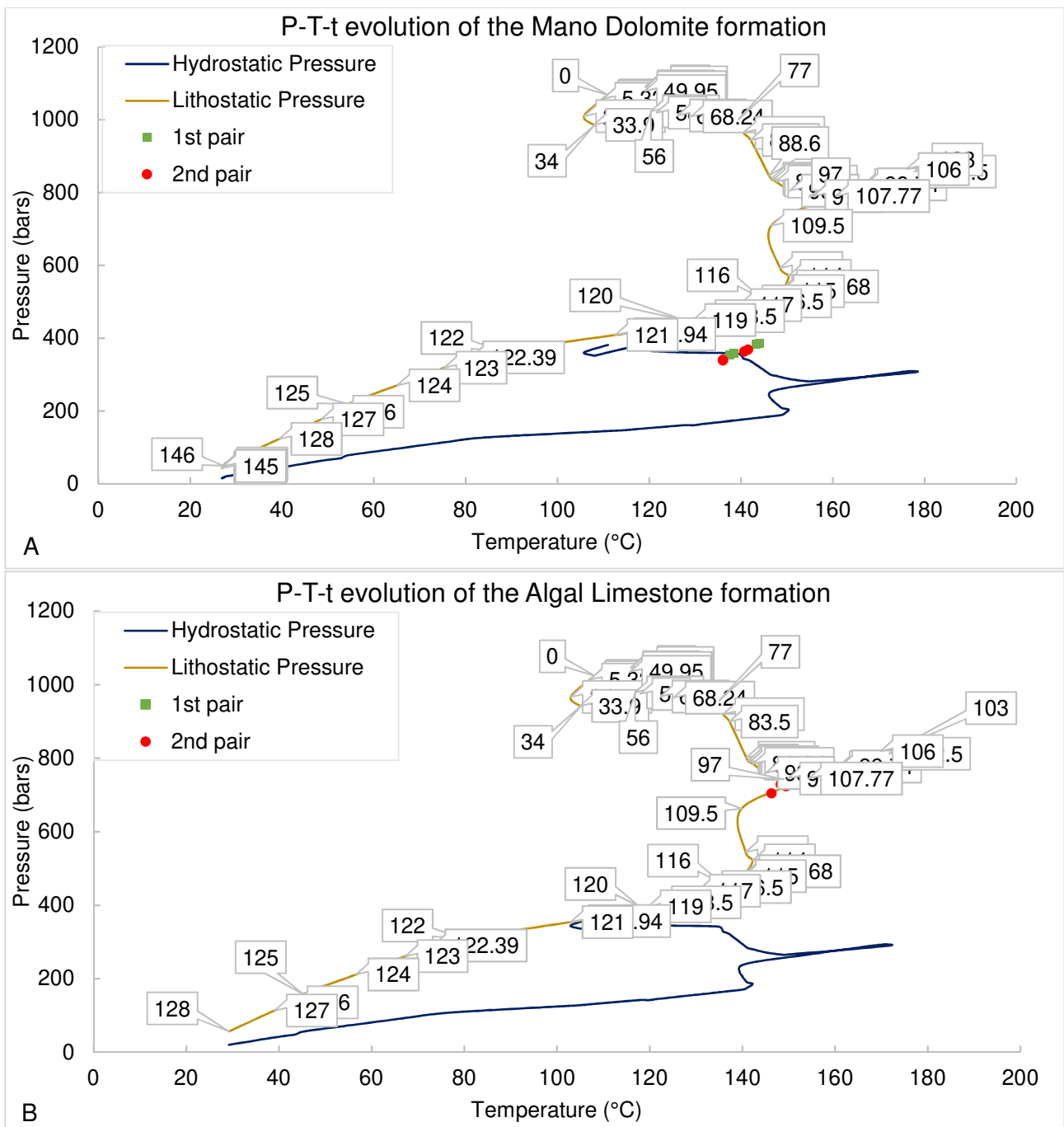
603 inclusions alongside an inclusion dominated by the vapor phase, typical of a necked-down
604 inclusions. Secondly, the L/V ratio, Fv%, is consistent and stable across all studied FIs in the
605 different cement phases. Necking down or thermal resetting would result in varying Fv%. Thirdly,
606 the shape of the histogram shows a confined Gaussian distribution of frequencies without any
607 high-temperature “stretching” or thermal re-equilibration trends. The confinement of the T_h
608 measurements to limited ranges of nearly 10°-15°C variability is still a very limited range giving
609 such a diagenetic system at depths of 3.5-4 km. Finally, the good reproducibility of the
610 homogenization temperatures at room conditions, where the pressure difference is the highest
611 between the internal fluid pressure and the external atmospheric pressure, is another indication
612 that the studied cement phases are not fragile and did not undergo post-entrapment alteration.
613 Another line of evidence that supports the absence of post-entrapment alteration is the near
614 absence of decrepitation of fluid inclusions, which further supports the good reproducibility of
615 obtained T_h and the absence of post-entrapment P-T modifications. Therefore, after validating the
616 primary nature of the fluid inclusions and confirming the absence of post-entrapment alteration,
617 pairs of aqueous and oil inclusions were selected for modeling. The selection criteria focused on
618 their petrographic location between growth zones or in clusters confined to individual crystals and
619 not extending to nearby cements. The pairs of aqueous and oil inclusions were also selected to
620 represent the most common homogenization temperatures, i.e. the modes, of the FI populations.
621 Following these criteria and the modeling procedure described in the paragraph 3.6, the aqueous
622 and oil inclusions observed in the Dol. 2 phase were modeled and indicate trapping temperatures
623 of 141-144°C and pressures between 367 and 386 bars. In the case of the Cal. 3 phase, at least
624 three fluid pulses are recognized. The first fluid pulse was a mixture of aqueous and oil fluids
625 trapped contemporaneously in the calcite cement. These fluids are documented in figure 11 – E
626 and F where their isochores intersect at high-pressure conditions. The two other fluid pulses are
627 evidenced by the aqueous and oil inclusions which were not trapped simultaneously as seen in
628 figure 11 – C and D. The anhydrite phase is also a witness of multiple fluid pulses, as the
629 isochores of the aqueous and oil inclusion trapped in it do not intersect (Figure 11- G and H).
630 Modeling of trapping P-T conditions of fluid inclusions should be evaluated relative to the
631 uncertainties and limitations of the used techniques. In this study, the error range of the T_h
632 microthermometry measurements is in the order of $\pm 1^\circ\text{C}$ and for T_{m_i} around $\pm 0.1^\circ\text{C}$. Quantification

633 errors of gases in the fluid inclusions using Raman spectroscopy should be less than 0.1 mol% as
634 the treatment and calculation steps were repeated several times to avoid any biased baseline
635 treatment of the Raman spectra. Finally, the processing of the 3-D image stack for the oil inclusion
636 modeling and the quantification of the vapor volume fraction by the confocal scanning laser
637 microscope could impose an error of no more than $\pm 2\%$. Combining all these error ranges and
638 extrapolating them to the trapping P-T estimation, a possible error range of ± 15 bars and $\pm 5^\circ\text{C}$ is
639 suspected. This uncertainty is deemed acceptable for diagenetic systems whose P-T conditions
640 range from 100-1500 bars and 40-250°C (Pironon, 2020).

641 6.2 Ages of the diagenetic fluids

642 To obtain age constraints on the diagenetic fluids, hydrostatic and lithostatic pressures obtained
643 from the basin model done with Petromod 1D are plotted on P-T diagram from the time the
644 formation was deposited until present day (Fig. 13). The spike in temperature between 116 and 97
645 Ma refers to the high thermal anomaly associated to the Apto-Albian rifting stage. On the same
646 figure, the trapping P-T conditions obtained from the double-isochore plots in figure 11 are also
647 plotted. This step puts the conditions of the trapped fluids in the framework of the P-T evolution of
648 their hosting formation and thus the timing of the trapping can be deduced. As seen in figure 13-A,
649 the two pairs of fluid inclusions trapped in the Dol. 2 intersect the P-T domain of the Mano
650 Dolomite formation at two age ranges: 120-116 Ma (Aptian) and 81-68 Ma (Campanian-
651 Maastrichtian). The difference between the fluid pressure and the hydrostatic pressure amount to
652 around 175 bars of overpressure if the 81-68 Ma age is considered. Similarly, fluids trapped in the
653 Cal. 3 phase intersect the P-T domain of the hosting Algal Limestone formation between 109.5 and
654 97 Ma (Albian – Cenomanian) (Fig. 13-B). These fluids have pressures closed to the lithostatic
655 pressure gradient, amounting to around 480 bars of overpressure relative to hydrostatic conditions.
656 Given the petrographic evidence that clearly places the Cal. 3 phase as paragenetically younger
657 than the saddle dolomites (Fig. 4, 6 – B and D), the Aptian age of the saddle dolomites is more
658 realistic than the Campanian-Maastrichtian. This Aptian age is further supported by the regional
659 dolomitization trend during the Aptian observed in the Rouse field (Renard et al., 2019) and in the
660 Chaînons Béarnais (Motte et al., 2021; Salardon et al., 2017).

661 It was not possible to place an age constraint on the fluids trapped in the Anhydrite phase,
 662 because the aqueous and oil inclusions are not co-genetic and were not trapped simultaneously.
 663 Nonetheless, it is still possible to approximate their relative timing of precipitation by petrographic
 664 evidence. Figure 7-A shows a stylolite, tectonic in origin, cross-cutting an anhydrite-filled fossil-
 665 moldic pore. These tectonic stylolites were likely formed as a result of the maximum compression
 666 during the Eocene. Also, figure 7-C shows Cal.3 being partially replaced by the engulfing Anhydrite
 667 cement. Hence, the anhydrites have an age younger than the Cal. 3 phase and before the Eocene
 668 maximum compression, thus the Late Cretaceous age.



670
 671 Figure 13. Evolution of the hydrostatic and lithostatic pressures deduced from Petromod 1D for the (A) Mano
 672 dolomite formation and Algal Limestone Formation (B). Number tags are ages in Ma. The green and red

673 points correspond to the trapping P-T conditions of the fluids determined from the intersection of the
 674 aqueous and oil inclusions isochores in figure 11.

675 Uncertainties about these age estimations are related mainly to the thermal model calibrated with
 676 the vitrinite data (Fig. 12). In the Mano, Algal and Lons formation, the modelled maturity is
 677 compatible with that recorded from the vitrinite data and fits well the vitrinite maturities. The misfit
 678 with the vitrinite data in the Sainte Suzanne marls is likely linked to the thermal blanket effect that
 679 these marls have, which isolated the vitrinites and prevented their higher maturities. The clays,
 680 shales and marls that constitute the Sainte Suzanne formation are characterized by significantly
 681 low thermal conductivities in the range of 0.2-1.0 Wm⁻¹/°C, almost half the conductivities in the
 682 surrounding media (Blackwell and Steele, 1989; Pollack and Cercone, 1994; Nunn and Lin, 2002;
 683 Clerc et al., 2015; Salardon et al., 2017).

684 6.3 Diagenetic model of Deep Lacq

685 A paragenetic sequence of diagenetic phases and events is proposed in figure 14.



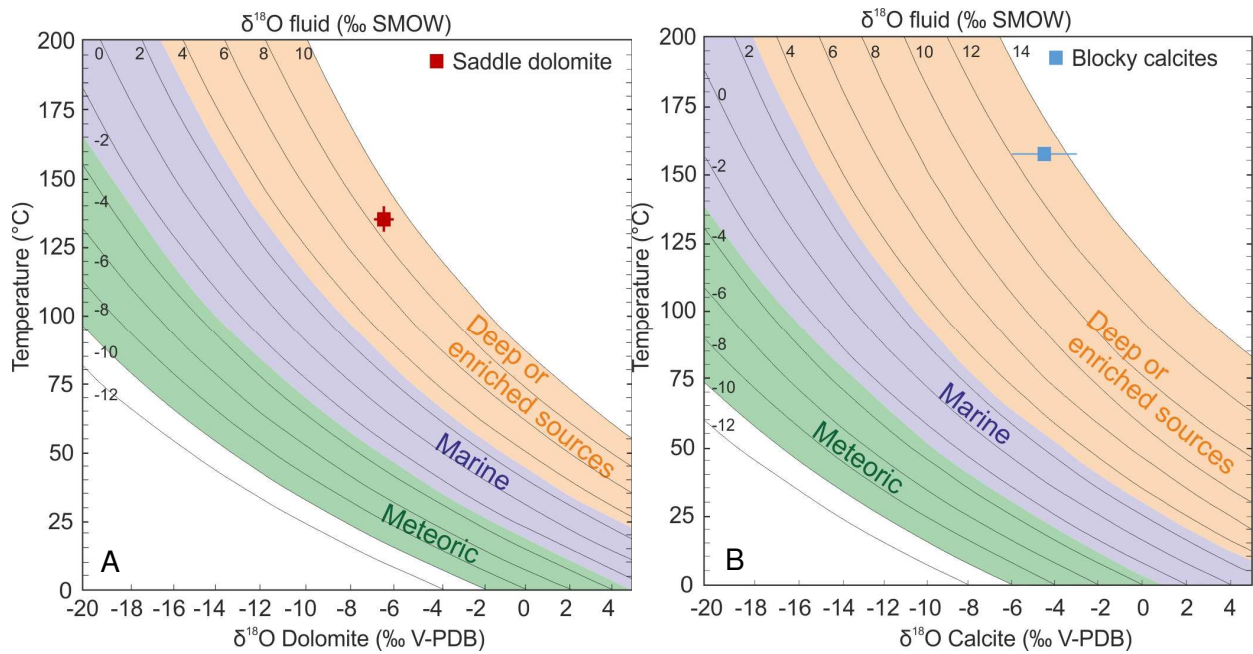
686
 687 Figure 14. Paragenesis of the Lacq gas field. T_t and P_t correspond to the trapping T-P conditions
 688 respectively as deduced from fluid modeling. T_h is the homogenization temperature. The evolution of H₂S
 689 mole % is also added.

690 The chalky mudstone, bioclastic wackestone to packstone, as well as the peloidal boundstone
 691 with microforaminifera, bivalves, mollusks, sponges, echinoderms and annelids are indicative of a
 692 shallow marine depositional environment, in a lagoon to reefal setting. This is also supported by

693 previous paleo-environmental reconstructions of Le Marrec et al. (1995) and Biteau et al. (2006).
694 $\delta^{18}\text{O}$ isotopes of the bulk host rocks (Fig. 8) are depleted relative to the late Jurassic seawater
695 (Veizer et al., 1999). This signature is typical of micrites which experienced crystal overgrowths at
696 higher temperatures during burial (Carpentier et al., 2014). The high-temperature impact is
697 supported by the orange CCL colors of the micritic host rocks (Figs. 4-E and 5-B,C,D) which are
698 indicative of the expected reducing conditions during burial diagenesis (Flügel, 2004). Negative
699 $\delta^{13}\text{C}$ values could be due to a possible contribution of organic matter hosted in the chalky matrix
700 (Jorgensen, 1987). Micrite rims and burrows engulfing and penetrating bioclasts (Fig. 4-A) are
701 evidences of micro-borers bacterial activity (e.g. micro-borers). The bacterial activity was
702 responsible of the precipitation of pyrite 1 phase since the framboidal clusters and their rounded to
703 anhedral shapes are typical by-products of bacterial sulfate reduction (BSR) (Machel, 2001). The
704 preferential distribution of framboidal pyrite in shells impacted by micritization is in accordance with
705 BSR, controlled by the local abundance of organic matter. Dolomite 1 is relatively early in the
706 diagenetic history (Fig. 14). It is probably the product of marine dolomitization in a shallow lagoonal
707 environment. This is partly supported by the dark core of the dolomite crystals (Fig. 4-C) and by
708 the shallow marine conditions that prevailed throughout the Jurassic and Cretaceous (Biteau et al.,
709 2006). These early dolomite phases (D1) can be considered as the analogs of the replacive
710 dolomites (RD1, RD2) that precipitated in near-surface to shallow burial environments found in the
711 Chaînons Béarnais with cretaceous ages (141-125 Ma) (Motte et al., 2021). The very small crystal
712 sizes and possible contamination with the host matrix prevented reliable collection of powders for
713 O and C isotopes analysis to further support the dolomitization model. At slightly deeper levels
714 within the meteoric eogenetic realm, local circulation in the meteoric phreatic zone, characterized
715 by fluctuations in redox conditions, precipitated the dog-tooth calcite 1 phase. The scalenohedral
716 morphology of crystals, their isopachous distribution around grains, and their zoning in CCL (Fig.
717 4-E) reflecting the fluctuations in redox states (Machel et al., 1991) support the precipitation in the
718 deep meteoric phreatic zone (Flügel, 2004; Moore, 2013). Transition to mesogenesis (Fig. 14) was
719 marked by the dominance of the reducing conditions in the calcite 2 phase with diminishing
720 influence of the percolating meteoric fluids. Calcite 1 and 2 are only observed at the level of the
721 Annelids formation and are considered as products of local pools of meteoric fluids possibly
722 supplied by the local surfaces of subaerial erosion at the Base Cretaceous unconformity (Fig. 2).

723 As the overburden pressure increased, mechanical compaction started and then evolved to
724 chemical with the development of stylolites 1 (Fig. 5-A). Since the samples were collected from
725 vertical wells that cut perpendicularly nearly horizontal strata, the orientation of the stylolite 1
726 planes, being nearly horizontal, allows its classification as bedding-parallel stylolites (BPS).
727 Stylolite 1 are cross-cut by fracture A. These fractures were initially lined by dolomite 2 before
728 being partially replaced by later calcites (Fig. 5-B). The relative $\delta^{18}\text{O}$ depletion ($\sim -6.2\%$ PDB) of
729 dolomite 2 is due to fractionation at elevated temperatures (O'Neil et al., 1969). The presence of
730 H_2S in the aqueous inclusions (Table 2) is likely generated by Thermochemical Sulfate Reduction
731 (TSR) as this process is well-documented in the area and is assumed to be responsible of the high
732 H_2S concentration (up to 15% mol) of the Lacq gas field (Connan and Lacrampe-Couloume, 1993).
733 The H_2S presence in these fluid inclusions suggests that TSR was already active when saddle
734 dolomites precipitated. However, no evidence of *in-situ* TSR is observed here, implicating
735 migration of TSR-derived fluids into the Deep Lacq reservoir. Saddle dolomites are mostly limited
736 to fractures and represent a typical burial cement whose Mg source was likely the dissolution and
737 recrystallization of the early dolomites as a result of pH lowering during the first stages of organic
738 matter maturation (Moore, 2013). Such a scenario can justify the elevated $\delta^{18}\text{O}$ values calculated
739 for the parent fluid of dolomite of 8.1-10.1 ‰SMOW (Fig. 15-B). The dissolution of low temperature
740 carbonates in semi-closed conditions is an efficient way to enrich the $\delta^{18}\text{O}$ of parent fluids and
741 provide a suitable Mg source. The partially dissolved and recrystallized early dolomites, with their
742 corroded rims and/or cores (Fig. 4-C) provide direct evidence of this Mg source.

743 The high salinities and enriched $\delta^{18}\text{O}$ values for the dolomite-forming fluids, indicate a mixture
744 between the Triassic evaporites at the core of the Lacq anticline (Fig. 3) and the recrystallization of
745 earlier dolomites during the Aptian hyperextension. Mg derived from a marine source influenced by
746 high-temperature serpentization reactions (Scribano et al., 2017) could also be possible given the
747 geodynamic context that was impacted by the thermal anomaly during the Aptian hyperextension.
748 This dolomite phase (Dol. 2) is similarly documented by Motte et al. (2021) as phases DC2, DC3
749 and DC4 in the Chaînons Béarnais, and as Dol 2 and Dol 3 by Renard et al. (2019) in the Rouse
750 field.



751

752 Figure 15. Fractionation curves of Horita (2014) to determine the δ¹⁸O (SMOW) values of the parent fluids
 753 that precipitated the saddle dolomite phase.

754 Oil has likely migrated to the Deep Lacq reservoir during the precipitation of Dol. 2 as suggested
 755 by the primary oil inclusions. Source rock maturity modeling indicate that the entrance to the oil
 756 windows, with vitrinite reflectance of ~0.6% PRv (oil) has occurred during the Lower Cretaceous on
 757 the crest and the NE limb of the Lacq structure (Sejourné, 1995). Such a model can explain the
 758 coeval entrapment of oil and aqueous inclusions in Dol. 2 and is in accordance with the timing
 759 suggested in figure 13-A.

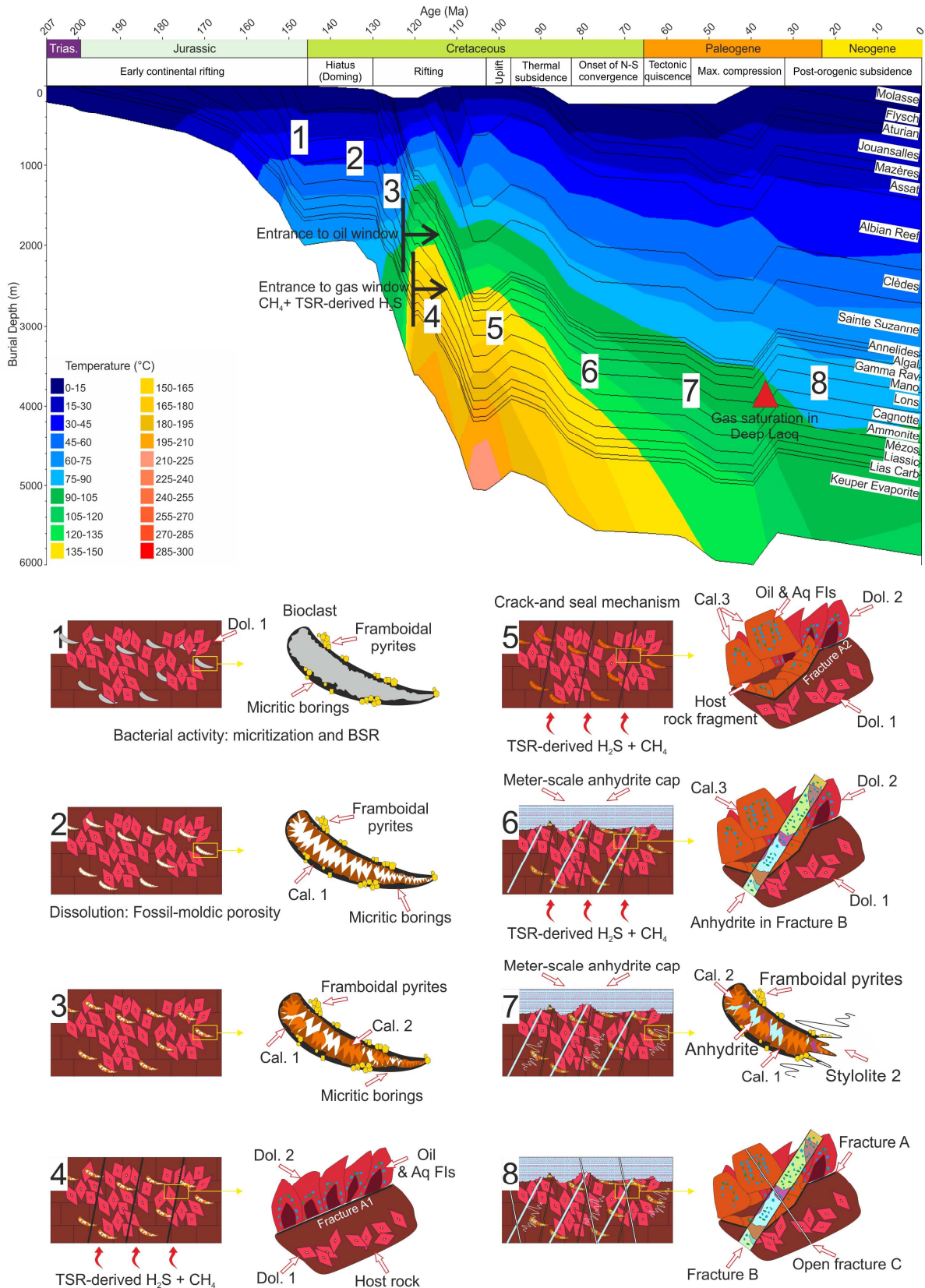
760 The Albian-Cenomanian age of the calcites correspond to a phase of post-rift uplift and thermal re-
 761 equilibration (Masini et al., 2014; Ford et al., 2016). This change in tectonic activity triggered a new
 762 fracturing event (A2) that reopened fracture set A1 (Fig. 5-B) and allowed Ca-rich fluids to
 763 circulate. The dull orange to yellow luminescence suggests a precipitation under reducing
 764 conditions (Machel, 1985). The depletion in δ¹⁸O (-2.8 and -5.63‰PDB) (Fig. 8) is also due to
 765 fractionation at elevated temperatures (Horita, 2014). The δ¹³C of the Cal.3 is in the same domain
 766 as that of the Dol. 2 and represents typical signatures of burial cements whose fluids have been
 767 buffered by the surrounding media (Moore, 2013). The δ¹⁸O of the parent fluids that precipitated
 768 the calcite 3 is enriched and reach +8.7 ‰ SMOW (Fig. 15-A). This enrichment is indicative of a
 769 contribution of a deep source of fluids. The most likely scenario is the depletion of Mg in the
 770 Triassic brine that have precipitated the saddle dolomites leaving behind fluids enriched in δ¹⁸O,
 771 depleted in Mg and relatively saturated in Ca. The similar salinities between the fluids trapped in

772 the saddle dolomites and the Cal. 3 (Fig. 10 – B, D, F and H) provide strong evidence to the
773 continued leaching of the Triassic saline source, but with the cessation of supply of mantellic Mg
774 during this phase.

775 After calcite cementation, the massive anhydrites precipitated along the fracture set C. Multiple
776 sulfur isotopes values in this anhydrite phase ($\delta^{34}\text{S}$ between 18.97 and 19.12‰ and $\Delta^{33}\text{S}$ between
777 0.017 and 0.023‰) are consistent with the overall Keuper seawater signature recorded in CAS at
778 $\delta^{34}\text{S} = 20.6\text{‰} \pm 5.4$ and $\Delta^{33}\text{S} = 0.000\text{‰} \pm 0.025$ (Wu et al., 2010 – 2014). This further supports a
779 direct link between Deep Lacq and the underlying Triassic evaporites at the core of the Lacq
780 Anticline (Fig. 3). The structural link between the Triassic core and the Jurassic reservoirs of Deep
781 Lacq was made possible through the major thrust fault that separates the Southern Lacq Basin
782 from the Lacq Anticline, as well as the extensional fractures on top of the Triassic core that cut
783 through the Jurassic formations (Fig. 3). The precipitation of anhydrites has most likely occurred
784 during the onset of the N-S compression which allowed the upward circulation of dissolved sulfates
785 by pressure gradients within the anticline, before being intercepted by the impermeable formation
786 of the Sainte Suzanne Marls. This circulation scheme explains the massive precipitation of
787 anhydrite just a few meters below this formation (Fig. 6-A).

788 The infilling and supply of gases to the Deep Lacq reservoir was likely a gradual process and not a
789 sudden pulse. This is witnessed by the fact that dissolved gases were trapped in the fluid
790 inclusions of the Dol.2, Cal. 3 and anhydrite phases (table 2). Basin modeling done in this work,
791 supported by Sejourné (1995), indicates an entrance to oil window and, shortly afterwards the gas
792 window, during the Aptian thermal anomaly associated to the hyperextension event. The H_2S
793 quantity was highest in the Aptian Dol. 2 and decreased gradually to the Paleogene anhydrites.
794 The difference in timing between fluid circulation, mineral precipitation and gas charging is the
795 reason for the absence of gas inclusions in the studied phases. Basin modeling done in this work
796 and supported by Michels et al. (2021) confirm the rapid maturity of organic matter and cracking
797 from oil to gas since the Cretaceous hyperextension, but it is only recently, during the Paleogene
798 compression, that the main gas infilling of the reservoir occurred. This is mainly due to the lack of
799 pathways between the source and reservoirs rocks, which was later made possible via the fracture
800 network formed during the Paleogene orogeny (Fig. 16). The fact that no secondary gas inclusions

801 were observed is supported by the fact that the petroleum system entered into the dry gas window
 802 without any aqueous phase to aid in the fracture-healing processes and fluid entrapment
 803 An illustrative summary of the main geodynamic events and associated diagenetic modifications is
 804 compiled in figure 16.



805
 806 Figure 16. Summary of the main geodynamic, diagenetic and fluid events relative to the burial history of
 807 Deep Lacq. Colors of the schematic diagrams are based on the CCL petrography.

809 This work highlights the power of applying a multi-modeling approach to reveal the diagenetic
810 history in a complex geodynamic setting. After a detailed petrographic analysis of all the phases
811 present, the multitude of equations of states used in AIT-PIT modeling provided valuable insight
812 into the fluids P-T conditions. These in turn were then validated against the basin history to place
813 age constraints on the timing of fluids circulation.

814 The diagenesis of the giant Deep Lacq reservoir has been dependent on a geological heritage with
815 two main events: a rifting phase and a convergence phase. Before these two major events,
816 diagenesis was insignificant and limited mostly to bacterial activity as revealed by micritization and
817 BSR-related framboidal pyrites.

818 The Aptian rifting and hyperextension enabled the circulation of hot deeply sourced fluids enriched
819 in Mg, which triggered a limited event of fracture-related hydrothermal dolomitization. Fractures
820 connected Deep Lacq to the Triassic salt core, which provided Mg^{2+} and NaCl to precipitate the
821 saddle dolomites (Dol. 2). Additional Mg for dolomitization was provided by the dissolution and
822 recrystallization of early dolomites. The connection to a serpentine-related source of fluids remains
823 a likely scenario given the geodynamic context and the mantle exhumation event during the Aptian
824 hyperextension. At the same time, the source rock entered the oil window and shortly afterwards
825 the gas window, TSR started and gas and oil started to migrate. The most likely kitchen for TSR
826 reactions is the deeper South Lacq Basin and not *in-situ* within the Lacq field as no evidence of
827 local TSR were recognized in all the studied samples.

828 Continued fracturing events resulted in the entrapment of multiple populations of aqueous and oil
829 inclusions in the Cal. 3 phase. This phase is a major pore-reducing cement that precipitated during
830 Albian to Cenomanian, at the time when the basin experienced post-rift doming and thermal
831 relaxation.

832 With the onset of the Pyrenean orogeny and the associated structural reconfiguration, new
833 pathways and fractures introduced sulphate-rich fluids that precipitated anhydrite cements. The
834 upper age limit was constrained by the Eocene compression tectonic stylolites that crosscut the
835 anhydrites. During the Pyrenean orogeny, the gas reservoirs of Deep Lacq have likely reached
836 their saturation capacity of dry gas that prevented further diagenetic cements to precipitate.

837 In this Jurassic reservoir, the number of diagenetic events appears to be less than in the
838 “Chaînons Béarnais”, a well-known zone where the absence of hydrocarbon trapping favors a
839 longer recording of the diagenetic history.

840 The multitude of analytical approaches used in this research opens new perspectives to unravel
841 the infilling and preservation history of petroleum reservoirs. Several basins have undergone a
842 complex diagenetic history due to their geodynamic context during a complete or incomplete
843 Wilson cycle. The combination of techniques used here is a key to deciphering the real timing, fluid
844 composition and P-T path of fluid circulations in such complex areas.

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