

Cross-frontal mode: An alternative methodology for Taylor dispersion analysis of monomodal sample

Jérémie Gouyon, Ariane Boudier, Arnaud Pallotta, Célia Boukoufi, Igor Clarot

▶ To cite this version:

Jérémie Gouyon, Ariane Boudier, Arnaud Pallotta, Célia Boukoufi, Igor Clarot. Cross-frontal mode: An alternative methodology for Taylor dispersion analysis of monomodal sample. Journal of Chromatography A, 2023, 1694, pp.463913. 10.1016/j.chroma.2023.463913 . hal-04026838

HAL Id: hal-04026838 https://hal.univ-lorraine.fr/hal-04026838v1

Submitted on 4 Oct 2024

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution 4.0 International License

1	Cross-frontal mode: an alternative methodology for Taylor
2	Dispersion Analysis of monomodal sample
3	Jérémie Gouyon ^a *, Ariane Boudier ^{a,b} , Arnaud Pallotta ^{a,b} , Célia Boukoufi ^{a,c} ,Igor
4	Clarot ^{a,b}
5	
6	^a Université de Lorraine, CITHEFOR, F-54000 Nancy, France
7	^b Nanocontrol, F-54000 Nancy, France
8	^c Pharmacy Department, University Hospital, 54511 Vandœuvre-lès-Nancy, France
9	
10	*Corresponding author: jeremie.gouyon@univ-lorraine.fr (ORCiD number: 0000-0002-
11	8834-9009), +33 3 72 74 73 10
12	Keywords
13	Capillary, Hydrodynamic radius, Methodology, Taylor Dispersion Analysis

14 Abstract

15 Taylor dispersion analysis (TDA) is a technique dedicated to the determination of the 16 molecular diffusion coefficient (D) of species, using band broadening of an analyte in 17 a laminar flow. Two modes are commonly used to perform TDA: pulse and frontal 18 modes. In each case, a fitting of the signal is required. We propose here a third mode 19 denoted as cross-frontal mode, combining two crossed sample fronts without 20 modification of a classical CE device for the rapid and accurate determination of D of 21 caffeine, reduced glutathione (GSH), insulin from bovine pancreas, bovine serum 22 albumin (BSA) and citrate-capped gold nanoparticles (AuNP). Theoretical aspects and 23 methodology are described, showing a good correlation between the so-called cross-24 frontal mode and usual frontal mode. Limitations of the techniques are also assessed, 25 and are similar to regular modes while no fitting is required. This new methodology 26 allows improving the sensitivity toward low concentrated sample compared to pulse 27 mode, and an alternative mathematical treatment compared to regular TDA modes.

28 **1. Introduction**

29 Taylor dispersion analysis (TDA) is known as a characterization technique for size 30 determination in the nm range (0.1 – 100 nm)[1]. It has been used for a wide variety of 31 analytes size, from small organic molecules[2] to biomolecules characterization[3] or 32 inorganic nanoparticle size determination[4]. Using capillary electrophoresis (CE) 33 instrument[5], TDA is easily accessible and complemental to other characterization 34 techniques such as Dynamic Light Scattering (DLS) or Transmission Electronic 35 Microscopy (TEM)[6]. Basically, TDA consists in the measurement of the peak broadening of a solute injected in a laminar Poiseuille's flow for sub-nm size range 36 37 determination. Peak width is, in this particular case, inversely proportional to the 38 diffusion coefficients of the analyte. Such a laminar flow is feasible with unmodified CE 39 devices[7] due to capillary dimensions, pressure control and UV-Visible detector 40 coupling.

Two modes are nowadays commonly used for Taylor dispersion: pulse and frontal 41 42 modes[8]. The pulse mode consists in the injection of a finite plug of sample before 43 mobilization through the capillary[9] (typically, the plug is inferior to 1% of the volume 44 of the capillary between the inlet and the detection window). The resulting Taylorgrams being described by a Gaussian curve, the peak width is linked to the analyte size. 45 46 Nonetheless, the sensitivity of the pulse mode is limited (due to low sample volume), 47 which can be counteracted using different types of detectors[10,11], or using the frontal 48 mode. It consists in the continuous injection of the sample[12], leading to a half-wave 49 signal reached after saturation of the capillary with the sample. Compared to pulse 50 mode, frontal mode allows reaching higher sensitivity since there is no limitation in 51 terms of introduced quantity of analyte. Each mode leads to different Taylorgram 52 profiles, which are summarized in Figure 1.

53 Mathematical interpretation has already been described in the literature[12,13] for both 54 modes, and is not the object of this paper. Nonetheless, it has to be mentioned that 55 fitting of the signal (either Gaussian or half-wave) using appropriate software is 56 required for the determination of the temporal variance σ^2 of the elution profile, directly 57 proportional to the size of the analyte (see following review[14] for more details). 58 Specific software and mathematical treatment are required for this operation, which 59 can be unappealing for the non-initiated.

In this paper, we aimed at presenting a third mode, denoted as "cross-frontal mode", 60 61 so named because it combines crossed sample fronts. Mathematical demonstration 62 and applications are described for monomodal sample. A comparison with regular 63 frontal mode is also presented, since both modes are comparable in terms of sensitivity 64 and methodology. The main interest of the proposed methodology relies on the 65 absence of fitting required to interpret the data. Similar methodology was applied using 66 capillary zone electrophoresis[15] and is here adapted to TDA application. This method 67 is promising to simplify the interpretation compared to frontal mode while being as 68 simple to perform, since no device modification is required. Demonstration is applied 69 in this paper for analyte size ranging from 0.2 to 5 nm hydrodynamic radius ($r_{\rm H}$), from 70 small organic molecule to biomolecules. Feasibility of the cross-frontal methodology is 71 thus expressed for a broad range of analytes.

72 2. Material and methods

73 2.1. Materials and reagents

74 All reagents and solvents were of analytic grade and were used without further 75 purification. Ultrapure deionized water (> 18.2 M Ω cm) was used for the preparation of 76 all solutions. Sodium phosphate monobasic monohydrate NaH₂PO4.H₂O (for analysis, 77 99 %) was purchased from Merck (Darmstadt, Germany). Sodium phosphate dibasic 78 dodecahydrate Na₂HPO₄.12H₂O (for analysis, 98 %) was purchased from Carlo Erba 79 Reagents (Val de Reuil, France). Sodium citrate dihydrate (≥ 99%), sodium borohydride NaBH₄ (ReagentPlus[®], 99 %), gold(III) chloride trihydrate HAuCl₄.3H₂O 80 81 (ACS reagent, \geq 49.0 % Au basis), caffeine (SigmaUltra), L-glutathione reduced (GSH, 82 \geq 98.0 %), Insulin from bovine pancreas (BioReagent, suitable for cell culture), Bovine 83 serum albumin (BSA, 98 % electrophoresis), hydrochloric acid, HCI, and nitric acid, 84 HNO₃, were purchased from Sigma Aldrich (St Louis, MO, USA). Solutions 85 preparation, citrate-capped gold nanoparticles (AuNP) synthesis[16,17] and the 86 related characterizations are detailed in Supplementary Information.

87 2.2. Apparatus

Taylor dispersion analyses were performed using a P/ACE MDQ CE system (Beckman Coulter) equipped with a diode array detector (DAD, wavelength ranging from 200 to 600 nm). Measurements were performed at 25 ± 1 °C using a 40 cm bare fused silica capillary (effective length: 30 cm, Polymicro Tech.) with inner diameter of 100 µm.

92 2.3. Cross-frontal TDA experimental methodology

- Depending on the analyte, mobilization pressure P and injection time t_{inj} and duration
 of the experiment were adjusted but the concept remains similar for each tested
 condition:
- 96 (i) The capillary was flushed with water, 0.1 M NaOH, and water.
- 97 (ii) the capillary was filled with the sample. Each sample was prepared in a specific
 98 solvent, depending on its properties as mentioned in (iii) below (see Supplementary
 99 Information for more details).
- (iii) the analyte-free solvent, denoted as "empty plug" (a solution of 1.5 mM sodium
 citrate for citrate-capped AuNP analysis, 20 mM phosphate buffer for caffeine and BSA
 analysis, 10 mM HCl for insulin analysis or 1 mM HCl for GSH analysis) was injected
 at P for a duration t_{inj} (P being equal to 0.4, 0.3 or 0.2 psi and t_{inj} varying from 5 to 100
- 104 s, depending on the analyte).
- 105 (iv) Starting the UV-Visible acquisition (at 514 nm for AuNP[18], 272 nm for caffeine,
- 106 214 nm for GSH and 220 nm for BSA and insulin), the empty plug was pushed107 hydrodynamically with the sample using the same P.
- (v) 1.5 mM sodium citrate, 20 mM phosphate buffer, 1 mM HCl or 10 mM HCl
 (depending on the analyte) were pushed using the same P after measurement of the
 negative peak.
- Figure 2 shows a schematic view of a cross-frontal Taylorgram obtained using the experimental methodology, indicating how the mean solute concentration across the cross-section of the capillary \bar{C} and the concentration of the front $\overline{C_0}$ were measured. Each measurement was repeated 3 times for each described condition. Experimental conditions are summarized in Supplementary Information section (Table S1). It is to be noted that higher P values were not tested since Taylor's conditions would not be fulfilled (as mentioned in Experimental results section).

118 **3. Results and discussion**

119 **3.1. Theoretical aspects of cross-frontal TDA mode**

Considering fundamental aspects of Taylor dispersion, it has been
demonstrated[19,20] that the concentration profile (Taylorgram) using frontal mode for
monomodal sample is given by eq. (1):

123
$$\frac{\bar{c}}{\bar{c}_0} = \frac{1}{2} \pm \frac{1}{2} \operatorname{erf}\left[\frac{t-t_R}{\sigma\sqrt{2}}\right] (1)$$

where t is time, t_R is the mean residence time, σ is the square root of the temporal variance of the elution profile and erf is the error function. This equation is adequate for symmetric profile.

The new proposed methodology consists in filling the capillary with the sample, followed by injection of variable volumes (for different t_{inj}) of an "empty plug". Using same velocity (related to P), the empty plug will then be pushed by re-introducing the sample, resulting in the following theoretical Taylorgram which is the sum of two front Taylorgrams i and j (Figure 3.B):

132 Considering the continuous laminar flow (Poiseuille's flow) occurring during the 133 experiment and for a precise value of t_{inj} , both sample profiles (shown by the * on the 134 Figure 3.A) can be described independently using eq. (1). The profile i being centred 135 on tr + $t_{inj}/2$ and j being centred on tr - $t_{inj}/2$, eqs. (2) and (3) give mathematical 136 description of profiles i and j.

137
$$\frac{\overline{C}_{l}}{\overline{C}_{0}} = \frac{1}{2} + \frac{1}{2} \operatorname{erf}\left[\frac{t - (t_{R} + \frac{t_{inj}}{2})}{\sigma\sqrt{2}}\right] (2)$$

138
$$\frac{c_j}{\bar{c}_0} = \frac{1}{2} - \frac{1}{2} \operatorname{erf}\left[\frac{t - (t_R - \frac{1}{2})}{\sigma\sqrt{2}}\right] (3)$$

The Taylorgram is equal to the sum of both eqs. (2) and (3). At $t = t_R$, *i.e.* the intersection of both Taylorgrams i and j (but also the minimum of the peak), the equation can be simplified and give eq. (4):

142

 $\frac{\bar{c}}{\bar{c}_0} = 1 - erf\left[\frac{t_{inj}}{\sigma_2\sqrt{2}}\right] (4)$

143 It is possible to express the previous equation as follow, giving eq. (5):

144
$$erf^{-1}\left[1-\frac{\bar{c}}{\bar{c}_0}\right] = \frac{t_{inj}}{\sigma^2\sqrt{2}}$$
(5)

For different t_{inj}, the previous expression will fit a linear curve with a slope equal to 1/($\sigma 2\sqrt{2}$). The proposed methodology remains in the possibility to inject variable volumes of empty plug (using variable t_{inj}) and to plot eq. (5) in order to calculate σ using slope of the linear curve. Using UV-visible detector, the absorbance is directly proportional to the concentration. It can be approximated that \bar{C} is correlated to absorbance value at t_R and \bar{C}_0 is correlated to absorbance value of the plateau. An illustration of this hypothesis is presented in Figure 3.C.

152

Fulfilling Taylor's conditions[12], the diffusion coefficient D of the analyte can bedetermined using eq. (6):

$$D = \frac{r_C^2 t_R}{24\sigma^2}$$
(6)

with r_c, capillary radius. If no particle interactions occur and analytes are considered as spheres, D can be linked to r_{H} using Stokes-Einstein equation (eq. (7)):

158

$$D = \frac{k_B T}{6\pi\eta r_H} \left(7 \right)$$

159 with k_B the Boltzmann constant (m².kg.s⁻².K⁻¹), T the temperature (K) and η the 160 viscosity of the medium (Pa.s).

161 3.2. Experimental results

162 Using cross-frontal TDA methodology for caffeine (194.2 g/mol), GSH (307.3 g/mol, 163 tripeptide), insulin (~ 5 kDa), BSA (~ 66 kDa) and AuNP (inorganic NP) size 164 determination, D and r_{H} were obtained and results are reported in Table 1. Peclet (Pe) 165 and dimensionless residence time τ are also indicated for the cross-frontal mode, to 166 insure Taylor's conditions were satisfied [12] (Pe > 69 and τ > 1.4, see Supplementary 167 Information). These conditions are dependent on P and capillary dimensions. Every 168 tested values of P meet the criteria for applicability of TDA modes, and are not 169 supposed to influence size determination.

- 170 As summarized, caffeine r_H was measured using cross-frontal mode and values are 171 comparable with reports from the literature for the lowest tested pressures (0.32 and 172 0.35 nm for 0.2 and 0.3 psi respectively against 0.32 to 0.49 nm[21-23]). Figure S6 173 shows the plots for caffeine analysis. These results tend to show the potential of cross-174 frontal TDA mode for small organic molecules size determination. Concerning the 175 lower value of r_H obtained at 0.4 psi (0.25 nm), the deviation can be attributed to the 176 low injection time applied, which cannot be fully approximated and undergoes 177 misinterpretation of the final result. It is therefore important to only use the cross-frontal 178 mode with higher tinj only (> 10 s), since ramp time to reach the pressure becomes 179 negligible. Results were also calculated using frontal mode, with values close to the 180 cross-frontal mode and the one reported in the literature (0.32 nm for frontal mode and 181 cross-frontal mode at 0.2 psi).
- For GSH, values obtained with the cross-frontal mode (0.43, 0.39 and 0.40 nm for 0.2, 0.3 and 0.4 psi) are comparable with reports from the literature (0.36 nm[24]) and with the frontal mode measurement (0.42 nm). Figure S7 shows the plots for GSH analysis. These results tend to show the potential of cross-frontal TDA mode for organic molecules size determination. Moreover, it is the first time TDA was used to measure GSH hydrodynamic radius.

For insulin, values from cross-frontal mode (1.21, 1.16 and 1.09 nm for 0.2, 0.3 and 0.4 psi respectively) are comparable with reports from the literature (1.05 nm[25]) and with the frontal mode measurement (1.04 nm). Figure 4 shows the plots for insulin analysis. These results support the potential of cross-frontal TDA mode for small protein size determination.

193 Concerning BSA, results on r_H are comprised between 5.00 and 3.99 nm (see Figure 194 S8 for cross-frontal Taylorgrams). These values are coherent with previously reported works (see Table 1). In this case, working at P = 0.2 psi, the injection time increased 195 196 up to 100 s, and gave a deviation from the linear fitting model (visible in Figure S8.B). 197 Since viscosity of the sample is different than the 'empty plug' (due to high 198 concentration of BSA), we suppose that a local change in media properties occurs and 199 leads to such a deviation. Another explanation is based upon adsorption of the protein 200 which will be explained later (see Limits and perspectives). With no consideration for 201 this tini, r_H becomes equal to 4.13 nm, closer to results obtained using other pressure 202 levels. Still, variations between r_H from different P are to be mentioned.

203 For citrate-capped AuNP, results for each tested P are closed (Figure S9) and coherent 204 with DLS measurement: a slightly inferior value is reported for TDA (2.82 nm at 0.3 psi 205 using cross-frontal mode against 3.32 nm using DLS). As already stated in the 206 literature for this type of NP (citrate-capped AuNP), DLS measurements are sensitive 207 to potential aggregates, thus leading systematically to a superior value than with 208 TDA[4] since DLS is an intensity-based method, while TDA is a mass-based method. 209 Comparing cross-frontal mode with frontal mode, rH estimated via TDA also gave 210 similar results (2.74 nm using frontal mode). No significant differences comfort the use 211 of cross-frontal mode as an alternative of frontal mode for inorganic NPs size 212 determination. Moreover, pulse analysis of citrate-capped AuNP is hardly feasible in 213 our case since the sample is not concentrated ($< \mu M$). It leads to low signal barely 214 exploitable since injected volume is limited in pulse mode. It is even more convenient 215 to use cross-frontal mode since there is a gain in terms of sensitivity, which makes it a 216 potential alternative for both pulse and frontal modes.

Since results between cross-frontal mode, frontal mode and reported values of the
literature are coherents, one can assume that the proposed methodology is adequate.
One advantage of such a methodology relies in the easy data handling, since no fitting
is required. Moreover, regular TDA conditions (such as values of P, t_{inj}, ...) can be kept,

while non-initiated can start using similar conditions for other molecule/nanoparticlesize determination.

223 3.3. Limits and perspectives

Few aspects need to be discussed concerning limitations of the cross-frontal mode, inherent to TDA. It is known that, depending on the operating conditions, measurable size range is variable. In the case of cross-frontal mode, several precisions are required.

228 For instance, during the development of the methodology, very high injection volume (related to t_{ini}) led to low \bar{C}/\bar{C}_0 ratio. The function erf⁻¹ being strongly impacted for ratio 229 230 values close to 0, strong variations of the ratio occur and deeply impact the final results, 231 with high standard deviation. In order to avoid such an effect, $1-(\bar{C}/\bar{C}_0)$ should not exceed 0.96 (with erf⁻¹ (0.96) = 1.45) since variation of 0.01 on $1-(\bar{C}/\bar{C}_0)$ leads to a 232 233 change of more than 0.08 on erf⁻¹(1-(\bar{C}/\bar{C}_0)) (> 5 % variability). In other words, 234 preliminary analysis using variable tini should be performed to determine the maximum 235 tinj to be used for TDA before reaching the specified value. As for frontal or pulse 236 modes, optimization of TDA conditions is required before use.

Moreover, low t_{inj} should be avoided to decrease the probable variation inherent to the CE device (due to ramp time to reach the pressure for instance), as illustrated for caffeine analysis at 0.4 psi. In our case, 10 s minimum was the best option. This value is dependent on the capacity of the apparatus to deliver a stable pressure profile during the injection.

242 Another aspect which needs to be considered is the sample adsorption. In the case of 243 BSA, no results were indicated for frontal mode (Table 1) since the Taylorgrams were 244 asymmetric and no suitable fittings were obtained for determination of σ . It suggests 245 that interactions between the protein and the inner surface of the capillary may occur, 246 which is detrimental for TDA application[5], no matter the mode used. Even if results 247 for BSA size determination using cross-frontal mode seem to agree with the 248 literature[26], no conclusion can be adopted since Taylor's conditions are not entirely 249 fulfilled. This adsorption can also explain the variations observed using the cross-250 frontal mode for BSA characterization. Use of coating (dynamic or covalent/semi-251 covalent) is suggested, but can be detrimental for detection of the analyte and was not 252 explored here for this reason.

- As a prospect, simplification of the interpretation and similarities between cross-frontal and frontal mode results suggest the possibility to consider the cross-frontal mode as
- an alternative if following conditions are respected:
- (i) Usual TDA conditions (on Pe and τ) are respected.
- 257 (ii) No adsorption of the sample occurs on the inner capillary wall.
- (iii) t_{inj} is sufficiently high to avoid strong variation due to the CE device and
 misinterpretation (> 10 s). If volume is too important, a decrease in P should be
 considered.
- 261 (iv) t_{inj} is not too high to insure $1 (\bar{C}/\bar{C}_0) < 0.96$.
- 262 (v) Sample is monomodal / monodisperse for the equation to be applicable.
- 263 Still, cross-frontal mode is more sensitive than pulse mode since the limitation in terms 264 of injected volume of sample is not to be considered (< 1 %). It may allow to measure 265 less concentrated samples or analytes with limited absorption properties, as illustrated 266 with citrate-capped AuNP. Moreover, cross-frontal mode requires no fitting of the 267 signal, in contrary to frontal mode which facilitates data manipulation, even if similar 268 sensitivity is expected. No particular software is necessary to estimate σ due to a 269 simplified data treatment, which is more appealing for non-initiated and can makes 270 TDA more accessible.

271 **4. Conclusions**

272 To conclude, cross-frontal mode TDA was successfully applied for r_H determination of 273 caffeine (0.32 nm), GSH (0.43 nm), insulin (1.21 nm) and citrate-capped AuNP (2.93 274 nm), with similar results compared to frontal mode TDA and literature. As well as for 275 the pulse and frontal modes, cross-frontal mode methodology is still limited by several 276 requirements. For instance, adsorption of analytes is incompatible with TDA 277 measurements. Still, this new methodology requires no fitting of the Taylorgram, 278 leading to an interesting simplification of the signal treatment. Moreover, the 279 applicability of the method seems feasible for a wide range of analytes (r_H from 0.3 to 280 5 nm), as stated in the paper with small organic molecules, biomolecules and inorganic 281 NP measurements. Larger particles and polydisperse / multi-modal samples should 282 also be tested for validation of the methodology.

283 Credit authorship contribution statement

Jeremie Gouyon: Conceptualization, Investigation, Methodology, Writing – original
draft & editing. Ariane Boudier: Writing - review & editing. Arnaud Pallotta: Writing -

- review & editing. Celia Boukoufi: Investigation, Writing review & editing. Igor Clarot:
- 287 Writing review & editing.

288 **Declaration of Competing Interest**

289 The authors declare that they have no known competing financial

290 interests or personal relationships that could have appeared to influence

the work reported in this paper.

292 Acknowledgements

- 293 This research did not receive any specific grant from funding agencies in the public,
- 294 commercial, or not-for-profit sectors.

295 Appendix A. Supplementary material

296 Supplementary information to this article can be found online.

297 **References**

- Adamo, G., Campora, S., Ghersi, G. In *Nanostructures for Novel Therapy*, Ficai,
 D., Grumezescu, A. M., Eds., Micro and Nano Technologies, Elsevier, 2017, pp.
 57–80.
- 301 [2] Ye, F., Jensen, H., Larsen, S. W., Yaghmur, A., Larsen, C., Østergaard, J. J.
 302 Pharm. Biomed. Anal. 2012, 61, 176–183.
- 303 [3] Hong, J., Wu, H., Zhang, R., He, M., Xu, W. Anal. Chem. 2020, 92, 5200–5206.
- 304 [4] Pyell, U., Jalil, A. H., Pfeiffer, C., Pelaz, B., Parak, W. J. *J. Colloid Interface Sci.* 305 2015, *450*, 288–300.
- 306 [5] R. Moser, M., A. Baker, C. *Anal. Methods* 2021, *13*, 2357–2373.
- 307 [6] Urban, D. A., Milosevic, A. M., Bossert, D., Crippa, F., Moore, T. L., Geers, C.,
 308 Balog, S., Rothen-Rutishauser, B., Petri-Fink, A. *Colloid Interface Sci. Commun.*309 2018, *22*, 29–33.
- 310 [7] Saetear, P., Chamieh, J., Kammer, M. N., Manuel, T. J., Biron, J.-P., Bornhop,
 311 D. J., Cottet, H. *Anal. Chem.* 2017, *89*, 6710–6718.
- [8] Leclercq, L., Reinhard, S., Chamieh, J., Döblinger, M., Wagner, E., Cottet, H.
 Macromolecules 2015, 48, 7216–7221.
- 314 [9] Cottet, H., Biron, J.-P., Martin, M. Anal. Chem. 2007, 79, 9066–9073.
- 315 [10] Casto, L. D., Do, K. B., Baker, C. A. Anal. Chem. 2019, 91, 9451–9457.
- [11] Moser, M. R., Smith, C. M., Gutierrez, G. G., Baker, C. A. Anal. Chem. 2022, 94,
 6089–6096.
- 318 [12] d'Orlyé, F., Varenne, A., Gareil, P. J. Chromatogr. A 2008, 1204, 226–232.
- [13] Labied, L., Rocchi, P., Doussineau, T., Randon, J., Tillement, O., Lux, F.,
 Hagège, A. *Anal. Chem.* 2021, *93*, 1254–1259.
- [14] Gouyon, J., Boudier, A., Barakat, F., Pallotta, A., Clarot, I. *ELECTROPHORESIS* 2022, *43*, 2377–2391.
- 323 [15] Jin, W., Chen, H. *Chromatographia* 2000, *52*, 17–21.
- Beurton, J., Lavalle, P., Pallotta, A., Chaigneau, T., Clarot, I., Boudier, A. *Int. J. Pharm.* 2020, *580*, 119244.
- Berthou, M., Pallotta, A., Beurton, J., Chaigneau, T., Athanassiou, A., Marcic, C.,
 Marchioni, E., Boudier, A., Clarot, I. *J. Chromatogr. B* 2022, *1198*, 123244.

- 328 [18] Tournebize, J., Sapin-Minet, A., Schneider, R., Boudier, A., Maincent, P., Leroy,
 329 P. *Talanta* 2011, *83*, 1780–1783.
- 330 [19] Taylor, G. I. Proc. R. Soc. Lond. Ser. Math. Phys. Sci. 1953, 219, 186–203.
- 331 [20] Sharma, U., Gleason, N. J., Carbeck, J. D. Anal. Chem. 2005, 77, 806–813.
- 332 [21] Hulse, W. L., Forbes, R. T. Int. J. Pharm. 2011, 411, 64–68.
- 333 [22] Leaist, Derek., Hui, Lu. J. Phys. Chem. 1990, 94, 8741–8744.
- 334 [23] Chamieh, J., Oukacine, F., Cottet, H. J. Chromatogr. A 2012, 1235, 174–177.
- 335 [24] Stricks, W., Kolthoff, I. M. J. Am. Chem. Soc. 1952, 74, 4646–4653.
- 336 [25] Yousefi, R., Taheri, B., Alavi, P., Shahsavani, M. B., Asadi, Z., Ghahramani, M.,
 337 Niazi, A., Alavianmehr, M. M., Moosavi-Movahedi, A. A. J. Biomol. Struct. Dyn.
 338 2016, 34, 362–375.
- 339 [26] Hulse, W., Forbes, R. Int. J. Pharm. 2011, 416, 394–397.
- Mazloum-Ardakani, M., Sheikh-Mohseni, M. A., Mirjalili, B.-F. *Electroanalysis* 2013, 25, 2021–2029.
- 342 [28] Jensen, S. S., Jensen, H., Cornett, C., Møller, E. H., Østergaard, J. J. Pharm.
 343 Biomed. Anal. 2014, 92, 203–210.
- 344 [29] Jachimska, B., Wasilewska, M., Adamczyk, Z. *Langmuir* 2008, *24*, 6866–6872.
- 345 [30] Le Saux, T., Cottet, H. Anal. Chem. 2008, 80, 1829–1832.



350 351 Figure 1 - Taylor dispersion analysis representation of (A) pulse and (B) frontal mode. [Reprinted, with permission, with modifications from ref [14]. Copyright © 2022 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim]



354 355 Figure 2. Schematic view of a cross-frontal Taylorgram. The coloured part indicates the time at which the sample is hydrodynamically pushed, before continuous injection of the solvent in order to decrease the signal to obtain $\overline{c_0}$ and \overline{c} .



Figure 3. (A) Schematic capillary profile of the proposed cross-frontal TDA mode and (B) example of resulting Taylorgram centred on t_R . Sample (blue) contains the analyte to be measured, and empty plug (clear) is composed of solvent free of analyte. (C) Simulated cross-frontal Taylorgrams using variable t_{inj} for $t_R = 5$ min and $\sigma = 0.3$ min and (D) eq. (5) plot for determination of σ , used to calculate diffusion coefficient D and r_H .

364



367 Figure 4. (A) Typical cross-frontal Taylorgrams obtained for insulin using different P and t_{inj} and 368 (B) resulting plot for determination of σ (n=3 for each level)

Table 1 – Results obtained using cross-frontal mode TDA and comparison with frontal mode TDA and literature data. Values were statistically evaluated by the Student's Test (p < 0.05).

Commis		Cross-frontal mode			Frontal*	r _н th.
Sample	P (psi)	0.2	0.3	0.4		Ret.
	σ (min) D (.10 ⁻¹⁰ m².s ⁻¹)	0.125 ± 0.013 6.26 ± 0.66	0.106 ± 0.021 5.65 ± 1.11	0.077 ± 0.013 8.10 ± 1.38	-	
Caffeine	r_н (nm) τ / Pe	0.32 ± 0.03 85.1 / 94	0.35 ± 0.07 49.9 / 160	0.25 ± 0.04 54.9 / 146	0.32 ± 0.00 -	0.32^a - 0.491 ± 0.008 ^a [21–23]
	σ (min) D (.10 ⁻¹⁰ m².s ⁻¹)	0.150 ± 0.005 4.64 ± 0.17	0.113 ± 0.007 5.16 ± 0.32	0.099 ± 0.004 4.98 ± 0.18	-	
GSH	r _н (nm)	0.43 ± 0.02	0.39 ± 0.02	0.40 ± 0.01	0.42 ± 0.00	0.36 ^b - 0.81 ± 0.05 ^c
	τ / Pe	66.6 / 120	47.3 / 169	33.6/238	-	[24,27]
	σ (min) D (.10 ⁻¹¹ m².s ⁻¹)	0.241 ± 0.009 1.66 ± 0.06	0.191 ± 0.002 1.72 ± 0.02	0.161 ± 0.004 1.84 ± 0.04	-	
Insulin	r_н (nm) τ / <i>P</i> e	1.21 ± 0.04 22.0/364	1.16 ± 0.01 15.0 / 535	1.09 ± 0.03 12.2 / 654	1.04 ± 0.01 -	1.05 ± 0.1 ^d - 1.7 ^a [25,28]
BCA	σ (min) D (.10 ⁻¹¹ m².s ⁻¹)	0.494 ± 0.032 4.01 ± 0.26	0.367 ± 0.017 4.51 ± 0.21	0.307 ± 0.016 5.02 ± 0.27		
БЭА	г_н (nm) τ / Ре	5.00 ± 0.32 5.4 / 1477	4.44 ± 0.21 3.8/2114	3.99 ± 0.21 3.3 / 2432	**	3.3^d - 4.84 ± 0.03 ^a [21,26,29]
AND	σ (min) D (.10 ⁻¹¹ m².s ⁻¹)	0.385 ± 0.021 6.84 ± 0.31	0.298 ± 0.010 7.10 ± 0.24	0.266 ± 0.017 6.54 ± 0.42	-	
AUNP	r_H (nm) τ / <i>P</i> e	2.93 ± 0.14 9.6/833	2.82 ± 0.10 6.2 / 1292	3.06 ± 0.20 4.2 / 1914	2.74 ± 0.10	3.32^d (2.9 ± 0.5) [16]

372 *Results for frontal analysis were obtained by fitting the experimental curve with eq. (1), see Supplementary
 373 Information for more details[30]

**Results for frontal analysis of BSA are not provided since no suitable fitting of the experimental curve were
 possible, due to asymmetric signals

376 ^a Result obtained by TDA /^b Result obtained by polarography /^c Result obtained by chronoamperometry /^d
 377 Result obtained by DLS