Adsorption of Bisphenol A on KOH-activated tyre pyrolysis char

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Supplementary information

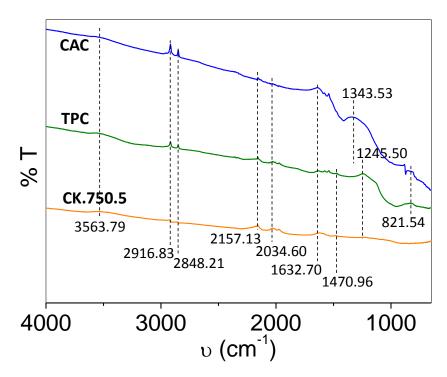
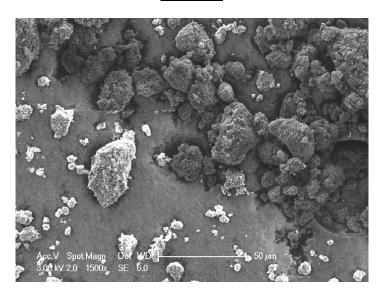


Figure S1 FTIR spectra for TPC, CAC and CK.750.5

The FT-IR spectra presented bands between 4000 and 650 cm⁻¹ that can be attributed to different chemical bond vibrations. The broad bands located between 3563.79 and 2916 cm⁻¹ are related to stretching of O–H bonds, asymmetrical and symmetrical stretching of C_{sp2}–H bonds, stretching of C_{Ar}–H and asymmetrical stretching of C–H in methylene groups. The bands located between 2916.83 and 2848.21 cm⁻¹ are assigned to symmetrical stretching of C_{sp3}–H bonds and asymmetrical stretching of C–H in methylene groups. The bands found between 2157 and 1632.70 cm⁻¹ can be assigned to stretching of C=C and C=O bonds, and those between 1632.70 and 1470.96 cm⁻¹ to stretching of C–O bonds in –COOH groups, C=C stretching and C–H bending. The intense band located at 1632.70–1343.53 cm⁻¹, together with that at 3563.79 and 2916 cm⁻¹, found for TPC and CAC may indicate the presence of abundant –COOH and alcohol groups. The bands appearing in the range of 1470.96–1245.50 cm⁻¹ can be related to C–H bending and C–C stretching vibrations, while those appearing below 1245.50 cm⁻¹ indicate the presence of C–C in CH₂ groups and carbon

linked to oxygen in C–O–C epoxy and C–O containing-groups. TPC and CAC exhibited intense bands in the region of 1245.50–821.54 cm⁻¹, which agrees with the presence of – COOH, alcohol and/or epoxy groups in these samples

<u>CK.750.5</u>



<u>CAC</u>

